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Magnetism of MnBi-Based Nanomaterials

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Nanostructured MnBi ribbons doped with impurity elements including B, C, Fe, Hf, Sm and Tb were prepared using the arc melting and melt-spinning techniques. The melt-spun ribbons were annealed in vacuum furnace at 350\textdegree C to obtain the intended hexagonal structure. The external impurity doping made a significant change in the magnetic properties of the nanostructured MnBi ribbons including a decrease in saturation magnetization (\textit{M}_s) and anisotropy energy (K) and an increase in coercivity (H_c). However, Hf and C co-doping showed the opposite effect with a small increase in both M_s and K. Interestingly, the anisotropy energy of the boron doped sample increased by about 15\% irrespective of the small decrease in magnetization. A significant increase in H_c of MnBi ribbons was found due to Hf, Tb and Sm doping. H_c as high as 13 kOe was achieved in Hf-doped sample after the sample was aligned in a magnetic field. A thermal hysteresis was observed at the structural phase transition of MnBi, which shifts by about 5 K towards higher temperatures due to impurity doping. The observed magnetic properties of the impurity doped MnBi ribbons are explained as the consequences of the disorder and the competing ferromagnetic and antiferromagnetic interactions.

Index Terms—Magnetic anisotropy, melt-spin ribbons, permanent magnet.

I. INTRODUCTION

The non-rare-earth materials having high magnetic anisotropy and high Curie temperature are gaining significant attention because of increasing cost, irregular supply and increasing demand of some of the rare-earth elements which are essential for producing powerful permanent magnets [1], [2]. A manganese-based ferromagnetic material MnBi is one of such materials that has a high magnetic anisotropy (K) of 16 Mergs/cm\textsuperscript{3} at room temperature and also has a high Curie temperature, well above room temperature. An interesting feature in the magnetic anisotropy of MnBi is that the value of K is negative below 100 K but increases with increasing temperature and attains its maximum value of 22 Mergs/cm\textsuperscript{3} at 490 K [3]. The high-temperature value of K is much higher than that of Nd\textsubscript{2}Fe\textsubscript{14}B magnet, the most powerful (highest energy product) permanent magnet developed so far. This has made MnBi a material of interest for high-temperature permanent magnet applications. Other interesting properties of MnBi in thin films are the high value of room temperature perpendicular anisotropy [4], the high value of transport spin polarization [5] and an unusual spin correlations leading to a Kondo effect when doped with heavy and noble metals such as Pt and Au [6], [7].

Although magnetic anisotropy of MnBi is promising for permanent-magnet applications, the maximum energy product (7.1 MGOe at room temperature) that determines the strength of a magnet is relatively low for most practical applications [8]. This is because the theoretical limit of saturation magnetization for MnBi is about 712 emu/cm\textsuperscript{3}, which is much smaller than that of the most other permanent magnet materials [6], [9]. Some of the other issues concerning the structural and magnetic properties of this material are (i) the structural phase transition before the Curie temperature is reached and (ii) the diffusion of Mn atoms into the empty interstitial sites causing the net magnetization to decrease. Also, MnBi is highly susceptible to moisture and oxygen contamination. Therefore, there have been extensive research efforts to increase the resistance to degradation, improve magnetic properties and form a phase that is stable and exhibits a high Curie temperature. These improvements are expected to be achieved by alloying MnBi with a suitable third element. Our interest is in the magnetic properties, where we have investigated the effect of external impurity doping on the saturation magnetization, coercivity, anisotropy energy and Curie temperature of the nanostructured MnBi ribbons. We have chosen six doping specimens; (i) a ferromagnetic transition metal (Fe) which has a high saturation magnetization (ii) two rare-earth elements; one (Sm) having positive and the other (Tb) having negative atomic quadrupole moments (iii) a 5d element (Hf) where the induced magnetic moment is expected to be small and to couple with the manganese moments and (iv) two non-magnetic elements (B and C) which are expected to help in developing structural stability.

II. EXPERIMENTAL METHODS

Nanostructured MnBi:X (X = B, C, Fe, Hf, Sm and Tb) ribbons were prepared by arc-melting, melt-spinning and annealing. The arc-melting process, which produces MnBi:X ingots from the respective metal pieces, was carried out on a water cooled Cu hearth in a highly pure argon environment. In order to produce nanostructured ribbons, a molten mixture formed by melting MnBi:X ingot was ejected onto the surface of a rotating copper wheel where it rapidly solidified into ribbons. The tangential speed of the rotating wheel was kept at 40 m/s for all the samples and the melt-spinning process was carried out in the melt-spinner chamber filled with a suitable pressure of highly pure argon gas. In order to obtain the intended crystal structure, the ribbons were annealed in a tubular vacuum furnace (base pressure ~ 10\textsuperscript{−6} Torr) at 350\textdegree C for 4 hours. We also prepared powder samples by mechanically grinding the annealed ribbons in acetone. The elemental compositions of the samples as determined by energy dispersive x-ray (EDX) spectroscopy are Mn\textsubscript{1.5}Bi\textsubscript{46}, Mn\textsubscript{1.5}Fe\textsubscript{4}Bi\textsubscript{43}, Mn\textsubscript{47}Hf\textsubscript{2}Bi\textsubscript{50}, Mn\textsubscript{51}Sm\textsubscript{4}Bi\textsubscript{45} and Mn\textsubscript{46}Tb\textsubscript{4}Bi\textsubscript{42}. Because of the big difference in the melting points of Mn and Bi, it is very difficult to control the elemental composition during melt spinning. Since the content of B and C could not be
determined by EDX spectroscopy, we estimated the amount of these elements from their weights in the starting mixture, which is about 1%. One of the samples we studied was co-doped with Hf and C (2% Hf and 1% C). The structural properties of the samples were studied using x-ray powder diffraction and the magnetic properties were investigated using a Quantum Design Magnetic Properties Measurement System (MPMS) and Physical Properties Measurement System (PPMS).

III. RESULTS AND DISCUSSION

As shown in Fig. 1, the most intense peaks in the room-temperature x-ray diffraction (XRD) patterns of the MnBi and MnBi:X (X = Fe, Tb, Sm, B, HfC) ribbons are indexed based on the hexagonal NiAs crystal structure. The diffraction patterns indicate that the impurity doping has not produced a significant change in the lattice symmetry but there is a noticeable change in both the c and a lattice parameters (see the inset of Fig. 1). The lattice parameters were calculated from the Rietveld analysis of the powder x-ray diffraction data. However, refinement of fractional occupancy of the different atomic species was excluded in this analysis. As a consequence of preferential orientation of crystallites and possible fractional occupation of dopant atoms at the crystallographic sites, there is a slight deviation of the calculated diffraction pattern. However, the accuracy of lattice parameters is consistent with the assumption that a fraction of the dopant atoms may have been incorporated into the MnBi lattice. Although there are reflections from elemental impurities, no indications of secondary alloy phases have been found within the limit of x-ray diffraction. All the MnBi:X ribbons are composed of nanocrystalline grains but the grain sizes are different for the ribbons of different elemental compositions. The grain sizes as determined using the Scherrer’s formula are between 27 nm and 43 nm (27 nm for MnBi:HfC and 43 nm for MnBi:Sm) showing that the grain size is sensitive to the nature of the dopant elements. We note that the ribbons were prepared under similar synthesis and annealing conditions.

The magnetic properties of the MnBi ribbons are highly sensitive to the external impurity doping. Although some of the samples contain a small amount of elemental Mn and Bi, the contribution from these elements to the observed magnetic properties can be considered to be negligibly small. Fig. 2 shows the M(H) hysteresis loops for four representative samples, namely, MnBi, MnBi:Hf, MnBi:B and MnBi:Tb. All the samples show strong ferromagnetic moment at room temperature but the M(H) loops are not saturated even at 7 T. At 7 T, we have measured a large room-temperature magnetization of about 675 emu/cm³ (75 emu/g) in the undoped MnBi ribbons which is very close to the theoretical value of 712 emu/cm³ (80 emu/g). Magnetization (M), a parameter related to the strength of a magnet, showed significant decrease with impurity doping, irrespective of whether the dopants carried magnetic moment or not except in the case of Hf and C co-doping where magnetization showed moderate increase. As shown in the lower panel of the inset of Fig. 2, the high-field room-temperature magnetizations (M_T) recorded for B, Fe, Hf, Sm and Tb doped ribbons are 583, 655, 563, 627, 506 and 534 emu/cm³ respectively. However, the observed value of M_T (686 emu/cm³) in Hf and C co-doped sample is little higher than the value measured in MnBi ribbon. In contrast to the magnetization, the anisotropy energy (K) shows little increase above the level of MnBi (11 Mergs/cm³ for MnBi and 13 Mergs/cm³ for MnBi:B) due to B doping. However, as shown in the upper panel of the inset of Fig. 2, there is a significant decrease (about 40%) in the value of K due to Fe doping. Tb and Sm doping produced between 10 to 15% decrease in K. The change in the value of K due to Hf and C is very small. The anisotropy energy was calculated using the approach to saturation method, where the high-field data was fitted to $M = M_0(1 - A/H^2) + \chi H$; $A = 4 K^2/15 M_T^2$. The parameters $M_0$, $A$ and $\chi$ are the spontaneous magnetization, a constant that depends on K and the high field susceptibility, respectively [10].
As mentioned above, one of the issues related to MnBi is the interstitial occupation of Mn or dopant atoms. Interstitial Mn moments couple antiferromagnetically with the other Mn atoms causing the net magnetization to decrease [6]. The exact positions of the impurity atoms in the MnBi lattice are uncertain but the decrease in magnetization is consistent with the competing ferromagnetic and antiferromagnetic interactions in the MnBi system. The small increase of magnetization in the Hf and C co-doped sample may indicate that the diffusion of Mn into the interstitial sites is relatively small in this case and possibly there is a ferromagnetic coupling between Mn and Hf moments. Magnetocrystalline anisotropy of a compound depends on its electronic structure and hence on the crystal structure. Hence, the change in atomic arrangement due to external impurity doping may also change the value of anisotropy energy [11]. Since the signs of the atomic quadrupole moments for Sm and Tb are opposite, their contributions to the anisotropy energy are also expected to be opposite, provided that the doping does not make any change in the crystal lattice. In our case, such an opposite trend was not found suggesting a different source for the change in the anisotropy energy of the impurity-doped MnBi ribbons. It is likely that the large decrease in K in Fe doped sample is due to the interstitial disorder where Fe occupies the Mn sites, pushing more Mn into the interstitial sites [12]. However, boron, that has a lot smaller atomic radius compared to that of Mn and Bi, may occupy vacant interstitial sites reducing the diffusion of Mn onto those sites leading to anisotropy energy to increase.

As shown in Fig. 3, there is a substantial increase in the value of \( H_c \) as the MnBi ribbons are doped with heavy elements including Hf, Sm, and Tb but the change is relatively small for B, C, and Fe doping. Basically, coercivity of a magnet depends on the magnetic anisotropy and also on the microstructure of the sample. In our case, there is an apparent change in both the grain size and the anisotropy energy of the MnBi ribbons due to impurity doping but this change does not directly correlate with the observed change in coercivity. However, since the samples doped with heavy elements show larger coercivities, the role of magnetic anisotropy cannot be completely ruled out. Furthermore, MnBi shows a high perpendicular anisotropy in thin films and the isotropic ribbons prepared at high wheel speed (40 m/s) are not expected to show very high coercivity.

As mentioned above, MnBi shows a high perpendicular anisotropy and also a large coercivity in thin films. This is because the easy axis of magnetization is along the c-axis, which is perpendicular to the film plane in the textured films. In order to align the particles in a magnetic field, we mechanically crushed the ribbons into fine powder in acetone and mixed with epoxy resin. A thin layer of this dilute mixture was placed on a clean silicon piece and allowed to set in a perpendicular magnetic field of about 9 kOe. Acetone is expected to reduce the oxidation of the samples during milling but x-ray diffraction patterns of the powder samples show that the oxidation was not completely prevented by acetone [13]. Fig. 4 shows the easy-axis and hard-axis M(H) loops of the magnetic-field aligned Mn\(_{54}\)Bi\(_{46}\) sample measured at room temperature where the easy-axis M(H) loop is wide and almost rectangular. The easy-axis M(H) loops of all other MnBi:X samples are also almost rectangular and there is a clear anisotropy between the easy-axis and hard-axis M(H) loops. Easy-axis M(H) loops of Hf- and B-doped samples are shown in the top-left inset of Fig. 4. However, the fairly wide hard-axis M(H) loop suggests that a substantial fraction of the particles are still not aligned with their c-axis parallel to the applied magnetic field. Magnetic field alignment has produced a large gain in both the coercivity and squareness ratio \((M_r/M_s)\) of MnBi and MnBi:X samples. As shown in Fig. 3, the easy-axis coercivites of most of the samples are in the range of 10 kOe with a maximum \( H_c \) of 13 kOe in Hf doped sample. Furthermore, the ratio \( M_r/M_s \) is greater than 90% for all the aligned MnBi:X samples. We note that the values of \( M_r/M_s \) measured in the powder samples are slightly smaller than the values measured in the corresponding ribbons. For example, the \( M_r/M_s \) measured in Mn\(_{54}\)Bi\(_{46}\) powder is about 10% lower than the value measured in the ribbon [see the bottom-right inset of Fig. 4].

For MnBi, the critical single-domain diameter in the demagnetized state \( (d_c) \) calculated using the Kittel formula \( d_c = (15\gamma \times 4\pi)/J_{sat}^2 \), where \( \gamma = 4(K_1)^{1/2} \) is the domain wall energy, is about 0.42 \( \mu \)m. We have used the magnetic exchange energy \( A = 1.2 \times 10^{-6} \text{ ergs/cm}^2 \) [4], the uniaxial anisotropy energy \( K_1 = 9 \text{ Mergs/cm}^3 \) [2] and the saturation polarization \( J_s = 4\pi M_s = 8.4 \text{ kG} \) [8] for this calculation. This suggests...
that the average grain sizes in all of these MnBi:X ribbons are within the critical single-domain diameter range. However, all the aligned particles are not expected to be single domain, because the mechanical milling is not sufficient to crush the ribbons into particles as small as few tenths of a micrometer in size. This is surprising that the milled powder was well aligned in the magnetic field, although the particles contain thousands of grains. This suggests that the particles are locally textured and the applied magnetic field easily rotates the particles with their easy axis along the field direction.

As shown in Fig. 5, MnBi$_{54}$Bi$_{46}$ ribbons undergo a coupled structural and magnetic phase transition from a ferromagnetic low-temperature phase (LTP) to a paramagnetic high-temperature phase (HTP) at 620 K when the sample is heated from room temperature. When the sample is cooled from 750 K, the sample transforms from HTP to LTP at 580 K. The phase transition temperatures were determined from the extrapolation of the straight portion of the M versus 1/T curves. These values are very close to the transition temperatures (633 K for LTP to HTP and 603 K for HTP to LTP) determined from the neutron diffraction studies of bulk MnBi [14]. The phase-transition temperature has increased by about 5 K due to impurity doping [see the inset of Fig. 5]. This indicates that the impurity doping has produced some modifications in the MnBi crystal lattice, which is consistent with our x-ray diffraction studies.

IV. CONCLUSIONS

In summary, we prepared MnBi ribbons doped with various impurity elements including B, C, Fe, Hf, Sm and Tb using arc-melting, melt-spinning and annealing. Magnetic properties of MnBi ribbons are very sensitive to the external impurity doping. Most of the samples show increase in coercivity and decrease in magnetization and anisotropy energy due to impurity doping. However, the sample doped with B shows significant increase in anisotropy energy and the sample co-doped with Hf and C shows a moderate increase in magnetization. The measured saturation magnetization is as high as 686 emu/cm$^3$ in Hf and C co-doped sample with a maximum coercivity of 13 kOe in the Hf-doped sample. All the MnBi:X samples show coupled structural and magnetic phase transitions around 625 K which is about 5 K higher than that of undoped MnBi. The observed magnetic properties are explained as the consequences of the disorder and competing ferromagnetic and antiferromagnetic interactions.

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