

9-17-2004

Kinetic Reaction Analysis of an Anhydride-Cured Thermoplastic Ep-oxy:PGE/NMA/BDMA

Wei Chian

Department of Chemistry and Chemical Engineering, South Dakota School of Mines, Rapid City, South Dakota 57701

Delmar C. Timm

University of Nebraska-Lincoln, dtimm1@unl.edu

Follow this and additional works at: <http://digitalcommons.unl.edu/chemengmolecular>



Part of the [Chemical Engineering Commons](#)

Chian, Wei and Timm, Delmar C., "Kinetic Reaction Analysis of an Anhydride-Cured Thermoplastic Ep-oxy:PGE/NMA/BDMA" (2004). *Papers in Molecular Chemistry*. 3.

<http://digitalcommons.unl.edu/chemengmolecular/3>

This Article is brought to you for free and open access by the Chemical and Biomolecular Engineering Research and Publications at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Papers in Molecular Chemistry by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.

Kinetic Reaction Analysis of an Anhydride-Cured Thermoplastic Epoxy:PGE/NMA/BDMA

Wei Chian* and Delmar C. Timm**

*Department of Chemistry and Chemical Engineering, South Dakota School of Mines, Rapid City, South Dakota 57701,

**Department of Chemical Engineering, University of NebraskaLincoln, Lincoln, Nebraska 68588

Macromolecules; (Article); 2004; 37; 8091-8097.

Received April 12, 2004; Revised Manuscript Received July 14, 2004

Published on web : 09/17/2004.

Copyright © 2004 American Chemical Society.

The American Chemical Society allows the posting of only the title, abstract, tables, and figures from articles appearing in the *Macromolecules*.

This article is published online at <http://pubs.acs.org/journals>

DOI: [10.1021/ma049293x](https://doi.org/10.1021/ma049293x)

ABSTRACT:

A comprehensive reaction analysis of a linear epoxy resin cured with an anhydride was performed to evaluate the reaction rate expressions. Monomers included phenyl glycidyl ether and methyl-5-norbornene-2,3-dicarboxylic anhydride or nadic methyl anhydride; the catalyst was *N,N*-dimethylbenzylamine; the initiator was *n*-propanol. Emphasis was initially placed on the molar dynamics of monomeric and oligomeric molecules. Molecular fractionations were achieved using reversed phase, high performance liquid chromatography. Chemical reaction rate constants were examined as a function of degree of polymerization. For the chain-initiated polymerization, the initiation rate constant was observed to be approximately 3 times greater than the propagation constant associated with oligomeric molecules. Both Poisson and Gold distributions were used to fit data. Examinations of polymeric fractions obtained by gel permeation chromatography in conjunction with a multiangle laser light scattering photometer revealed a minor side reaction that broadened the polydispersity index and resulted in the reduction of the cumulative, molar concentration of molecules as a function of conversion.

© 2004 American Chemical Society

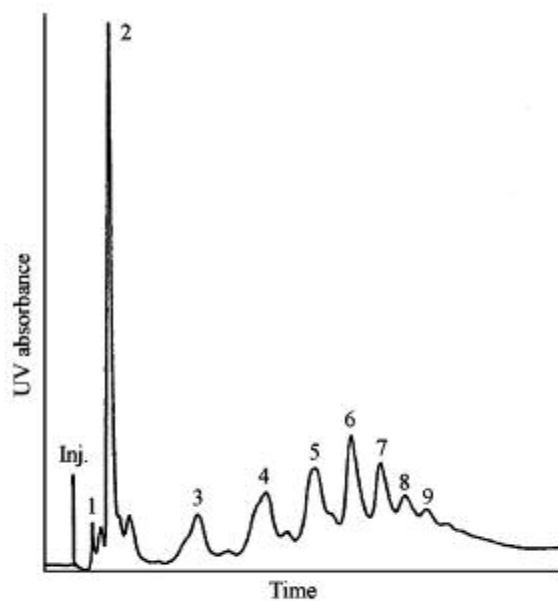


Figure 1. Representative HPLC chromatogram for the PGE/NMA/BDMA cure (initiator/anhydride ratio 0.15, 80 °C, 240 min): (1) NMA; (2) PGE; (3) P₁; (4) P₂; (5) P₃; (6) P₄; (7) P₅; (8) P₆; (9) P₇.

8094 Chian and Timm

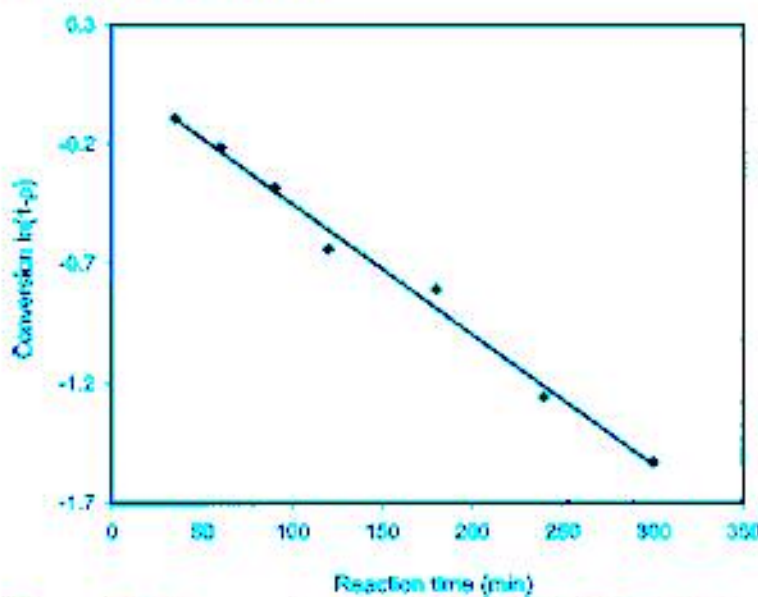


Figure 2. Monomer decay for PGE/NM (80 °C, initiator/anhydride ratio 0.15).

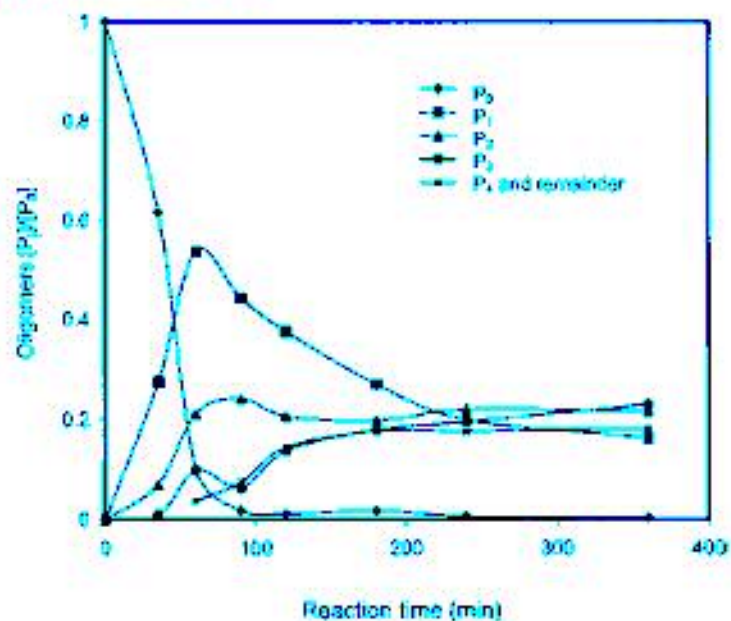


Figure 3. Oligomeric dynamics for PGE/NMA/BDMA (80 °C, initiator/anhydride ratio 0.15).

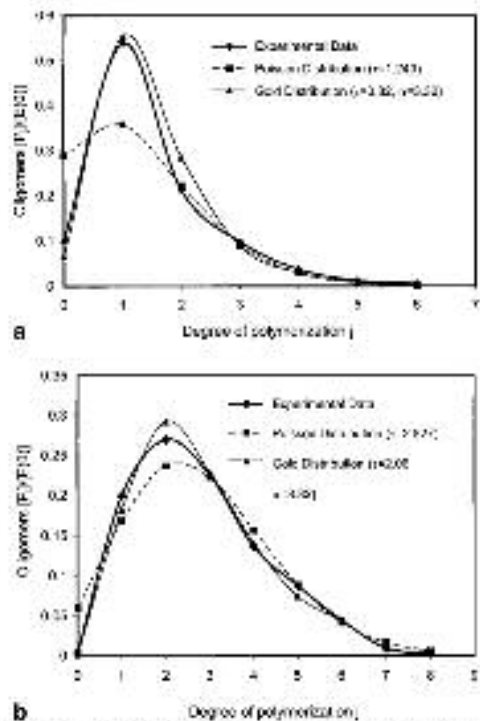


Figure 4. Theoretical experimental comparison of PBDs: (a) PGE/NMA/BDMA (60 min at 80 °C, initiator/anhydride ratio 0.01); (b) PGE/NMA/BDMA (190 min at 80 °C, initiator/anhydride ratio 0.15).

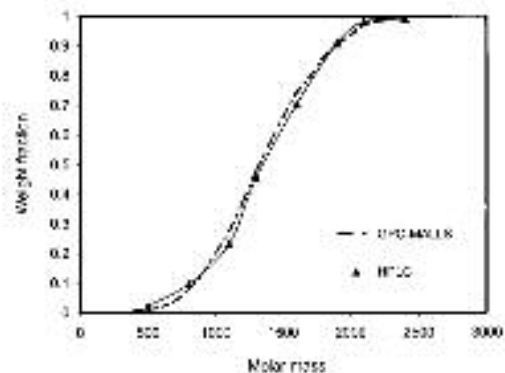


Figure 5. Comparison of cumulative weight fractions by SEC and GPC-MALLS (PGE/NMA/BDMA 240 min, 80 °C, initiator/anhydride ratio 0.15).

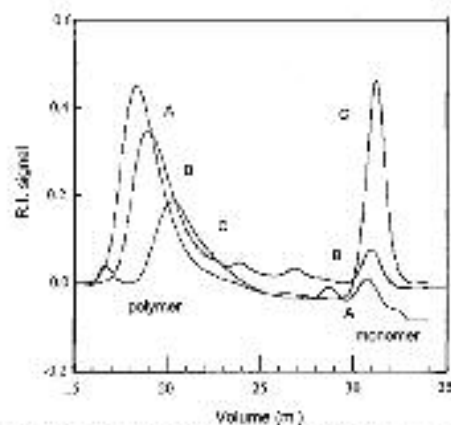


Figure 6. Chromatograms of PGE/NMA/BDMA reaction mixture (80 °C, initiator/anhydride ratio 0.01): (A) 240 min; (B) 190 min; (C) 60 min.

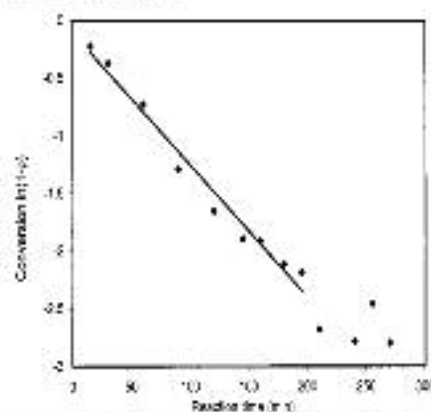
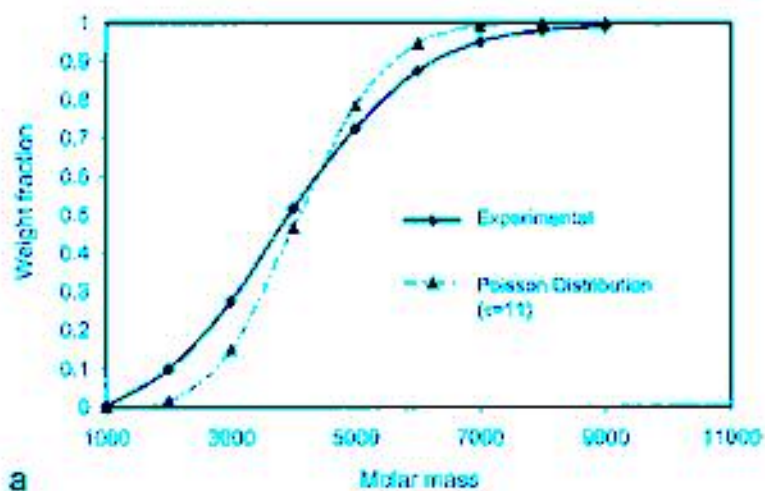
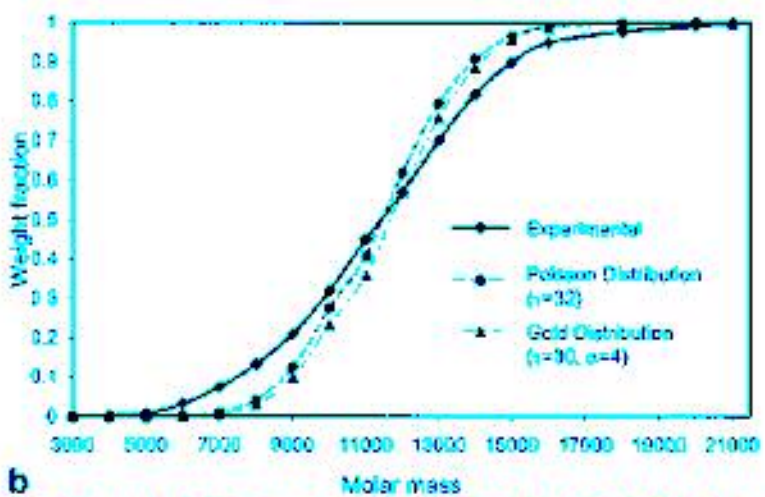


Figure 7. Monomer decay for PGE/NMA/BDMA at 80 °C (initiator/anhydride ratio 0.01).



a



b

Figure 8. Molecular weight distribution of (a) PGE/NMA/BDMA (60 min, 90 °C, initiator/anhydride ratio 0.01) and (b) PGE/NMA/BDMA.