Phase Transitions in Mechanically Milled Mn-Al-C Permanent Magnets

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Abstract: Mn-Al powders were prepared by rapid solidification followed by high-energy mechanical milling. The rapid solidification resulted in single-phase ε. The milling was performed in both the ε phase and the τ phase, with the τ-phase formation accomplished through a heat treatment at 500 °C for 10 min. For the ε-milled samples, the conversion of the ε to the τ phase was accomplished after milling via the same heat treatment. Mechanical milling induced a significant increase in coercivity in both cases, reaching 4.5 kOe and 4.1 kOe, respectively, followed by a decrease upon further milling. The increase in coercivity was the result of grain refinement induced by the high-energy mechanical milling. Additionally, in both cases a loss in magnetization was observed. Milling in the ε phase showed a smaller decrease in the magnetization due to a higher content of the τ phase. The loss in magnetization was attributed to a stress-induced transition to the equilibrium phases, as no site disorder or oxidation was observed. Surfactant-assisted milling in oleic acid also improved coercivity, but in this case values reached >4 kOe and remained stable at least through 32 h of milling.

Keywords: permanent magnets; mechanical milling
1. Introduction

Mn-Al-based permanent magnets have attracted renewed attention as a permanent magnet material because of the supply-chain issues associated with rare earth elements [1]. The ferromagnetic τ-MnAl, which has the L1₀ structure, has relatively good intrinsic magnetic properties ($K_I = 1.7 \, \text{MJ/m}^3$, $M_s = 7.5 \, \text{kG}$ [2]), suggesting a potential maximum energy product of ~12 MGOe with appropriate microstructural development [1]. The L1₀ phase is metastable and forms from the ε phase, which is stable above 870 °C. The equilibrium phases at room temperature that normally result from the decomposition of the ε phase are γ₂ and β-Mn [3].

The ε phase can be stabilized at room temperature by rapid quenching from the ε phase field [4–6]. This can be accomplished after conventional solidification and annealing in the ε phase field, or by rapid solidification where the high cooling rate suppresses equilibrium phase formation. The compositional limit of the τ phase is essentially limited by the composition achievable in the ε phase, which is reported to be $0.40 < x_{\text{Al}} < 0.46$ [7]. The τ phase is usually formed via a moderate-temperature annealing (~500 °C), and the phase transformation proceeds via an ε' intermediate phase [8]; too high of annealing temperature results in decomposition to the γ₂ and β-Mn phases. The addition of C tends to increase the stability of the metastable τ phase, likely by suppressing formation of the equilibrium phases. The C addition to the L1₀ does not deleteriously affect the intrinsic properties, except for the Curie temperature, which is decreased from 392 °C to 331 °C with 1.7 atomic percent C.

As is the case with all magnetic materials, the microstructure, particularly the grain size, plays a critical role in maximizing the extrinsic magnetic properties, namely coercivity and remanence. Phase purity additionally impacts overall saturation (and, consequently, remanence). Rapid solidification by melt spinning bypasses the primary solidification of γ₂ and β-Mn phases, resulting in single-phase ε in the as-solidified case [5]. Further, upon conversion to the τ phase, the coercivity of melt spun τ-MnAl is reported to be 1.25 kOe [5]. Microstructural modification via mechanical milling results in a decrease in grain size, leading to increases in coercivity [9]. However, it causes formation of the β phase which decreases the magnetization.

2. Results and Discussion

X-ray diffraction of the as-melt spun material revealed single-phase ε, as all the diffraction peaks observed were indexed to the ε phase (Figure 1). The ε phase was effectively transformed to τ-MnAl after heat treating at 500 °C for 10 min, with all peaks indexing to the τ phase (Figure 1). Transmission electron microscopy showed a grain size of ~500 nm (Figure 2), while the corresponding selected area electron diffraction pattern (inset) indexed to the τ phase, with diffraction arising from multiple grains, consistent with the X-ray diffraction results. The high-energy mechanical milling thus was initiated on either single-phase ε (ε-milled samples) or single-phase τ (τ-milled samples). The ε-milled samples were transformed to τ using the above-mentioned heat treatment. No post-milling heat treatment was done on the τ-milled samples, unless specifically noted.

Mechanical milling of the initially τ phase samples resulted in a dramatic increase in the X-ray diffraction peak widths (Figure 3). Additionally, the appearance of additional diffraction maxima was observed between the two primary τ peaks (~45° 20). The major β-Mn peaks are in this region, but
the peaks are too diffuse to identify conclusively. After 3 h of milling, the τ peaks are very broad, and further high-energy mechanical milling results in amorphization. For the ε-milled samples, the peak widths of the τ phase (note that the sequence was milling in the ε phase, followed by conversion to τ at 500 °C for 10 min) were notably more narrow (Figure 4). The increased peak widths of the τ-milled samples were due to both grain size reduction and increased structural disorder induced by the mechanical milling (notably, strain). Conversely, the τ phase formed in the ε-milled samples would lack the structural disorder induced by milling, similar to nearly defect-free grains that result from recrystallization, and the breadth of the diffraction peaks would solely be due to grain size effects. A secondary phase also evolves in the ε-milled samples, and in this case the peaks were positively identified to be from β-Mn.

**Figure 1.** X-ray diffraction patterns for (a) as-spun Mn₅₄Al₄₃C₃, and (b) after heat treatment at 500 °C for 10 min.

![Figure 1](image)

**Figure 2.** Transmission electron micrograph of melt spun Al-Mn-C after conversion to τ (inset: Selected area electron diffraction pattern of the region).
In both ε-milling and τ-milling, the high-energy mechanical milling resulted in a significant increase in the coercivity (Figure 5). Also in both cases, the initial increase in coercivity was followed
by a subsequent decrease upon further milling (Figure 6). The maximum coercivity was similar in both cases (4.6 kOe for τ-phase milling, 4.1 kOe for ε-phase milling). Additionally, the maximum was at similar milling times (1 h and 1.5 h, respectively). However, the coercivity of the ε-milled samples did not decrease as dramatically with milling time. The coercivity decrease in the τ-milled samples is likely due to amorphization, where the amorphous phase is magnetically soft. In both cases, a dramatic loss of magnetization was also observed. The loss in magnetization is much more significant in the τ-milled samples, but is also dramatic in the ε-milled samples.

**Figure 5.** (a) Hysteresis loops for ε phase-milled samples for different milling times, after conversion of ε to τ; (b) Hysteresis loops for samples milled in the τ phase.

**Figure 6.** Relationship between coercivity and milling time and magnetization (at 8 kOe) and milling time for both the (a) ε milled (solid line), and (b) τ milled samples (dashed line).

The loss of magnetization with milling time may be attributed to the formation of the β-Mn phase, which reduced the phase fraction of the ferromagnetic τ phase. However, the loss in magnetization appeared more significant than the phase fraction of the β-Mn, particularly for the τ-milled sample. The loss of magnetization could possibly result from oxidation during milling, or from site disorder induced by the milling of the τ phase. The latter is essentially a loss of the long-range order, and the resulting antisite disorder resulting in antiferromagnetic coupling between Mn atoms along the c-direction.

If oxidation occurred during milling, then an effective change in Mn:Al ratio would result and the phase relationships would be altered. Thus, if a milled sample is annealed in the ε phase field (above 1050 °C), then the phase transformation to the β phase can be reversed back to the ε phase, and
ultimately form the $\tau$ phase. If oxidation occurred, then the composition would shift so that it would no longer be in the $\varepsilon$ single-phase field. Figure 7 shows a sample that was milled, annealed at 1050 °C for 10 min, and then annealed at 500 °C for 10 min. The XRD pattern revealed single-phase $\tau$, suggesting that no significant change in composition occurred that resulted in the formation of $\beta$-Mn during milling (Figure 5). Interestingly, the original coercivity was recovered upon conversion to $\tau$. Presumably, the grain size increased as a result of the high temperature heat treatment but with no effect on the coercivity. Additionally, energy dispersive X-ray spectroscopy revealed only trace oxygen levels up to 1 h of milling, although long-term milled samples (10 h) showed significant oxidation [6]. It should be noted that the mechanical milling protocol of the latter sample was different, notably that intermittent milling was not employed.

**Figure 7.** X-ray diffraction pattern (a) 1 h $\varepsilon$-milled sample, converted to $\tau$ showing significant phase fraction of $\beta$-Mn, and (b) same sample after annealing at 1050 °C to transform the material back to $\varepsilon$, and then conversion to $\tau$ at 500 °C for 10 min. It returns to single-phase $\tau$.

In order to determine if site disorder was responsible for the loss of magnetization, a sample was milled for 15 min and the intensity ratio $I_{(001)}/I_{(111)}$ was compared to the original intensity. If site disorder increased, then the intensity of the (001) superlattice peak would decrease (relative to the (111) fundamental peak). The peaks were fitted with a Gaussian fit to determine the area. Figure 8 shows the XRD data and the Gaussian fits. The intensity ratio of the 15 min milled sample was virtually identical to the non-milled sample (both $0.6 \pm 0.05$), indicating that site disorder did not increase upon milling. It should be noted that a ~15% loss in magnetization was observed even with the relatively short-term milling.

With neither oxidation nor site disorder contributing to the loss of magnetization, it appears that the formation of the equilibrium $\beta$-Mn (and presumably $\gamma_2$ below detection limits) is the predominant contributor. As mentioned, the nominal composition of the alloy lies in the equilibrium phase field of
β-Mn + γ2; thus any decomposition of the metastable τ phase, or equilibrium decomposition of the ε phase, would result in the formation of these phases. Since neither of these phases is ferromagnetic (or ferrimagnetic), their presence would reduce the phase fraction of the ferromagnetic τ phase, and lower magnetization. Since the high-energy mechanical milling was done using short term (1 min) intermittent “pulses” to alleviate temperature excursions, the transformation to the equilibrium phases is likely stress-induced. This is similar to what has been observed in other systems, where crystallization of amorphous structures (i.e., transformation from metastable to stable structures) was stress-induced [10]. The crystallization in this case is thought to be enabled by an increase in defect density [11]. However, it is also probable that some local heating occurred, even with the short-term milling, which assisted in the phase transformation.

Figure 8. X-ray diffraction data and Gaussian fits of the (001) and (111) peaks for (a) unmilled and (b) 15 min milled samples. The numbers shown are areas (in arbitrary units).

τ-phase samples were also mechanically milled using a surfactant (surfactant-assisted mechanically milling). The surfactant eliminates re-welding of fractured particles that occurs during dry milling. This consequently leads to a reduction in particle size [12–16]. Figure 9 shows X-ray diffraction patterns of the surfactant-assisted milled samples. Note that significant broadening occurs, but even at long milling times amorphization does not occur (compare this behavior with dry milling, shown in Figure 3). Also, the predominant phase remains τ, with a very small fraction of equilibrium phases. The coercivity behavior is similar to the dry-milled samples, with an increase with milling time.
(Figure 10). Here, however, the dramatic loss of coercivity observed in the dry-milled samples does not occur, as a relatively stable coercivity results even for times up to 32 h. The decrease in magnetization also occurs here, however, but the decrease is less dramatic than in the dry-milled τ phase samples, and more along the lines of the ε-milled samples. The morphology of the milled samples was interesting as well. After 16 h of milling, the particles consisted of very flat platelets with a tremendously high aspect ratio. The thickness of the plates after milling for 16 h is on the order of tens of nanometers, while the length and width are on the order of several microns (Figure 11). Presumably, the X-ray peak broadening arises due to preferential alignment of the particles during sample preparation for X-ray diffraction (i.e., they lie flat on the sample holder so size broadening arises from the particle thickness). Interestingly, the fracture of the particles appears to be mica-like (that is, in sheets). Typically, fracture occurs either transgranular or intergranular. The microstructure (Figure 2) consists of predominantly equiaxed grains, so the exact fracture behavior of the melt spun ribbon is not clear. This is in contrast to surfactant-assisted milling studies in other systems, which produced distinct nanoparticles after similar times [12–16].

**Figure 9.** X-ray diffraction patterns of surfactant-assisted mechanical milling (SAMM) as a function of milling time.
Figure 10. Relationship between surfactant-assisted milling time and coercivity.

![Graph showing the relationship between surfactant-assisted milling time and coercivity.]

Figure 11. Scanning electron micrograph showing the particle morphology of surfactant-assisted mechanically milled Mn-Al-C particles.

![Scanning electron micrograph showing the particle morphology of surfactant-assisted mechanically milled Mn-Al-C particles.]

3. Experimental Section

The composition of Mn$_{54}$Al$_{43}$C$_3$ was chosen to maximize τ phase fraction. Alloys were obtained by arc melting of high purity elements (>99.9% purity). The ingot was then melt spun at a tangential speed of 40 m/s. The materials were heat treated at 500 °C for 10 min after sealing under UHP Ar. Samples were quenched in water from the heat treatment temperature. Mechanical milling was done in a SPEX 8000 Mixer/Mill (SPEX Sample Prep, Metuchen, NJ, USA) using hardened steel vials and balls. The Mn-Al-C alloys were loaded into the vials and sealed in a nitrogen atmosphere glove box.
Milling was done on an intermittent basis of one minute on followed by one minute off to reduce temperature excursions. Surfactant-assisted mechanical milling was accomplished using oleic acid (powder-to-oleic acid ratio of 20:1) and heptane as the medium. In this case, the milling protocol was 10 min on, 5 min off. X-ray diffraction (XRD) analysis was done using a Rigaku Multiflex (Rigaku Corporation, Tokyo, Japan) with Cu Kα radiation and a post-sample graphite monochromator. Powdered samples were placed on an off-cut SiO₂ single crystal to minimize diffraction affects from the holder. Magnetic measurements were completed using a Lakeshore 8500 VSM (Lakeshore Cryotronics, Westerville, OH, USA) and were conducted at room temperature with a maximum field of 8 kOe.

4. Conclusions

Two-step processing involving melt spinning followed by high-energy mechanical milling results in increased coercivity but a loss of magnetization for Mn₅₄Al₄₃C₃ alloys. The milling was accomplished on material starting in both the ε phase, which is formed upon melt spinning, and the τ phase, formed upon heat treatment of ε at 500 °C for 10 min. While significant coercivity values were obtained (>4 kOe in both cases), a loss of magnetization with milling time was observed. The loss of magnetization was more significant for τ phase milling. In both cases the magnetization loss was attributed to stress-induced formation of room-temperature stable phases, notably β-Mn. No increase in site disorder was induced by short-term milling, and phase formation induced by milling was reversible. Surfactant-assisted milling resulted in platelet-like particles and similar increases in coercivity as observed in the dry milling.

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Author Contributions

Michael Lucis, Timothy Prost and Meiyu Wang conducted the experiments, collected data, and interpreted results. Xiujuan Jiang contributed the electron microscopy, including sample preparation, image and diffraction collection, and interpretation/analysis. Jeffrey Shield directed the research, interpreted results, and prepared the manuscript.

Conflicts of Interest

The authors declare no conflict of interest.

References


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