

University of Nebraska - Lincoln

DigitalCommons@University of Nebraska - Lincoln

Agronomy & Horticulture -- Faculty Publications

Agronomy and Horticulture Department

1988

Colorimetric Determination of Cyanide in Enzyme-Hydrolyzed Extracts of Dried Sorghum Leaves

Francis A. Haskins

University of Nebraska-Lincoln, fhaskins@neb.rr.com

Herman J. Gorz

University of Nebraska-Lincoln

Robert M. Hill

University of Nebraska-Lincoln

Follow this and additional works at: <https://digitalcommons.unl.edu/agronomyfacpub>



Part of the [Plant Sciences Commons](#)

Haskins, Francis A.; Gorz, Herman J.; and Hill, Robert M., "Colorimetric Determination of Cyanide in Enzyme-Hydrolyzed Extracts of Dried Sorghum Leaves" (1988). *Agronomy & Horticulture -- Faculty Publications*. 179.

<https://digitalcommons.unl.edu/agronomyfacpub/179>

This Article is brought to you for free and open access by the Agronomy and Horticulture Department at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Agronomy & Horticulture -- Faculty Publications by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.

- Kolb, B. "Quantitative Analysis with Dynamic Headspace Gas Chromatography". *Chromatographia* 1982, 15, 587-592.
- Kolb, B.; Pospisil, P.; Auer, M. "Quantitative Analysis of Solid Samples; A Classification of Various Sample Types". *Chromatographia* 1984, 19, 113-122.
- Novak, J.; Zluticky, J.; Kuaberka, V.; Mostecky, J. "Analysis of Organic Constituents Present in Drinking Water". *J. Chromatogr.* 1973, 76, 45-50.
- Page, D. D.; Charbonneau, C. F. "Contamination of Several Breakfast Cereals by Methyl Chloroform". *J. Food Safety* 1978, 1, 129-136.
- Reinert, K. H.; Hunter, J. V.; Sabatino, T. "Dynamic Heated Headspace Analysis of Volatile Organic Compounds Present in Fish Tissue Samples". *J. Agric. Food Chem.* 1983, 31, 1057-1060.
- Rook, J. J. "Formation of Haloforms During Chlorination of Natural Waters". *Water Treat. Exam.* 1978, 23, 858-869.
- Suzuki, M.; Tsuge, S.; Takeuchi, T. "Gas Chromatographic Estimation of Occluded Solvents in Adhesive Tape by Periodic Introduction Method". *Anal. Chem.* 1970, 42, 1705-1709.
- Vitenberg, A. G.; Reznik, T. L. "Gas Chromatographic Headspace Analysis with Pneumatic Sampling". *J. Chromatogr.* 1984, 287, 15-27.

Received for review May 27, 1987. Revised manuscript received February 22, 1988. Accepted March 10, 1988.

Colorimetric Determination of Cyanide in Enzyme-Hydrolyzed Extracts of Dried Sorghum Leaves

Francis A. Haskins,* Herman J. Gorz, and Robert M. Hill

The purpose of this study was to develop a simple and effective procedure for assaying large numbers of sorghum leaf samples for their potential to release cyanide. The procedure involves drying the tissue at 75 °C (this was accomplished without loss of the cyanogenic glucoside dhurrin), grinding the dry tissue, extracting with water, digesting the extract with an almond meal extract to release cyanide, and determining cyanide in the digest colorimetrically. Results obtained were comparable to those based on distillation of digested samples and potentiometric determination of cyanide in the distillates. The equipment and supplies required for the procedure are relatively inexpensive, and a skilled technician can conveniently assay 70-80 samples/day.

Cyanide occurs in the leaves of sudangrass [*Sorghum bicolor* (L.) Moench] and sorghum (also *S. bicolor*) plants as the cyanogenic glucoside dhurrin [*p*-hydroxy-(*S*)-mandelonitrile β -D-glucopyranoside]. Degradation of dhurrin yields equimolar amounts of HCN, glucose, and *p*-hydroxybenzaldehyde (*p*-HB). One of the objectives of sudangrass and sorghum breeding programs is reduction in the level of dhurrin and thus in the possibility that cyanide released from the plant tissues will be harmful to consuming livestock. Plant-breeding and genetics programs typically require the examination of large numbers of individual plants. The HCN potentials (HCN-p) of large numbers of young sorghum seedlings can be assayed conveniently by autoclaving seedling leaves in water to extract and hydrolyze the dhurrin, diluting the extract in alkali, and reading the absorbance at 330 nm, the absorbance maximum of *p*-HB in basic solution, as described by Gorz et al. (1977). However, this simple procedure is not satisfactory for mature sorghum leaves (Haskins et al., 1984), and assays of mature leaves are required when measurements on the forage actually consumed by animals are needed. Blaedel et al. (1971) described an assay for cyanide in sudangrass forage in which emulsin was used to hydrolyze dhurrin, and cyanide in the hydrolysate was then determined directly with a cyanide-selective electrode. In our experience, however, the electrode was affected

adversely by constituents of the crude extracts; equilibration was slow, and misleading results were sometimes obtained.

The objective of this study was to adapt published procedures to provide for assay of the HCN-p of large numbers of samples with relatively inexpensive equipment and supplies. Specific information was sought on (1) the effect of tissue drying on HCN-p, (2) the feasibility of using an extract of almond meal to hydrolyze dhurrin in crude extracts of sorghum leaves, and (3) the suitability of the colorimetric procedure of Lambert et al. (1975) for determination of cyanide in these hydrolyzed extracts.

MATERIALS AND METHODS

Reagents. Succinimide, *N*-chlorosuccinimide, and barbituric acid for use in the procedure of Lambert et al. (1975) were obtained from Sigma Chemical Co., as was defatted almond meal. Other chemicals were obtained from customary sources.

Enzyme Preparation. Defatted almond meal was suspended in water (8 mg/mL), and the mixture was allowed to stand at room temperature for 3 h with occasional gentle shaking, after which it was filtered (Whatman No. 1 filter paper). The filtrate was used as the enzyme preparation. Such filtrates could be stored at 4 °C for at least 2 days without apparent loss in activity.

Plant Material. Week-old seedlings were grown in pans of a soil mixture in growth chambers at 27 °C under continuous cool white fluorescent light at about 150 $\mu\text{mol}/\text{m}^2$ per s, and shoots were excised just above the soil surface. For older plants from the field or greenhouse,

Departments of Agronomy (F.A.H., H.J.G.) and Biochemistry (R.M.H.), University of Nebraska, Lincoln, Nebraska 68583.

blades of upper leaves were excised and midribs were removed and discarded. Samples (usually less than 10 g) of the fresh tissue were placed in small baskets made of plastic screen and were oven-dried at 75 °C for about 3 h. The dry tissue was ground through a 1-mm screen and stored in tightly capped plastic vials at -18 °C prior to extraction for assay. The ground tissue was extracted as previously described (Gorz et al., 1986) by suspending weighed portions in water (usually 5 mg/mL) and shaking on a Gyrotory shaker for 2-3 h at room temperature. Suspensions were then filtered through Whatman No. 1 filter paper, and portions of the filtrates were used for assay.

Assay Procedures. Colorimetric Assay. The following procedure reflects the results of numerous preliminary experiments in which concentrations of enzyme preparation and plant extract and duration of digestion were varied. To 1 mL of tissue extract in a 20 × 150 mm test tube was added 1 mL of enzyme preparation, and the tube was capped with Parafilm and allowed to stand for 1.25 h at room temperature. The cap was then removed, and 8 mL of 0.1 M NaOH was added immediately. This basic solution was assayed for cyanide by the procedure of Lambert et al. (1975) as modified in the laboratory of Dr. E. E. Conn, University of California—Davis (personal communication). To 1 mL of the basic solution was added 0.5 mL of 1 M acetic acid followed by 5 mL of a solution containing 2.5 g of succinimide and 0.25 g of *N*-chloro-succinimide in 1 L of water. A 1-mL portion of a solution consisting of 6 g of barbituric acid, 30 mL of pyridine, and 70 mL of water was then added, and the tube was shaken vigorously and allowed to stand at room temperature for precisely 20 min. Absorbance at 580 nm was then read with a Spectronic 20 colorimeter.

Readings were made in sets of 20 or less, with each set including a colorimeter blank (1 mL of 0.1 M NaOH plus the color reagents), an enzyme control (1 mL of enzyme preparation plus 1 mL of water, digested, diluted, and sampled as indicated above), a standard of 1 mL of 3×10^{-5} M NaCN plus the color reagents, and up to 17 samples for assay. Various modifications of this procedure, made for purposes of comparison, are indicated in Results and Discussion.

Potentiometric Assay. This procedure was patterned after the AOAC (1945) method for the determination of cyanide in various feeds and foods. A 5-mL portion of enzyme preparation was added to 50 mL of plant extract in a standard Kjeldahl flask. The flask was stoppered and allowed to stand at room temperature for 2 h. The digest was diluted to 400 mL with water, the diluted digest was boiled, and HCN in the distillate was trapped in 20 mL of 1 M NaOH. The distillate was diluted to 200 mL with water, and cyanide concentration was determined with a cyanide-selective electrode. For some samples, portions of the distillate also were assayed colorimetrically essentially as described above.

RESULTS AND DISCUSSION

Use of Almond Meal To Enhance Dhurrin Hydrolysis in Homogenates of Fresh Tissue. The AOAC (1945) procedure for assaying cyanogenic glucosides in feeds suggested that addition of small amounts of emulsin may be necessary to release HCN, but quantities of enzyme or tissue were not specified. Easty et al. (1971) and Blaedel et al. (1971) specified the quantities of emulsin (almond β -glucosidase) used to hydrolyze dhurrin in sudangrass leaf extracts, but their values were in terms of emulsin rather than almond meal. Our comparison utilized seven sets of four 5-g samples of fresh leaves (midribs removed) rep-

resenting different stages of development of greenhouse- and field-grown plants of the sorghum-sudangrass hybrid Redlan × Greenleaf and the grain sorghum line N32 (Ross et al., 1980). Each 5-g sample was homogenized in 150 mL of water. The homogenate was transferred to a Kjeldahl flask with 300 mL of water, and the flask was sealed. Two of the samples in each set were incubated at room temperature without added almond meal; 200 mg of almond meal was added to each of the other two samples before incubation. After a 2-h incubation, all samples were distilled into alkali and cyanide was determined with the selective electrode. For the seven sets, the ratio of HCN-p based on incubation without almond meal to the value obtained with almond meal digestion was 0.69 ± 0.04 (mean \pm SE). Endogenous enzymes apparently did not bring about complete hydrolysis of the dhurrin present in the leaf tissue, nor was the distillation process itself sufficient to effect the complete breakdown, even though dhurrin in aqueous solution is heat-labile (Gorz et al., 1977).

The influence of a second digestion with almond meal was investigated by cooling homogenates following the first distillation, introducing an additional 200-mg portion of almond meal, incubating 2 h at room temperature, and again distilling. In several trials, the amount of cyanide detected following the second digestion and distillation was less than 3% as great as that from the initial digestion and distillation.

Effect of Drying the Tissue prior to Assay. In this comparison, 5-g portions of fresh leaf tissue and an equivalent amount of tissue that had been oven-dried at 75 °C (with the quantities used, 2 h at this temperature was sufficient to dry the tissue) were used. Both fresh and dried samples were homogenized in water, digested with almond meal, and distilled for cyanide determination as described above. A total of 13 comparisons were made, using leaves from Redlan × Greenleaf and N32 plants of various ages. On the average, HCN-p of the dried samples was 1.10 (SE = 0.05) times as great as that of the corresponding fresh samples. On the basis of these and other (Haskins et al., 1984) results with oven-dried leaves, it appears that leaf tissue can be dried at temperatures of 65-85 °C with little or no loss in HCN-p. However, drying for 2 h at 95 °C resulted in about 20% loss in HCN-p.

The use of properly dried tissue for HCN-p assay reduces the danger of spoilage during transport and storage of samples. This is especially important when the assay of fresh samples is inconvenient or impossible. An added advantage of drying is that it permits easy grinding of the tissue, thereby improving the uniformity of subsamples drawn for extraction and assay.

Comparison of Colorimetric and Potentiometric Assays. Fifteen dried and ground samples of sorghum leaf tissue, chosen to represent a wide range in HCN-p, were subjected to extraction, the extracts were digested with almond meal, and the digests were assayed both colorimetrically and potentiometrically as described in Materials and Methods. Single extracts of each sample were prepared. Duplicate 50-mL portions of each extract were digested and distilled for potentiometric and colorimetric assays of the distillates, and duplicate 1-mL portions of each extract were digested for direct colorimetric assay of the digests. Agreement among the three sets of values (Table I) was generally good. Overall means were 1422, 1447, and 1510 $\mu\text{g/g}$ for colorimetric assay of crude digests, colorimetric assay of distillates, and potentiometric assay of distillates, respectively. Correlation coefficients were calculated for the three pairs of sets in Table I; all three

Table I. Hydrocyanic Acid Potential (HCN-p) of Dried and Ground Sorghum Leaves Determined by Colorimetric Assay of Crude Enzyme Digests and by Colorimetric and Potentiometric Assays of Distillates from Such Digests

no.	entry ^a designation	HCN-p ^b (μg/g dry wt) based on		
		crude digest, colorimetric	distillate	
			colorimetric	potentiometric
1	KS8	7	10	53
2	KS8	30	25	64
3	Colman × White Collier-1	424	389	370
4	Colman × White Collier-2	691	657	683
5	Colman	795	746	793
6	White Collier	894	778	785
7	Redlan × Greenleaf	380	355	343
8	Redlan × Greenleaf	419	417	443
9	Redlan × Greenleaf	820	860	899
10	Redlan × Greenleaf	1883	1890	2033
11	N32	936	902	977
12	N32	1109	1114	1165
13	N32	1235	1155	1177
14	N32	3884	3769	3850
15	Martin	7828	8645	9020
mean		1422	1447	1510

^aEntries 14 and 15 are shoots from 1-week-old chamber-grown seedlings; other entries are upper leaves from field-grown plants at various stages of development. ^bValues shown are means of duplicate determinations which, in most cases, were within 5% of each other.

values exceeded 0.99. On this basis, the direct colorimetric assay of crude digests appears to be satisfactory for HCN-p determination.

Hydrolysis with Almond Meal Compared to Hydrolysis with NaOH. Duplicate extracts of the 15 entries listed in Table I were prepared, and duplicate 1-mL portions of each extract were subjected to digestion with almond meal and colorimetric assay as described above. Also, duplicate 1-mL portions were diluted with 9 mL of 0.1 M NaOH, and initial and 3-h scans from 400 to 240 nm were done. Mao and Anderson (1965) reported that dhurrin in 0.067 M NaOH has a 21-min half-life; thus, the 3-h treatments should effect virtually complete hydrolysis of the glucoside. Calculations of HCN-p were based on (a) colorimetric assay of the alkaline hydrolysates for cyanide and (b) gain in absorbance of the solutions at 330 nm during the 3-h period of hydrolysis in NaOH.

Colorimetric assay of extracts following digestion with almond meal gave HCN-p values (Table II) that agreed well with the values shown in Table I, indicating that reproducible results were obtained with these extraction and assay procedures. Hydrolysis of the extracts with NaOH yielded much lower HCN-p values. This was especially true of the values based on A_{330} gain; the overall mean for these assays was 774 μg/g compared with 1413 μg/g for the colorimetric assays of enzyme-hydrolyzed extracts. Recovery of cyanide added to enzyme-hydrolyzed extracts and of cyanide and *p*-HB added to base-hydrolyzed extracts exceeded 95%; therefore, the lower values observed following hydrolysis with NaOH cannot be attributed to reduced recovery. Rather it appears that, in dilute NaOH, dhurrin reacted with some constituent(s) of the leaf extracts in such a way that dhurrin was lost without yielding HCN or *p*-HB. Sucrose and glucose have been observed to have such an effect on dhurrin, whereas glycine does not (Haskins et al., 1984). In the present study we observed that small amounts of ethanol had an effect similar to that of glucose and sucrose. Therefore,

Table II. Hydrocyanic Acid Potential (HCN-p) of Dried and Ground Sorghum Leaves Determined by Colorimetric Assay of Enzyme- and NaOH-Hydrolyzed Extracts and by Gain in Absorbance at 330 nm in NaOH-Hydrolyzed Extracts

entry ^a	HCN-p ^b (μg/g dry wt) based on		
	enzyme digest, colorimetric	NaOH hydrolysate	
		colorimetric	gain in A_{330}
1	10 ± 2	15 ± 2	0
2	42 ± 2	19 ± 1	17 ± 1
3	440 ± 5	129 ± 1	75 ± 2
4	699 ± 4	301 ± 3	206 ± 1
5	800 ± 7	284 ± 4	197 ± 3
6	915 ± 5	450 ± 2	330 ± 3
7	395 ± 5	110 ± 1	47 ± 1
8	396 ± 3	298 ± 3	155 ± 2
9	820 ± 4	442 ± 2	286 ± 2
10	1851 ± 17	1085 ± 8	710 ± 7
11	940 ± 4	524 ± 5	393 ± 1
12	1120 ± 23	640 ± 7	481 ± 3
13	1216 ± 10	655 ± 4	486 ± 3
14	3736 ± 53	2746 ± 17	2400 ± 13
15	7820 ± 39	6611 ± 64	5827 ± 13
mean	1413	954	774

^aSee Table I for designation of entries. ^bDuplicate determinations on duplicate extracts; thus, $n = 4$ for each mean ± SE.

ethanol should not be used to prepare sorghum leaf extracts that are to be subjected to alkaline hydrolysis prior to assay for cyanide.

Despite the large effects of hydrolysis and assay methods illustrated by the values in Table II, correlation coefficients between the sets of values all exceeded 0.99. Therefore, any of the three procedures used for these assays would probably be satisfactory for ranking a group of samples representing a wide range of HCN-p. For accuracy, however, the procedure utilizing almond meal is preferable to those employing alkali to hydrolyze dhurrin.

Other Precautions. To determine whether capping the test tubes during enzymatic hydrolysis of dhurrin was beneficial, 1-mL portions of six different sorghum leaf extracts were digested with 1-mL portions of almond meal extract in capped tubes as described in Materials and Methods, and six comparable tubes were left uncapped during digestion. Colorimetric assays yielded values for the uncapped digests that were 85–89% as high as those obtained for the capped digests. Evidently the Parafilm caps were effective in reducing loss of HCN during the digestion step.

Solutions of KCN and NaCN are stable at room temperature for at least 1 week, and this was also found to be true of the cyanide in distillates from sorghum leaf digests. However, when almond meal digests of sorghum leaf extracts were diluted with NaOH and allowed to stand at room temperature for 1 or 3 days before the colorimetric assay was done, losses in HCN-p were as high as 25 and 50%, respectively. Therefore, it is important to proceed with the colorimetric assay as soon as possible after the digests have been diluted with NaOH.

In the potentiometric assay described in Materials and Methods, the extra water for distillation should be added following the digestion, as indicated. When 50 mL of tissue extract was diluted to 400 mL prior to introduction of the enzyme preparation, the HCN-p values obtained were about 80–85% as high as those based on dilution following digestion.

Suitability of the Procedure for Large Numbers of Samples. Starting with dried and ground plant samples, a skilled technician can conveniently accomplish the required weighing, extraction, digestion, and colorimetric

determination on 70-80 samples per day. Thus, the described procedure combining enzymatic hydrolysis and colorimetric assay is suitable for research projects that generate substantial numbers of samples.

ABBREVIATIONS USED

H₂CN-p, hydrocyanic acid potential; *p*-HB, *p*-hydroxybenzaldehyde.

ACKNOWLEDGMENT

The excellent technical assistance of Carol Caha and John Toy is gratefully acknowledged.

Registry No. CN⁻, 57-12-5; dhurrin, 499-20-7.

LITERATURE CITED

- Association of Official Agricultural Chemists *Methods of Analysis*, 6th ed.; AOAC: Washington, DC, 1945; p 416.
- Blaedel, W. J.; Easty, D. B.; Anderson, L.; Farrell, T. R. "Potentiometric Determination of Cyanide with an Ion Selective Electrode: Application to Cyanogenic Glycosides in Sudan Grasses". *Anal. Chem.* 1971, 43, 890-894.
- Easty, D. B.; Blaedel, W. J.; Anderson, L. "Continuous Electrochemical Determination of Cyanide: Application to Cyanogenic Glycosides in Sudan Grass". *Anal. Chem.* 1971, 43, 509-514.
- Gorz, H. J.; Haag, W. L.; Specht, J. E.; Haskins, F. A. "Assay of *p*-Hydroxybenzaldehyde as a Measure of Hydrocyanic Acid

- Potential in Sorghums". *Crop Sci.* 1977, 17, 578-582.
- Gorz, H. J.; Haskins, F. A.; Vogel, K. P. "Inheritance of Dhurrin Content in Mature Sorghum Leaves". *Crop Sci.* 1986, 26, 65-67.
- Haskins, F. A.; Gorz, H. J.; Hill, R. M.; Youngquist, J. B. "Influence of Sample Treatment on Apparent Hydrocyanic Acid Potential of Sorghum Leaf Tissue". *Crop Sci.* 1984, 24, 1158-1163.
- Lambert, J. L.; Ramasamy, J.; Paukstelis, J. V. "Stable Reagents for the Colorimetric Determination of Cyanide by Modified König Reactions". *Anal. Chem.* 1975, 47, 916-918.
- Mao, C.-H.; Anderson, L. "Cyanogenesis in *Sorghum vulgare*. 2. Mechanism of the Alkaline Hydrolysis of Dhurrin (*p*-Hydroxymandelonitrile Glucoside)". *J. Org. Chem.* 1965, 30, 603-607.
- Ross, W. M.; Gorz, H. J.; Haskins, F. A.; Webster, O. J. "Registration of Ten Sorghum Parental Lines (Reg. No. PL49 to PL58)". *Crop Sci.* 1980, 20, 834.

Received for review September 22, 1987. Accepted March 29, 1988. Contribution from the USDA-ARS and the Departments of Agronomy and Biochemistry, Nebraska Agricultural Research Division. Published as Paper No. 8435, Journal Series, Nebraska Agricultural Research Division. Reference to a company or product name does not imply approval or recommendation of the product by the USDA or the University of Nebraska to the exclusion of others that may be suitable.

Resolution of Peanut Seed Proteins by High-Performance Liquid Chromatography

Sheikh M. Basha

Peanut (*Arachis hypogaea* L.) seed proteins were extracted from the defatted meal with 0.01 M sodium phosphate buffer, pH 7.0, containing 0.5 M NaCl and 0.05% NaN₃ and resolved into four major and six minor fractions by HPLC. The molecular weight of these fractions ranged from 80 000 to 480 000. By this technique, variation in protein composition was detected in peanut seeds of different genotypes, boiled seeds, and seeds of different maturities. The data showed that HPLC can be successfully used to detect qualitative and quantitative differences in peanut seed protein composition.

Peanut seed proteins have been broadly classified into arachin, con-arachin (globulins), and albumins (Johnson et al., 1950; Dawson, 1968; Neucere, 1969; Basha and Cherry, 1976). Arachin and con-arachin together comprise approximately 87% of the seed proteins (Irving et al., 1946). Classical methods for separation of peanut proteins include ammonium sulfate precipitation (Jones and Horn, 1930; Dawson, 1968), CaCl₂ precipitation (Tombs, 1965), cryoprecipitation (Neucere, 1969; Basha and Pancholy, 1982), NaBr precipitation (Shetty and Rao, 1974), and ion-exchange chromatography (Dechary et al., 1962; Cherry et al., 1973; Neucere and Conkerton 1978). However, these methods are tedious and time-consuming. Recently, Basha and Pancholy (1981) separated peanut proteins into 10 arachin and non-arachin fractions employing gel filtration on a Sephacryl S-300 column. However, this method requires about 2 days to achieve the desired resolution, making it less attractive for use in large-scale sample monitoring such as screening of germplasm, following

compositional changes due to environmental factors, and during processing. Hence, an HPLC method has been developed for seed protein fractionation and characterization that would greatly increase sample-screening capabilities. This paper describes an HPLC method that would yield a peanut seed protein pattern similar to a conventional gel filtration column and require only 15 min/analysis.

MATERIALS AND METHODS

Instrumentation. The HPLC system consisted of a Model 510 pump, a variable-wavelength Model 490 UV/vis detector, Model 840 data station, and a manual U6K injector (Waters Chromatography Division, Millipore Corp., Milford, MA). The 7.8 mm × 30 cm PROTEIN PAK 300 SW column was obtained from Waters.

Materials. Peanut seeds of different lines and species [*Arachis hypogaea* L. (Nambiquare, Jenkins Jumbo, T2376, Chico); *Arachis batizogaea*, *Arachis stenosperma*; *Arachis monticola*, *Arachis villosulicarpa*] were gifts from Drs. R. O. Hammons and D. W. Gorbet, University of Georgia and University of Florida, respectively. For the boiling study, 25 g of freshly harvested green peanuts (*A. hypogaea* L., cv. Florunner) was boiled for 0-120 min in

Peanut Research Laboratory, Division of Agricultural Sciences, Florida A&M University, Tallahassee, Florida 32307.