UDP-glucose Dehydrogenase Activity and Optimal Downstream Cellular Function Require Dynamic Reorganization at the Dimer-Dimer Subunit Interfaces

Annastasia S. Hyde
University of Nebraska-Lincoln

Ashley M. Thelen
University of Nebraska-Lincoln

Joseph J. Barycki
University of Nebraska-Lincoln, jbaryckii2@unl.edu

Melanie A. Simpson
University of Nebraska - Lincoln, msimpson2@unl.edu

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Annastasia S. Hyde, Ashley M. Thelen, Joseph J. Barycki and Melanie A. Simpson

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UDP-glucose Dehydrogenase Activity and Optimal Downstream Cellular Function Require Dynamic Reorganization at the Dimer-Dimer Subunit Interfaces


Department of Biochemistry, University of Nebraska, Lincoln, NE 68588-0664.

Running title: UDP-glucose dehydrogenase quaternary structure

*These authors contributed equally to this work.

Address correspondence to: Dr. Melanie A. Simpson, Department of Biochemistry, University of Nebraska, N246 Beadle Center, Lincoln, NE 68588-0664; tel (402) 472-9309; fax (402) 472-7842; email: msimpson2@unl.edu

OR Dr. Joseph J. Barycki, N114 Beadle Center, Lincoln, NE 68588-0664; tel (402) 472-9307; fax (402) 472-7842; email: jbarycki2@unl.edu

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Background: UDP-glucose dehydrogenase (UGDH) mutants were engineered to perturb hexamer:dimer quaternary structure equilibrium.

Results: Dimeric species of UGDH have reduced activity in vitro and in supporting hyaluronan production by cultured cells.

Conclusion: Only dynamic UGDH hexamers support robust cellular function.

Significance: Manipulation of UGDH activity by hexamer stabilization may offer new therapeutic options in cancer and other pathologies.

SUMMARY

UDP-glucose dehydrogenase (UGDH) provides precursors for steroid elimination, hyaluronan production, and glycosaminoglycan synthesis. The wild-type UGDH enzyme purifies in a hexamer-dimer equilibrium, and transiently undergoes dynamic motion that exposes the dimer-dimer interface during catalysis. In the current study, we created and characterized point mutations that yielded exclusively dimeric species (obligate dimer, T325D), dimeric species that could be induced to form hexamers in the ternary complex with substrate and cofactor (T325A), and a previously described exclusively hexameric species (UGDHA132), to investigate the role of quaternary structure in regulation of the enzyme. Characterization of the purified enzymes revealed a significant decrease in the enzymatic activity of the obligate dimer and hexamer mutants. Kinetic analysis of wild-type UGDH and the inducible hexamer, T325A, showed that upon increasing enzyme concentration, which favors the hexameric species, activity was modestly decreased and exhibited cooperativity. In contrast, cooperative kinetic behavior was not observed in the obligate dimer, T325D. These observations suggest that the regulation of the quaternary assembly of the enzyme is essential for optimal activity and allosteric regulation. Comparison of kinetic and thermal stability parameters among the hexameric wild-type enzyme and the engineered mutants revealed structurally-dependent properties consistent with a role for controlled assembly and disassembly of the hexamer in the regulation of UGDH. Finally, both T325A and T325D mutants were significantly less efficient in promoting downstream hyaluronan production by HEK293 cells. These data support a model that requires an operational dimer-hexamer equilibrium in order to function efficiently and preserve regulated activity in the cell.

UDP-glucuronate is an essential cellular metabolite that serves as the precursor for multiple...
diverse and compartmentalized processes. Notably, its production is critical in tissues such as liver, prostate, and breast for ER-localized phase II detoxification of lipophilic hormones and xenobiotics by glucuronidation (1). In adult tissues, in wound healing, and during development, high levels of UDP-glucuronate are used to synthesize extracellular glycosaminoglycans such as heparan sulfate and hyaluronan (HA)(2), and the decarboxylation of UDP-glucuronate to UDP-xylose initiates proteoglycan production in the Golgi (3). In embryonic development, temporal and spatial elevation in UDP-glucuronate synthesis are indispensable for rapid extrusion of HA to the extracellular space to effect the dramatic morphological changes that generate anatomical structures such as cardiac valves (4,5).

UDP-glucuronate is exclusively synthesized by UDP-glucose dehydrogenase (UGDH), which is an essential enzyme for circulatory system development in multiple organisms from nematodes to humans (4,6-11). In prostate tumor cell lines, UGDH is a constitutively expressed but androgen-stimulated gene product, upregulation of which can specifically drive excess steroid elimination through glucuronidation in hormone dependent cells (12). In the developing heart, UGDH point mutations that alter the integrity of quaternary structure, normally hexameric, fail to support maximal HA production that is needed for cardiac valve formation (4). Clinical and cellular consequences associated with UGDH defective function implicate stable but dynamic association of subunits, and suggest useful therapeutic applications would focus on sustaining robust UGDH activity.

Several reports have provided structural and enzymological insights into UGDH function in multiple species (6,7,10,11,13-26). In all cases, the enzyme oxidizes UDP-glucose through two NAD⁺-dependent electron transfers. Of particular interest to studies relating quaternary structure to cellular function is that bacterial UGDH adopts a strictly dimeric conformation in the presence or absence of its substrate and cofactor (27), whereas higher eukaryotic UGDH orthologues purify in a hexamer-dimer equilibrium that is stabilized to hexameric association under conditions representative of the cellular environment (4,11,16,28). Kinetic data comparing wild-type UGDH to an obligate (inactive) hexamer mutant in which residue 132 was deleted (25), together with the interesting observation of an “open” conformation of the wild-type enzyme observed in co-crystals with the potent UGDH inhibitor UDP-xylose, suggest that the transient capacity to dissociate and reorganize the hydrogen bond network at the interface between dimeric units is an important element of the enzyme’s normal catalytic cycle.

For the current study, we sought to determine the effects of quaternary structure on intrinsic and cellular functional properties of UGDH. Crystallographic data are available for several wild-type or point mutant UGDH forms (pdb codes 3ITK (29) and 3KHU (30)), including an apo structure, as well as ternary forms with each permutation of the abortive complex (NAD⁺ cofactor and UDP-glucuronate product, 2QG4; NADH cofactor and UDP-glucose substrate, 2Q3E (29)). These structural data provide a high resolution view of the non-covalent bonding network that sustains the subunit interface in the context of the hexamer (Fig 1). We used these coordinates to rationally generate point mutations expected to perturb the interface and alter the hexamer-dimer equilibrium without impacting catalytic activity directly. With these tools, we additionally evaluated whether the hexameric conformation is required for cellular function, and how its integrity relates to steady state activity. Our results demonstrate for the first time that the UGDH hexamer is more competent to support high levels of HA production than the dimer, and that intrinsic stability of the enzyme is conformationally dependent.

**EXPERIMENTAL PROCEDURES**

*Generation and purification of UGDH point mutants.* Molecular graphics and analyses were performed with the UCSF Chimera package (31), developed by the NIH-supported Resource for Biocomputing, Visualization, and Informatics at the University of California, San Francisco. Point mutants of human UGDH were generated from the wild type construct, UGDH-pET28a (for protein purification), using the QuikChange site-directed mutagenesis kit (Stratagene, La Jolla, CA) according to the manufacturer’s protocol. Sequences were verified by Eurofins MWG Operon (Huntsville, AL). Wild-type UGDH and
point mutants were expressed in *E. coli* strain Rosetta2(DE3)pLysS (EMD Biosciences, Inc., San Diego, CA), induced overnight at 18°C, and purified as N-terminal 6-His fusions essentially as previously described (11). Following nickel affinity chromatography, proteins were dialyzed against 20mM Tris-HCl, pH 7.4, and 1mM DTT. Protein was stored at a concentration of 2 mg/ml.

Enzymatic activity measurement and kinetic characterizations. Activity of the point mutants was assayed by monitoring the change in absorbance at 340 nm that accompanies reduction of NAD⁺ to NADH as reported previously (11). For standard measurement of enzymatic activity, the assay was done in a 96-well plate format at room temperature for 5 minutes in 0.1 M sodium phosphate buffer, pH 7.4. T325A and T325D mutants exhibited saturable kinetics, while the activity of the Δ132 mutant was found to be negligible. Michaelis constants $K_m$ and $V_{max}$, for UDP-glucose and NAD⁺, were determined independently by holding [NAD⁺] constant at 10 mM and varying [UDP-glucose] from 0-1 mM. Similarly, NAD⁺ measurements were made by holding [UDP-glucose] constant at 1 mM and varying [NAD⁺] from 0-10 mM. Specific activities were calculated from absorbance values using the molar extinction coefficient for NADH of 6220 M⁻¹cm⁻¹. Triplicate determinations for each concentration increment were plotted with PRISM (GraphPad Software, Inc., San Diego, CA). $K_m$ and $V_{max}$ were calculated by fitting the data to the Michaelis-Menten equation and assuming a single binding site per subunit for substrate and cofactor.

Stability of enzyme activity in vitro. Wild type and mutant UGDH (1mg/mL each in 0.1M sodium phosphate, pH 7.4) was incubated in the absence and presence of saturating NAD⁺ (10mM) and UDP-glucuronate (1mM). After a 1 minute equilibration at 37°C, aliquots were assayed for initial activity as above with saturating NAD⁺ and UDP-glucose, and a final enzyme concentration of 670nM. For each time point, absorbances were measured in triplicate, divided by the average absorbance at time zero, and plotted as percent relative activity over time. Using PRISM software, the graphs were fitted to the one phase exponential decay equation in order to calculate half-lives at 37°C.

Molecular Mass Determination. Oligomeric state of the apo enzyme was determined by size exclusion chromatography as previously described (11). Each sample was precleared by centrifugation prior to loading. To determine the effect of substrate and cofactor inclusion on oligomeric state, purified recombinant UGDH wild-type and point mutants were individually loaded on a Superdex 200 HR 10/300 column (GE Healthcare Life Sciences, Piscataway, NJ) using a 250 μL loop and separated by FPLC in 1X PBS in the presence of 1 mM UDP-glucose and/or 5 mM NAD⁺ at a flow rate of 0.5 ml/min. Substrate and cofactor alone or together were included in each 250 μL load sample at the same final concentrations. Because of the interference from cofactor in UV absorbance readings from the column fractions of enzyme complexes, protein content of these samples was measured at 595 nm upon the addition of Bradford reagent. The standards used were: thyroglobulin, 699 kDa; ferritin, 440 kDa; albumin, 67 kDa; β-lactoglobulin, 35 kDa; RNase A, 13.7 kDa. Resolution was sufficient to calculate molecular masses unambiguously in multiples of 57 kDa (the calculated molecular mass of monomeric Histagged UGDH).

Thermal stability measurement. Purified recombinant UGDH wild-type and point mutants were analyzed using a modified version of the Thermofluor Stability Assay as described by Ericsson et al (32). Solutions of the wild-type and mutant apo enzymes (~5-20 μg of protein) were prepared in 1X PBS containing Sypro Orange dye (Invitrogen, 1:500 dilution). Samples containing substrate and/or cofactor were prepared in the same manner with UDP-glucose and NAD⁺ added to a final concentration of 1 mM and 5 mM respectively. All samples were handled at room temperature throughout the preparation and transferred to an iCycler MyiQ Thermocycler (Bio-Rad) for analysis. The assay protocol was executed with the block initially at 20°C. After addition of samples, temperature was raised over a period of 76 min from 20°C to 95°C in 0.5°C increments. The change in fluorescence was monitored for each sample at an excitation and emission of 490 and 575 nm respectively. Transitions representing the melting temperature, $T_m$, were obtained from the derivative of fluorescence with respect to temperature for at least 3-10 replicates, and plotted as the mean ±
Proteolytic sensitivity of UGDH. Purified recombinant UGDH wild-type and point mutants (10 μg protein) were digested with 10 ng of trypsin in 1x PBS for 2.5 hours in the absence or presence of 1mM UDP-glucose, 1mM UDP-glucuronate, 5mM NAD+, 5 mM NADH, or abortive ternary complexes (at saturating conditions). Samples were analyzed by reducing SDS-PAGE.

Cell Culture and Transfection. HEK293 cells were cultured in DMEM supplemented with 10% fetal bovine serum (FBS, Fisher), and seeded in a 6-well plate 24 hours prior to transfection. Cells were transiently co-transfected at 80% confluence using Fugene HD (Roche) with hyaluronan synthase 3 (HAS3) and wild-type or mutant UGDH constructs in the pIRES2-EGFP eukaryotic expression vector. Cells were harvested 48 hours post-transfection in 1X RIPA buffer and incubated on ice for 10 minutes. The whole cell lysate was separated by centrifugation.

HA Quantification and Western Analysis. HA content of cell culture conditioned media was determined by a competitive binding assay (23). Human umbilical cord HA (Sigma) was coupled to Covalink 96-well plates at 100 μg/ml in 1X PBS in the presence of 25 μg SulfoNHS/EDC. Plates were rocked for 1.5 h at room temperature, then washed twice with Pierce SuperBlock prior to sample incubation. Serial dilutions of cell culture media were incubated with biotinylated HA binding protein (bHABP, Calbiochem) at a final concentration of 1 μg/mL. Standards and samples were incubated in the HA-precoated wells at room temperature overnight. Excess HA was removed with four washes of PBS/0.05% Tween 20. The plate was developed using an avidin-biotin HRP system (Vectastain) with TMB as substrate, and absorbance was measured at 650 nm. HA concentration was interpolated from a standard curve and normalized to total protein. Western analysis of whole transfected cell lysates was performed as previously described (24). Primary antibodies were used at a dilution of 1:1000 for α-UGDH and α-FLAG, and 1:750,000 for α-tubulin.

RESULTS

Kinetic comparison of wild-type and mutant UGDH. To determine the role of quaternary structure in the activity and properties of UGDH, we designed, expressed, and purified mutant species of the enzyme that were expected to exhibit disruption from primarily hexameric to exclusively dimeric association. Initially, we selected glutamate 110 and threonine 325 because of their pivotal functions in dimer-dimer association as shown in the ribbon diagram of the ternary complex structure (Fig 1). The E110A mutant, although dimeric in the apo form, exhibited only about 50% reduction in V max, but was highly unstable in solution and in cultured cells, so it could not be evaluated unambiguously. The carboxylate side chain of this residue on one subunit forms salt bridges with two positively charged moieties, one of which is on the adjacent subunit (K329) and one of which stabilizes the same helix that contains E110 (R114, Fig 1).

Therefore, the loss of E110 not only disrupts a direct series of interactions between subunits, but also within individual subunits, since this charge likely screens electrostatic repulsion between the two positive charges. This may account for its instability. Instead, we pursued the other highly networked interface residue, T325.

In the apo form, T325 directly forms a hydrogen bond with D105 of the opposite subunit (Fig 1B). A similar view of the interface in the abortive ternary complex structure illustrates a conformational change that reorganizes the hydrogen bond network such that T325 and D105 now coordinate through a highly ordered water molecule (Fig 1C). T325 residue was altered to an alanine because it would be incapable of hydrogen bonding, and to aspartate, which would introduce a new formal negative charge at the interface. Both mutants expressed abundant soluble protein, similarly to the wild-type enzyme, and exhibited saturable kinetics. Data were fitted to the Michaelis-Menten equation to calculate steady state kinetic constants and evaluate potential cooperativity. Wild-type UGDH and the T325A mutant exhibited similar V max and K m values with respect to both substrate and cofactor when examined under standard assay conditions (Table 1). However, the T325D mutant exhibited approximately 5-fold reduced V max relative to the wild-type UGDH. Interestingly, when the wild-type and T325A mutant enzyme were assayed at higher concentrations, previously reported to yield sufficient NADH to inhibit activity (K i for NADH = 27 μM (21)), the turnover and catalytic
efficiency of both were significantly reduced (Fig 2). In contrast, the efficiency of the T325D mutant was not altered over this enzyme concentration range, despite accumulation of NADH to the previously reported inhibitory levels and an unaltered apparent $K_m$ for NAD$^+$ relative to the wild-type enzyme. These results suggest the perturbation of the dimer interface by charge introduction eliminates negative cooperativity and product inhibition.

**Half-life of UGDH catalytic activity in vitro is reduced by mutations at the dimer interface.** We previously observed a loss of stability in the catalytic activity of UGDH when quaternary structure was disrupted by mutations that were found in human congenital heart defects. However, we were not able to assert unambiguously that the loss of activity was attributable to the effects of those mutations on UGDH quaternary structure. The T325 mutants afforded this opportunity, since they were rationally designed to disrupt hexamer assembly without directly affecting the integrity of the active site. In addition, we had not tested the effect of substrate and cofactor concentrations that simulated cellular conditions on overall duration of the enzyme’s catalytic competence. Purified wild-type or mutant UGDH was incubated with or without NAD$^+$ and UDP-glucuronate, an abortive ternary complex, and activity was monitored over time. Formation of the wild-type ternary complex extended the half-life of enzymatic activity $\approx 30\%$ from 59h to 88h (Fig 3A). The activity of the T325A mutant was the most labile, with a half-life of only 24h that was not extended significantly by substrate and cofactor addition (Fig 3B). Interestingly, UGDH T325D retained its activity similarly to the wild-type enzyme but did not exhibit increased stability in the abortive ternary complex (Fig 3C).

**UGDH quaternary structure disrupted by mutagenesis of T325 is differentially restored by ternary complex formation.** We directly assessed the quaternary structure of the T325 mutants by size exclusion chromatography. As expected, the wild-type enzyme purified in a hexamer-dimer equilibrium that was stabilized to the hexamer conformation by inclusion of saturating NAD$^+$ cofactor and UDP-glucose substrate in the column running buffer. Moreover, we further found that the presence of either substrate or cofactor alone was also able to shift the eluted peak to the higher mass profile of the hexamer (Fig 4). As a control, we generated an additional point mutant, Δ132, in which valine 132 was deleted and which had been reported to be exclusively hexameric (25). Consistent with the prior results, this mutant had undetectable activity (Table 1) and eluted at the molecular mass of the hexamer, irrespective of substrate or cofactor addition. In contrast, both T325A and T325D mutants were exclusively dimeric in the apo form, indicating the loss of the neutral hydrogen bond network formed by T325 was sufficient to disrupt the dimer-dimer association interface. However, the inclusion of either substrate or cofactor alone with UGDH T325A partially shifted the dimer equilibrium to higher molecular mass complexes, and addition of both to form the ternary complex fully shifted the peak to the molecular mass of the hexamer. Neither the presence of substrate or cofactor alone, nor the inclusion of both together, affected any change in the apparent molecular mass of the T325D mutant, which retained its dimeric structure. Thus, we termed the UGDH mutant Δ132 an obligate hexamer, T325A an inducible hexamer, and T325D an obligate dimer.

**Intrinsic thermal stability and resistance to proteolysis are reduced by inability of UGDH to hexamerize.** With these tools in hand, we sought to assess whether structural properties of the mutants were altered so that we could relate the intrinsic characteristics to the activity of the enzymes in vitro and in vivo, and draw conclusions about the importance of hexameric conformation to cellular function. A thermal denaturation assay was previously used to show that the melting temperature ($T_m$) of the UGDH apoenzyme was increased by $20^\circ C$ in the presence of saturating substrate and cofactor to form the holoenzyme (4,29). We extended our prior characterization of the wild-type enzyme using this assay in the presence of either reduced or oxidized cofactor alone, substrate or product alone, and each permutation of the ternary complex (e.g.; UDP-glucose and NAD$^+$, UDP-glucuronate and NAD$^+$, etc.). As expected, the ternary complexes exhibited $\approx 20^\circ C$ higher $T_m$ relative to the wild-type UGDH apoenzyme (Fig 5, solid dark bars compared to the white bar). However, adding either cofactor, substrate, or product alone also increased thermal stability,
though to a lesser extent (≈10°C, Fig 5, patterned bars). Consistent with the maximal thermal stabilization being attributable to the ability to hexamerize, the UGDH Δ132 mutant had a $T_m$ equivalent to the wild-type ternary complex, with or without any additives. The $T_m$ of the UGDH T325A mutant was not altered by addition of substrate or cofactor alone, but was increased by ≈20°C in each of the ternary complexes, which would be predicted by its apparent hexameric conformation in the gel filtration assay. The T325D mutant, also as predicted, was not stabilized by substrate or cofactor alone in the thermal denaturation assay, but the $T_m$ increased ≈10°C in the ternary complexes, suggesting there is some intrinsic stability conferred even to the dimeric species that is independent of dimer-dimer association. However, this demonstrates that the hexameric wild-type UGDH enzyme has greater potential to be stabilized in the presence of its metabolites than does the obligate dimer mutant.

As an independent measure of structural integrity in vitro, we evaluated susceptibility of each wild-type and mutant UGDH species to limited trypsin proteolysis. We chose conditions in which the wild-type enzyme was almost completely degraded in the apo form (Fig 6, top, first two lanes). As observed in the thermal denaturation assay, addition of substrate, product, reduced or oxidized cofactor, or each permutation of ternary complex components fully protected the wild-type enzyme against proteolysis. Similarly, the obligate hexamer UGDH Δ132 was minimally susceptible to trypsinization in all conditions. The inducible hexamer, T325A, was fully degraded in the apo form and in the presence of substrate alone. Addition of NAD$^+$ stabilized the enzyme fully, and either NADH or UDP-glucurionate partially protected, while the ternary complexes were almost completely undigested. These results are consistent with the T325A mutant’s ability to adopt a hexameric conformation that is thermally and proteolytically stabilized to an increasing degree upon cofactor and substrate addition. Not surprisingly, the T325D obligate dimer mutant was fully degraded by trypsin in the apo form and in the presence of substrate, product or cofactor individually. Consistent with an increase in its intrinsic subunit stability though without the ability to hexamerize, the T325D mutant experienced only partial protection against trypsin digestion in the ternary complex.

**Cellular function of UGDH is dependent on hexameric stability.** Though cellular ratios of NAD$^+$/NADH and UDP-glucose/UDP-glucuronate are likely to undergo considerable flux depending upon local or systemic nutrient and oxygen status, the pool of NAD cofactors in a given cell type is relatively fixed. Therefore, we compared the effects of obligate and inducible quaternary structure on the ability to provide precursors for HA synthesis in a cellular context. HEK293 cells, which express very low endogenous UGDH and HA synthase (HAS), the enzyme that uses UDP-glucuronate to produce HA, were transfected with eukaryotic expression constructs for wild-type or mutant UGDH, with or without HAS3. Equivalent steady state expression of all UGDH species was observed in each transfection condition (Fig 7, upper panel). HA was quantified in the conditioned media of each transfectant and normalized to total protein in the transfected cell lysate analyzed by western in the upper panels. Concurrent expression of wild-type UGDH and HAS3 yielded ≈3-fold more HA production than transfection with HAS3 alone, as we have previously observed (12). Strikingly, expression of the T325A mutant with HAS3 was not fully effective in promoting HA synthesis, yielding approximately 60% less HA than the wild-type cotransfection. The UGDH T325D mutant did not support HA production in excess of HAS3 alone, and coexpression of HAS3 with the obligate hexamer Δ132 modestly but significantly suppressed HA production relative to HAS3 alone. Thus, the cellular activity of UGDH is maximized when its conformational sampling statistically favors the hexameric species but permits dynamic movements in quaternary structure. These results support a model in which the hexamer-dimer ratio of UGDH senses flux in the presence of cellular cofactor, substrate and product.

**DISCUSSION**

UGDH targeted disruption and loss of function polymorphisms that interfere with the enzyme’s quaternary structure and cellular stability have been implicated in cardiac malformation in multiple organisms (4,5,33,34). Moreover, the androgen-stimulated expression of UGDH is important for prostate epithelial androgen solubilization and elimination. Our goal
was to engineer obligatory forms of the enzyme that represented its two most stably sampled oligomeric states, while minimally disrupting overall or active site stability, to determine the mechanistic function of this hexamer-dimer equilibrium in vitro and in a cellular context. The engineered UGDH mutants allowed us to determine that the hexameric enzyme has only modestly greater maximal catalytic activity than the dimer, but is significantly more responsive to structural stabilization and activation by substrate and cofactor, and capable of supporting greater output of the extracellular matrix product HA. Importantly, however, the dimeric mutant retains significant activity and is considerably less sensitive to product inhibition by reduced cofactor.

Examination of the dimer-dimer subunit interface revealed an extensive network of charge interactions and hydrogen bonding that coordinately stabilize the hexamer in the presence and absence of its cofactor or substrate. Besides T325, we had also identified a glutamate residue, E110, as a critical contributor to non-covalent interactions. However, the intrinsic instability of the E110A mutant made it unsuitable for subsequent cellular studies. T325 is ideally positioned to accommodate reorganization of the interface hydrogen bonds that occurs as the active site shifts to form binary and ternary complexes, and exchange cofactors, during the catalytic cycle. Although elimination of the hydroxyl by substituting alanine for threonine was expected to prevent dimer-dimer association, stability of the hexamer in the ternary complex indicates that there was sufficient redundancy and flexibility in the network to prevent loss of activity. Placement of a negative charge was necessary and sufficient to prevent subunit association, since the new negative charge was proximal to both the D105 and E110 side chains of the adjacent subunit, which probably was electrostatically repulsive.

It was important to characterize the mutants sufficiently to ascertain that an obligate dimer had been achieved since we had previous mutants that purified in a dimeric form as an apoprotein and later we discovered that although they had activity, they also had restored hexameric structure in the active ternary complex. K339A purified as a dimer, but had been designed as a substrate binding mutant. Consistent with this, its $V_{\text{max}}$ was identical to the wild-type enzyme but the $K_m$ for UDP-glucose was increased several hundred-fold. We interpreted this to mean that the dimeric form of the enzyme was equally active, but we did not characterize the solution oligomerization state of the ternary complex. In contrast, a previously reported mutant, K94E, was described as a dimer with 100-fold reduced activity, and was used to argue that the dimeric species was inactive (26). The $K_m$ for both substrate and cofactor was also elevated 1-2 orders of magnitude, so the mutant was severely compromised, but the conformation of the ternary complex was also not verified.

Inspection of the positioning of K94 at the dimer-dimer interface reveals that its side chain directly forms hydrogen bonds with the E360 side chain and the M98 backbone carbonyl of the opposite subunit, and that within its own subunit, its backbone amide and carbonyl hydrogen bond with the backbone carbonyl of A103 and the backbone amide of G101, respectively. Significantly increased disorder is evident for residues 88-110. Thus, residue K94 is critically involved in stabilization of a major element of secondary structure that directly communicates with the active site. For these reasons, the K94E dimer did not unambiguously address the relation of UGDH quaternary structure to its function in vitro and is also not suitable for studies in vivo. Our finding that substrate, product, and/or cofactor addition to either the wild-type or T325A mutant UGDH could partially or completely shift the structural equilibrium to the hexameric species led us to examine multiple features of the enzyme that could report on its structural integrity in reaction conditions. In particular, we wanted to identify a mutant that was minimally perturbed on an overall structural level other than being unable to oligomerize. In doing so, we affirmed that the dimeric species had reduced catalytic activity but not as a consequence of altered substrate or cofactor affinity in the context of the reaction.

Gel filtration shows UDP-glucose alone and NAD alone stabilize the wild-type hexamer. However, the T325A mutant was only induced to form a hexamer if both substrate and cofactor were present. In addition, induction of T325A hexamer formation required concentrations of substrate and cofactor that were saturating for wild-type UGDH and for maximal catalytic rate in the T325A mutant. In these conditions, the hexameric conformation of T325A is then fully stabilized. In
contrast, the obligate dimer remains dimeric in all conditions. When we compare to thermal stability, there is a strong increase in melting temperature of the wild-type when substrate, product, or cofactor are added alone, and an additional increase when either a productive or an abortive ternary complex is formed. An interpretation of increased stability in the hexameric form is further supported by the observed protection from proteolysis, with all of the binary and ternary permutations tested, as well as in the obligate hexameric mutant. Results of proteolysis and thermal denaturation further support only a partial restoration in complex formation and its resultant increase in stability for the T325A mutant as indicated in gel filtration, and suggest greater intrinsic structural stability is conferred by cofactor binding than by UDP-glucose binding. Also consistently, the ternary complexes of T325A and wild-type UGDH are equally stable when assessed both by thermal denaturation and limited proteolysis. The obligate dimer shows no intrinsic increase in thermal or proteolytic resistance except in the ternary complex, which suggests that approximately half the increase in quantified stability parameters is due to stability of the dimer, and half is the result of further ability to hexamerize. Moreover, since the UDP-glucose substrate and UDP-glucuronate product minimally protected the T325A mutant against proteolysis, the stability conferred in the dimeric ternary complex likely derives from the binding of cofactor following substrate association.

Previous kinetic data have shown that substrate binds first, with cofactor more surface exposed to facilitate exchange during catalysis (30). This is consistent with the above proteolytic results, and together with the observed product inhibition in the hexameric form, suggests that the availability of the substrate may be a key regulatory condition. The cooperative rate increase of the enzyme in the presence of increasing substrate is evidence that a conformational shift may precede achievement of full catalytic potential. In addition, it has been suggested that the dynamic hexamer is the relevant catalytic species on the basis of relatively low activity in both the K94E dimer and the A132 obligate hexamer. Our results with a single substitution at position T325 are consistent with lower activity of the obligate dimer, but with a more modest reduction in turnover than in K94E. This retention of significant catalytic activity in the less perturbed T325D dimeric mutant suggests the dimeric species may also be catalytically relevant in some cellular conditions, particularly in the presence of product inhibition, but does not negate the more significant role of the hexamer during maximum cellular demand.

The wild-type enzyme overexpressed in cells transfected with HA synthase, which exhibit a high demand for HA production, generates a 3-fold increase in achievable steady state HA levels. In contrast, the T325D obligate dimeric and A132 obligate hexameric enzyme species were unable to support the acute increase in cellular demand for HA precursors to any detectable extent, which suggests that in the cellular context, neither a dimeric nor a hexameric conformation can be exclusively considered the catalytically relevant species. This is not surprising, since the transiently dissociable interface of the wild-type enzyme is likely essential for cofactor exchange between the two oxidation events that occur in a single turnover. However, it is interesting that the T325A mutant has equivalent steady state kinetic parameters to the wild-type UGDH enzyme and a comparable ternary complex stability, but demonstrates reduced support for HA production. We attribute this to the increased lability of the dimer-dimer interface in cellular conditions that normally support activity, since the individual substrate/cofactor dissociation events leave the subunits more susceptible to loss of maximally active hexameric conformation. The equilibrium between hexamer and dimer presumably does not imply full dissociation in the cell but rather involves a spectrum of enzyme species with transiently exposed regions of the dimer-dimer interface. At cellular levels of substrate, cofactor, and product, the exposure of the wild-type enzyme interface is likely to be low frequency, in consideration of our observation that these individual components stabilize all properties we measured for the hexameric conformation. Since the T325A mutant is not fully stabilized to a hexamer when only one of these components is present, and is minimally stabilized if any component is subsaturating, this mutant may be considerably less functional in the cell as a result of increased dissociation. One additional
explanation may be that the subunit interfaces of the intact hexamer versus the hexamer with a transiently dissociated but catalytically engaged dimer interface may interact differently with other cellular regulatory factors. Such interactions could covalently or non-covalently modify the subunit interface to favor or discourage subunit association and thereby alter activity of the enzyme.

REFERENCES


FOOTNOTES
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Figure 1. Overall diagram of UGDH and the molecular interactions at the subunit interface. (A) The human hexameric enzyme is depicted in a ribbon representation of the ternary complex. Active sites contain the substrate UDP-glucose and the reduced cofactor NADH, which are shown in ball and stick form. Residue T325, which was mutagenized in this study, is shown in space-filling form and colored gold. Individual subunits are colored in dimeric pairs of dark/light blue, dark/light green, and dark/light purple. The closeup views show proximal ordered water molecules and hydrogen bonded amino acid side chains that interact to maintain integrity of the subunit interface. Black lines indicate hydrogen bonds, nitrogen atoms are colored blue, and oxygen atoms are red. Ordered water molecules are depicted as red spheres. (B) Apoprotein interface, pdb 3itk. (C) Abortive ternary complex interface, pdb 2Q3E. Figure representations were made in Chimera (31).

Figure 2. Steady state kinetic analysis reveals selectively reduced catalytic activity and relief from product inhibition by perturbation of threonine 325. Purified recombinant human UGDH wild-type and T325 mutants were assayed for concentration dependence of (A) turnover number (kcat), (B) K_m, and (C) catalytic efficiency (kcat/Km) with respect to UDP-glucose in the presence of saturating NAD^+.

Figure 3. Half-life of wild-type but not mutant UGDH activity in vitro is stabilized by formation of a ternary complex. Purified proteins were placed at 37°C in the absence (apo, circles, solid line) or presence (holo, squares, dotted line) of UDP-glucuronate and NAD^+. At the indicated time points, an aliquot was removed and activity was assayed by addition of the enzyme to phosphate buffer containing saturating concentrations of NAD^+ and UDP-glucose. Values plotted are the mean ± SD for triplicate determinations. Absolute turnover was measured as the change in NADH absorbance at 340 nm in a 1-min assay. Absorbance values were normalized to the value at time 0 to illustrate the rate of activity loss. Half-lives were calculated from curve fits of the data from triplicate assays.

Figure 4. Quaternary structure of wild-type UGDH and T325A is stabilized from dimeric to hexameric by substrate and cofactor while T325D remains an obligatory dimer. Purified wild type and mutant UGDH was fractionated by gel filtration FPLC in the absence (dark solid line) or presence of NAD^+ alone (gray hatched line), UDP-glucose alone (gray dotted line), or both (dark hatched line). Absorbance was monitored at 280 nm for the apoproteins (left ordinate axis) and 595 nm for the holoproteins (right ordinate axis). Size determination of each peak was made by comparison to molecular weight standards. Representative traces are plotted for wild-type UGDH, Δ132, T325A, and T325D. On each plot, the peaks corresponding to hexamer and dimer are labeled H and D, respectively.

Figure 5. UGDH is thermally stabilized by addition of substrate and cofactor but thermal stability is compromised by mutations that occur at the dimer interface. Purified wild type UGDH and point mutants were thermally denatured in the presence of Sypro Orange by increasing temperature from 20°C to 95°C in 0.5 °C increments. The change in fluorescence with respect to temperature was used to calculate a melting point (T_m) for each enzyme species in the absence (white bars) or presence of 5 mM NAD^+ or NADH, 1 mM UDP-glucose or UDP-glucuronate, and each of three ternary complex combinations, as indicated in the legend. Mean ± SD is plotted for the T_m of each enzyme.

Figure 6. Differential sensitivity to limited proteolysis reveals altered conformation conferred to the mutant UGDH enzymes by impaired ability to assemble in hexamers. Purified wild-type and mutant UGDH was prepared in the absence or presence of the indicated substrate and/or cofactor combinations and then treated with trypsin and analyzed by SDS-PAGE. Results are representative of at least three
different experimental replicates.

Figure 7. Loss of UGDH intrinsic activity impacts cellular HA production. HEK293 cells were cotransfected with constructs encoding GFP (vector control) and human HAS3, or GFP and wild-type or mutant UGDH, or with HAS3 and each UGDH construct. After 48 hours, conditioned media were analyzed for HA content and cell lysates were analyzed by western for UGDH, HAS3 (Flag), or tubulin (control). HA was calculated in units of mass per volume of conditioned medium, and normalized to protein content (in mg) of the whole cell lysates. Mean ± SEM is plotted for quadruplicate measurements repeated three times. Statistical significance was evaluated by Student’s t-test; * p<0.01, ** p<0.05, relative to HAS3 only. Western data are shown for the relevant lysates.
Table 1. Summary of wild type and mutant UGDH kinetic constants.

<table>
<thead>
<tr>
<th>Enzyme</th>
<th>UDP-Glucose</th>
<th>NAD⁺</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Vmax</td>
<td>Km</td>
</tr>
<tr>
<td></td>
<td>nmol/min/mg</td>
<td>μM</td>
</tr>
<tr>
<td>Wild-type</td>
<td>313 ± 8</td>
<td>34 ± 3</td>
</tr>
<tr>
<td>T325A</td>
<td>324 ± 8</td>
<td>48 ± 4</td>
</tr>
<tr>
<td>T325D</td>
<td>65 ± 1</td>
<td>15 ± 1</td>
</tr>
<tr>
<td>Δ132</td>
<td>ND</td>
<td>ND</td>
</tr>
</tbody>
</table>

ND, not detectable
Figure 1
Figure 2

A

B

C
Figure 3

A

B

C

Relative Activity

Relative Activity

Relative Activity

Time (hours)

Time (hours)

Time (hours)

$\tau_{1/2} = 88\, \text{h}$

$\tau_{1/2} = 26\, \text{h}$

$\tau_{1/2} = 49\, \text{h}$

$\tau_{1/2} = 59\, \text{h}$

$\tau_{1/2} = 24\, \text{h}$

$\tau_{1/2} = 48\, \text{h}$
Figure 4

- Wild-type
- Δ132
- T325A
- T325D

Legend:
- Apo
- NAD$^+$
- UDP-glucose
- Holo
Figure 7