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Evaluation of Liquid Nitrogen Freeze Drying and Ethanol Dehydration as Methods to Preserve Partially Cooked Starch and Masa Systems¹

Roxana Yglesias² and David S. Jackson^{2,3}

ABSTRACT

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Preservation of starch structure/properties, including structures formed during partial or complete cooking, are important when the impact of processing conditions is being studied. Two preservation techniques used to study changes in starch during thermal-mechanical processing are commonly cited in the literature: 1) rapid freezing followed by lyophilization, and 2) a dehydration procedure using alcohols. A comparative determination on how these methods affect various starch structures has not been widely reported. Corn starch samples were collected from the Rapid Visco-Analyser (RVA) at 3 min (swollen granules, 30°C), at the top of the pasting peak (gelatinized granules, 95°C), at the bottom of the trough (dispersed polymers, 95°C), and a completed RVA sample stored

for 120 hr at 4°C (retrograded starch). Samples of masa were obtained by nixtamalizing corn. Differential scanning calorimetry (DSC) endotherms of starch and masa, and X-ray diffraction (XRD) patterns of masa were evaluated after being preserved by alcohol- or freeze-drying. No significant differences ($P > 0.05$) between methods were found for onset, end, and peak temperatures (°C), enthalpy (J/g) and % relative crystallinity in any of the samples analyzed. Liquid nitrogen freeze-drying and ethanol dehydration are both effective methods of preserving various starch systems for structural changes detectable by DSC and XRD; freeze-drying is generally less expensive and time-consuming.

Staling is a significant problem in corn tortillas. Aged tortillas are firmer, more rigid, and less rollable than fresh tortillas (Suhendro et al 1998). To prevent or retard staling, several investigations have tried to elucidate how tortillas stale (Fernandez-DeCastro 1988; Hebeda et al 1991; Friend et al 1993; Yau et al 1994; Fernandez et al 1999). Starch is the major component (≈80% db) in corn tortillas (Fernandez et al 1999). Therefore, attempts to explain staling by characterizing starch physical and chemical changes during storage have been made (Ghiasi et al 1979; Martin et al 1991; Ruan et al 1996). Due to the rapid and substantial changes in starch structures in the first hours after corn tortillas are cooked (Fernandez et al 1999), it is vital to preserve and stabilize starch properties and characteristics for accurate analysis and study. In general, there are two methods widely used for this purpose: 1) rapid freezing followed by lyophilization (freeze-drying), and 2) a dehydration procedure using ethanol (Fernandez et al 1999). Very little is known about the effect these treatments have, if any, on starch. Some researchers believe that the process of lyophilization causes gelatinized molecules to retrograde with consequent changes in the properties of the starch being evaluated. Retrogradation has been used to describe starch structural changes following gelatinization (Cameron and Donald 1991) and also as a stage of increased order (degree of crystallinity) from initially dispersed starch molecules (Atwell et al 1988). Differential scanning calorimetry (DSC) and X-ray diffraction methods are widely used as tools to investigate the order-disorder transitions that occur to an aqueous suspension of starch granules upon heating (Eliasson 1985; Cameron and Donald 1991; Liu et al 1992).

Masa can be considered a system of solubilized starch polymers supporting uncooked, swollen, dispersed, and retrograded starch granules (Gomez et al 1990).

This work was designed not only to study masa as a network of different starch organizations but to be able to investigate each molecular starch structure individually. A starch model system was generated for this purpose and samples of swollen, gelatinized, dispersed, and retrograded corn starch granules were investigated.

Thermal properties of samples measured by DSC, and relative crystallinity quantified by X-ray diffraction of nixtamalized masa and the starch model system were studied as affected by the two preservation methods described above.

MATERIALS AND METHODS

Sample Preparation of Starch and Masa Systems

Slurries of native corn starch (3.85 g of starch and 25.15 g of water) were loaded into a Rapid Visco-Analyser (RVA). The scan profile used started at 30°C (held for 10 min), ramped to 95°C (in 6 min), and was maintained at 95°C for 5 min. Samples were cooled to 50°C (in 4 min) and held at 50°C for 10 min. The RVA paddle was programmed to turn at 160 rpm to help mix the sample. Swollen granules, gelatinized, dispersed, solubilized, and retrograded starch were obtained by collecting four samples from the RVA profile 1) at 3 min, 2) at the viscosity peak (16.06 min), 3) at the trough (21.16 min), and 4) a completed RVA sample stored for 120 hr at 4°C, respectively. Every 29 g of RVA starch/water system sample was divided in two 14.5-g subsamples. Each subsample was placed in two 50-mL disposable/graduated centrifuge tubes with screw caps until preserved (≈7 g/container).

Masa was prepared using a small-scale laboratory technique. Clean corn (500 g) was cooked at 90°C for 30 min in a 2,000-mL glass beaker with 1,500 mL of water and 5 g of food-grade lime Ca(OH)₂, (Mississippi Lime Co., Alton, IL). Corn was steeped for 8.5 hr, rinsed twice using 700 mL of water per wash, drained and ground into masa dough with a small-scale stone grinder as generally described by Yglesias et al (2005).

Samples of the starch model system and masa were preserved within the first 5 min after they were prepared. The order in which each sample was preserved with a given method was randomly assigned.

Starch Preservation by Liquid Nitrogen Freeze-Drying

Two centrifuge tubes containing a sample of the starch model system (total of 14.5 g/sample) or 30 g of nixtamalized masa were placed in a 600 mL wide-mouth freeze-drying flask (The VirTis Co., Gardiner, NY) with ≈300 mL of liquid nitrogen until completely frozen. Flasks were connected to an 8L benchtop freeze-drier (VirTis) at –50°C for 24 hr. Samples were then ground using a cyclone sample mill with a 1-mm mesh screen (model 3010-030, Udy Corp., Fort Collins, CO) for further analysis.

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Starch Preservation by Ethanol Dehydration

Two centrifuge tubes of each starch sample (≈ 7 g/container) or four centrifuge tubes for each 30-g nixtamalized masa subsample (≈ 7.5 g/container) were filled with 30 mL of 100% ethanol. Slurries were transferred to a mortar, manually ground for 1 min, and replaced in the containers. Tubes were shaken manually for 1 min and centrifuged for 10 min at $3,400 \times g$. The supernatant was discarded and the pellet was extracted again. Solids were dried in a forced-air oven (model 1350F, Sheldon Manufacturing, Cornelius, OR) at 45°C for 2 hr as specified by Fernandez et al (1999). Samples were then ground before analysis using the same cyclone sample mill used for freeze-dried samples.

DSC Analysis

Thermal properties of masa and starch samples, after being preserved by liquid nitrogen-freeze drying or ethanol dehydration, were investigated using a differential scanning calorimetry (Pyris 1, Perkin Elmer Co., Norwalk, CT). Samples (≈ 10 mg) were accurately weighed into stainless steel pans and hydrated with 55 μL of distilled water to $\approx 80\%$ moisture content. Pans were sealed and stored at room temperature ($23 \pm 2^\circ\text{C}$) for ≈ 12 hr. Endotherms were obtained after heating the aluminum pans from 30 to 110°C at a rate of $5^\circ\text{C}/\text{min}$. Onset temperature ($^\circ\text{C}$), peak temperature ($^\circ\text{C}$), end temperature ($^\circ\text{C}$), and enthalpies (J/g) were calculated by endotherm peak analysis.

X-ray Diffraction

X-ray diffraction patterns were obtained using an X-ray diffractometer (D-Max/B Horizontal Q/2Q, Rigaku Denki Co.,

Japan). Masa samples were placed in a zero-cavity flat-mount glass and a metal spatula was used to even out the sample thickness; 100% ethanol was added to fix it to the mount. Slides were placed into a sample holder inside the diffractometer and scanned from $2\theta = 3\text{--}30^\circ$ with $2^\circ/\text{min}$ increments. The X-ray diffractometer was operated at 40 kV tube voltage and at 30 mA tube current. Diffraction patterns were analyzed using MacDiff X-ray diffraction peak analysis software (v. 4.2.5, Geologisch-Paläontologisches Institut, Frankfurt, Germany). The lower points of the diffraction peaks were connected by a smooth curve (crystallinity line). Area under the peaks, but over the crystallinity line, was estimated as the crystalline component. A baseline connecting the starting and end points of the crystallinity line was traced to estimate the area corresponding to the amorphous region; the amorphous region was the area between the crystallinity line and the baseline. Percentage of relative crystallinity was calculated by dividing the area of the crystalline region by the area of the amorphous region (Roe 2000).

Statistical Analysis

The starch model system was analyzed as a split-plot where type of starch structure (swollen, gelatinized, dispersed, and retrograded) was the whole plot and preservation method (liquid nitrogen freeze-drying and ethanol dehydration) was the split-plot factor. The experiment was repeated three times and the analysis of variance performed with SAS Proc Mixed, adjusting degrees of freedom for testing by the Satterthwaite's approximation (Mead 1988). Means were separated using the lsmeans procedure (v. 8, SAS Institute, Cary, NC). Masa was analyzed as a randomized complete block design (RCBD) where the masa production batches were the blocking factor and preservation methods the treatments. The experiment was repeated three times. The analysis of variance was run using SAS Proc Mixed, and means separated with lsmeans procedure.

RESULTS AND DISCUSSION

Starch Model System

Thermal properties of representative samples of the starch model system are shown in Fig. 1. Swollen granules showed a peak at $\approx 62^\circ\text{C}$, indicating melting of the crystallites or double helices within the range expected for corn starch gelatinization (Morris 1990; Sahai 1999). Gelatinized and dispersed starch granule samples showed no peak at this temperature, which is evidence of com-

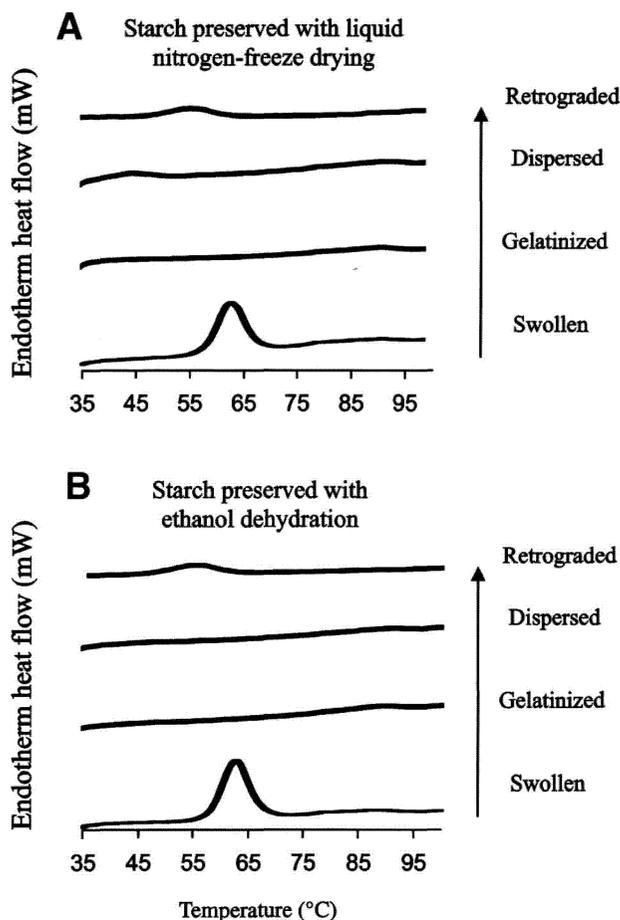


Fig. 1. Representative DSC endotherms for the starch model system after being preserved by liquid nitrogen-freeze drying and by ethanol dehydration.

TABLE I

DSC Thermal Properties Means, Standard Errors, and *P* Values Comparing Starch Swollen Granules by Method of Preservation^a

Treatment	Temperature ($^\circ\text{C}$)			Enthalpy (J/g)
	Onset	Peak	End	
Liquid nitrogen-freeze drying	58.03	62.33	66.40	11.10
Ethanol dehydration	58.16	62.51	65.68	11.67
Standard error	0.42	0.53	0.98	0.51
<i>P</i> value	0.9110	0.1334	0.4891	0.8950

^a Mean values for three treatment replicates.

TABLE II

DSC Thermal Properties Means, Standard Errors, and *P* Values Comparing Retrograded Starch by Method of Preservation^a

Treatment	Temperature ($^\circ\text{C}$)			Enthalpy (J/g)
	Onset	Peak	End	
Liquid nitrogen-freeze drying	47.26	54.87	61.61	4.08
Ethanol dehydration	47.53	56.39	64.45	3.71
Standard error	0.42	0.53	0.98	0.51
<i>P</i> value	0.9110	0.1334	0.4891	0.8950

^a Mean values for three treatment replicates.

plete gelatinization. A peak at $\approx 55^\circ\text{C}$ for the retrograded starch endotherm illustrates the retrograded amylopectin crystal melting site as established by other authors (Fernandez et al 1999; Seetharaman et al 2002). These data demonstrate that specific starch structures at the phases investigated (swollen, gelatinized, dispersed, and retrograded) can be successfully obtained from an appropriately programmed RVA instrument.

Means and standard errors for DSC thermal properties for swollen and retrograded starch are shown in Tables I and II, respectively. *P* values for all the variables measured showed no differences ($P > 0.05$) when comparing the two methods of starch preservation. There were no detectable endotherms for gelatinized and dispersed starch. These findings demonstrate that for individual molecular forms (swollen, gelatinized, dispersed, and retrograded), liquid nitrogen-freeze drying and ethanol dehydration are equally appropriate in preserving characteristics for further analysis.

Masa Systems

Starch gelatinization during alkaline cooking and steeping is restricted by insufficient heat and moisture (Gomez et al 1992) and by amylose-calcium interactions (Robles et al 1988). Masa can be considered to be a network of solubilized starch polymers supporting uncooked, swollen, and dispersed starch granules (Gomez et al 1990). In addition, retrogradation of gelatinized starch granules takes place very rapidly upon masa cooling; DSC traces have previously shown melting of an amylose-lipid complex peak at $\approx 95^\circ\text{C}$, and after storage, an amylopectin recrystallization peak at $\approx 55^\circ\text{C}$ (Seetharam et al 2002). None of these retrogradation peaks were observed in this study; presumably there was insufficient time for molecules to reorganize, as starch preser-

vation methods were started when masa was still warm (first 10 min after grinding). Figure 2 shows representative DSC endotherms for nixtamalized masa. The peak at $\approx 69^\circ\text{C}$ is indicative of starch gelatinization.

One of the other starch forms thought to exist in masa is annealed starch (Gomez et al 1990). Annealing takes place during steeping as starch is heated in excess water for a period of time at subgelatinization temperatures (Gomez et al 1992). This alteration within the masa crystallites can explain the increase in masa gelatinization temperature ($\approx 69^\circ\text{C}$, Table III), when compared to native corn starch ($\approx 62^\circ\text{C}$, Table I). Krueger et al (1987) suggested that annealing of starch caused a higher maximum temperature of gelatinization. It is also possible that this increase in temperature is solely due to the fact that the starches being compared are from different sources.

Gomez et al (1989) observed that crystallinity of corn starch decreased during tortilla processing. However, they did not find differences in A-type diffraction patterns of nixtamal and masa even though nixtamal was exposed to high mechanical shearing and warm temperatures during grinding into masa. Figure 3 shows A-type diffraction patterns for nixtamalized masa after being preserved by liquid nitrogen freeze-drying or ethanol dehydration. No significant differences were found in relative crystallinity (Table IV).

These findings support the hypothesis that neither liquid nitrogen freeze-drying nor ethanol dehydration caused significant alterations in starch molecular structures detectable by DSC or XRD.

CONCLUSIONS

Both liquid nitrogen freeze-drying and ethanol dehydration were effective methods for starch stabilization/preservation of properties detected during DSC and XRD analysis. These procedures can be used to preserve starch molecular structures at a specific point in time. Ethanol dehydration does not need sophisticated equipment but involves several analytical steps that logistically complicate evaluation of sequential samples. Running multiple tests on closely time-dependant samples requires increasing numbers of trained personnel. Liquid nitrogen freezing followed by freeze-drying is a simple and efficient procedure that allows studying samples taken at short time intervals. The method, however, requires a freeze drier and is limited by the freeze drier capacity. The starch preservation method to select will depend on the scope of the research project and resource availability.

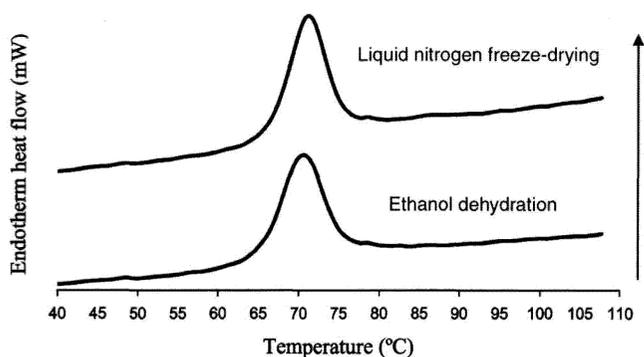


Fig. 2. Representative DSC endotherm for nixtamalized masa after being preserved by liquid nitrogen-freeze drying and ethanol dehydration.

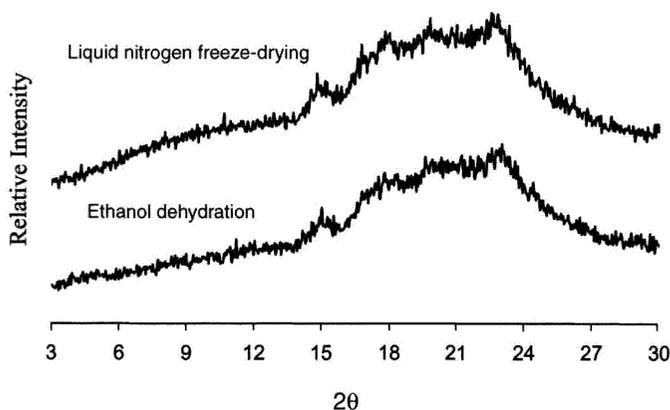


Fig. 3. Representative X-ray A-type diffraction pattern for nixtamalized masa after being stabilized and preserved by liquid nitrogen freeze-drying and ethanol dehydration.

TABLE III
DSC Thermal Properties Means, Standard Errors, and *P* Values of Nixtamalized Masa by Method of Preservation^a

Treatment	Temperature ($^\circ\text{C}$)			Enthalpy (J/g)
	Onset	Peak	End	
Liquid nitrogen-freeze drying	65.24	69.72	74.37	6.12
Ethanol dehydration	65.12	69.83	75.00	6.63
Standard error	0.21	0.25	0.39	0.62
<i>P</i> value	0.7246	0.7654	0.3687	0.6197

^a Mean values for three treatment replicates.

TABLE IV
Means, Standard Error, and *P* Value for Relative Crystallinity of Nixtamalized Masa by Method of Preservation^a

Treatment	Relative Crystallinity (%)
Liquid nitrogen-freeze drying	30.05
Ethanol dehydration	27.72
Standard error	0.92
<i>P</i> value	0.6633

^a Mean values for three treatment replicates.

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