

2009

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Mackowiak, C. L.; Grossl, P. R.; and Wheeler, R. M., "Crop Effects on Closed System Element Cycling for Human Life Support in Space" (2009). *NASA Publications*. 180.

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# Crop Effects on Closed System Element Cycling for Human Life Support in Space

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## ABSTRACT

Nutrient recycling in a space-based Bioregenerative Life Support System (BLSS) will require an understanding of nutrient dosage effects on crop production, plant tissue partitioning, and geochemical fates within crop systems. Sodium ( $\text{Na}^+$ ), fluoride ( $\text{F}^-$ ), and iodide ( $\text{I}^-$ ) are found in human waste streams. These elements were examined using crops in hydroponic systems. Lettuce, radish, spinach, and beet were used to study  $\text{Na}^+$  uptake and tolerance. Spinach, lettuce, and radish growth were inhibited at  $8.0 \times 10^{-2}$  M  $\text{Na}^+$  compared to the control. Beet growth improved at 2.0 and  $4.0 \times 10^{-2}$  M  $\text{Na}^+$  compared to the control. Rice plants were used to study  $\text{F}^-$  and  $\text{I}^-$  uptake and tolerance. Rice growth was inhibited at  $5.0 \times 10^{-4}$  M  $\text{F}^-$  and at  $5.0 \times 10^{-6}$  M  $\text{I}^-$ . Solution redox and sorption reactions were predicted with the aid of a chemical equilibrium model. A simulation model was used to predict element fates.

**Keywords:** beet, fluoride, geochemical model, iodide, lettuce, life support, radish, rice, sodium, spinach

## INTRODUCTION

Plants will play a key role in the life support functions of human habitats for extended space missions to planetary surfaces (Drysdale, 1995). This includes nutrient cycling between plants and humans (Barta et al., 1999; Wheeler and Strayer, 1997; Wheeler et al., 2001). Inorganic elements lost through plant

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Received 18 January 2007; accepted 17 May 2007.

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and human waste streams will be captured and recycled back into the human diet. There are several elements essential to humans but not to plants. The list of human essential elements is periodically reviewed and updated (Neilsen, 1998; Neilsen, 2000; Spears, 1999; Ulthus and Seaborn, 1996). Many of these elements or nutrients are not routinely tested for plant responses in cropping systems, unless a plant-derived element is an important contributor (e.g., calcium, iron) to the human diet.

Of all elements essential to humans but not plants, sodium (Na) is required in the largest quantity and thus, it will likely have the greatest short-term impact on human waste recycling in the plant system. Sodium is quite soluble (existing as  $\text{Na}^+$  in aqueous solutions) and will likely remain bioavailable once it enters the crop production system. The United States recommended daily allowance (RDA) is 3 g and the NASA recommendation for space is from 1.5 to 3.5 g (Lane et al., 1996). In general, Na is nonessential to higher plants, but it is essential to some  $C_4$  species, e.g., amaranth, saltbush (Brownwell, 1979) and is beneficial to several plant species, including a few on the NASA Bioregenerative Life Support System (BLSS) candidate crop list (Wheeler and Strayer, 1997). The BLSS crops benefiting from Na belong to the Chenopodiaceae family, e.g., spinach, chard, and beet. These crops not only tolerate Na but they can substitute Na for potassium (K), which results in higher Na accumulation in edible biomass (Subbarao et al., 1999).

The RDA for fluorine (F) has not been established but NASA recommends  $4 \text{ mg d}^{-1}$  F as  $\text{F}^-$  (Lane et al., 1996). There has not been definitive proof that F is an essential element but studies have shown that F nutrition mitigates dental caries (tooth decay) and may improve bone mass (National Academy of Sciences, 1989). Unlike Na, F might become less bioavailable in crop production systems over time. Fluorine, which typically exists as the fluoride ion ( $\text{F}^-$ ) in solution, readily reacts with calcium (Ca) and magnesium (Mg) to form solid-phases. In addition,  $\text{F}^-$  is incorporated into phosphorus (P) solid-phases to form fluoroapatite. Plants do not appear to profit from F nutrition and several reports have shown  $\text{F}^-$  supplied to plants above 1 mM may be toxic (Bar-Yosef and Rosenberg, 1988; Hara et al., 1977; Mackowiak et al., 2003; Stevens et al., 1997).

Iodine (I) is a proven essential nutrient where RDA and NASA concur with a daily requirement of  $0.15 \text{ mg d}^{-1}$ . This element exists in multiple valence states and thus is affected by the reduction/oxidation (redox) of its environment. In a highly oxidized environment, I exists as iodate ( $\text{IO}_3^-$ ) and under more reduced conditions iodide ( $\text{I}^-$ ) dominates. However, if  $\text{I}^-$  becomes oxidized, it may convert to volatile elemental iodine ( $\text{I}_2$ ) and escape into the atmosphere. In addition, I has been shown to combine with carbon in waste streams to form an array of potentially harmful halogenated organic compounds (Silverstein et al., 1994). Due to the ability of  $\text{I}_2$  to function as a bactericide, NASA has used it in their water treatment system for space. In this instance  $\text{I}_2$  is the active form (bactericide) that quickly reduces to  $\text{I}^-$  after contact, resulting in some

I<sup>-</sup> remaining in the water after treatment (Colombo et al., 1978). This potentially provides an additional source of I for plants, and humans (Gillman et al., 2001). Plant toxicity has been observed with I<sup>-</sup> concentrations as low as 10  $\mu$ M but plants can tolerate nearly 10-fold higher IO<sub>3</sub><sup>-</sup> concentrations (Mackowiak and Grossl, 1999).

The objectives of this research were 1) Compile plant yield and nutrient uptake data from NASA candidate crop dose-response studies with Na<sup>+</sup>, F<sup>-</sup>, and I<sup>-</sup>, 2) determine target element fates in a hypothetical BLSS hydroponic-based crop system using the dose-response data, geochemical equilibrium modeling and simulation modeling of target element mass transfers within a BLSS, and 3) discuss target element recycling scenarios and contaminant contingencies.

## MATERIALS AND METHODS

### Plant Culture

Representative crop species were 1) *Lactuca sativus* L. 'Waldmann's Green', *Raphanus sativus* L. 'Cherry Belle', *Beta vulgaris* L. 'Klein Bol', and *Spinacia oleracea* L. 'Nordic IV' for Na studies; 2) *Oryza sativa* L. '29-Lu-1' for I studies; and 3) 'USU-Super Dwarf' for F studies. Studies with Na were conducted in reach-in controlled environment chambers, whereas I and F studies were conducted in a greenhouse. The environmental parameters for all studies were within the envelope recommended for BLSS crop systems (Barta et al., 1999; Wheeler et al., 2001) and the environmental conditions for each nutrient study are summarized (Table 1). In the Na studies, water additions and nutrient solution pH were automatically controlled, whereas manual control (daily basis) was used for the other studies. The nutrient solution composition and maintenance methods used for the Na studies are the same used in a previous study with tomato and soybean (Mackowiak et al., 1999). Details of nutrient solution

Table 1  
Environmental conditions for plant studies

Element	Crop	Duration (days)	Irradiance (mol m <sup>-2</sup> d <sup>-1</sup> )	Temperature (°C)	RH* (%)	CO <sub>2</sub> ( $\mu$ mol mol <sup>-1</sup> )
Na	beet	35	13	23 $\pm$ 0.3	74 $\pm$ 9	1180 $\pm$ 166
	lettuce	21	13	23 $\pm$ 0.8	76 $\pm$ 10	1185 $\pm$ 137
	radish	21	13	23 $\pm$ 0.3	74 $\pm$ 9	1168 $\pm$ 158
	spinach	21	13	23 $\pm$ 0.4	73 $\pm$ 9	1195 $\pm$ 62
F	rice	63	29	28/22 $\pm$ 5	ambient	ambient
I	rice	113	16	29/22 $\pm$ 5	ambient	ambient

\*RH = relative humidity.

Table 2  
Plant culture conditions

Element	Crop	Salt	Hydroponic system*	pH <sup>†</sup>
Na	beet	NaCl	Recirculating nutrient film technique	5.8 ± 0.3
	spinach		”	5.6 ± 0.9
	radish		”	5.7 ± 0.2
	lettuce		”	5.7 ± 0.5
F	rice	KF	Aerated solution culture	5.0 ± 0.5
I	rice	KI and KIO <sub>3</sub>	Aerated solution culture	5.4 ± 0.4

\*The hydroponic systems varied as to their construction and nutrient solution recipes. Details for the Na, F, and I studies are given in Mackowiak et al. (1999), Mackowiak et al. (2003) and Mackowiak and Grossl (1999), respectively.

<sup>†</sup>The solution pH was automatically monitored and controlled in the Na studies and it was manually checked and adjusted each day in the F and I studies. All studies used either HNO<sub>3</sub> or KOH, as needed for pH control.

composition and maintenance for the F and I studies are described in detail elsewhere (Mackowiak et al., 2003; Mackowiak and Grossl, 1999). The hydroponic system descriptions and pH set points are given (Table 2). The iodine study included I<sup>-</sup> as part of the refill nutrient solution, where the concentration in the refill solution contained 20% of the initial I<sup>-</sup> treatment concentration, in addition to the plant essential elements. In contrast, the other studies only used initial doses of the target elements rather than continual additions via the hydroponic nutrient refill solutions.

### Harvest and Analyses

All plant biomass was oven-dried (65°C), weighed and ground (2 mm mesh) prior to tissue analysis. Roots were patted with paper towel to remove excess water and then dried. Roots were not rinsed in order to maintain any solid phases that might have developed during some of the studies. Sodium was determined by inductively coupled plasma spectrometry (ICP). Iodine (as I<sup>-</sup>) was determined by an alkaline dry-ashing procedure (Joepke et al., 1996; Mackowiak and Grossl, 1999) and digests analyzed colorimetrically, using a Lachat autoanalyzer (Lachat Instruments, Milwaukee, WI). Fluorine (as F<sup>-</sup>) was determined by an alkaline fusion ashing and selective ion electrode procedure (McQuaker and Gurney, 1977). Nutrient solution Na was determined by ICP. Nutrient solution I was determined colorimetrically, using the Lachat autoanalyzer, and nutrient solution F (as F<sup>-</sup>) determined with an ion selective electrode. Oven-dried root subsamples were taken from the F study, mounted on stubs, coated with carbon using an ion beam sputterer and examined with a

Hitachi S-4000 scanning electron microscope (SEM). Energy dispersive x-ray (EDX) digital dot mapping for Ca and F was performed with the Link EXL 1000 ECX system (Oxford Instruments, Oak Ridge, TN).

A completely randomized experimental design was used in the Na trials. Due to chamber size constraints, the study was run three times in order to obtain three treatment replications. Analysis of variance (ANOVA) proc GLM (SAS, Cary, NC) was used to analyze yield data and means comparisons performed using least significant differences (LSD). Tissue was combined among trials so only single values are available for tissue elemental content. The F and I trials were conducted as randomized block designs with three replicates (tubs) per treatment. The F treatments were analyzed by ANOVA (Prism, Graphpad Software, San Diego, CA) and means comparisons were performed using LSD. The I treatments were analyzed by ANOVA (Minitab 9.1, Minitab Inc., State College, PA) and planned comparisons were used to compare treatments with the control and among treatments.

### **Geochemical Modeling**

A chemical equilibrium model was used to individually model Na, F, and I in hydroponic nutrient solutions. Although there are several equilibrium models available, GEOCHEM-PC (Parker et al., 1995) was chosen because it adequately models hydroponic nutrient solutions. Assumptions and conditions for the model were, 1) the nutrient solution composition remained constant during the plant study, 2) solution pH remained constant, 3) redox conditions were not considered, and 4) mixed solids were not allowed to form. In addition, the model's  $\text{CaF}_2$  equilibrium constant ( $K^\circ$ ) was greater than several reported values, so it was replaced with a value from Elrashidi and Lindsey (1986).

### **Simulation Modeling**

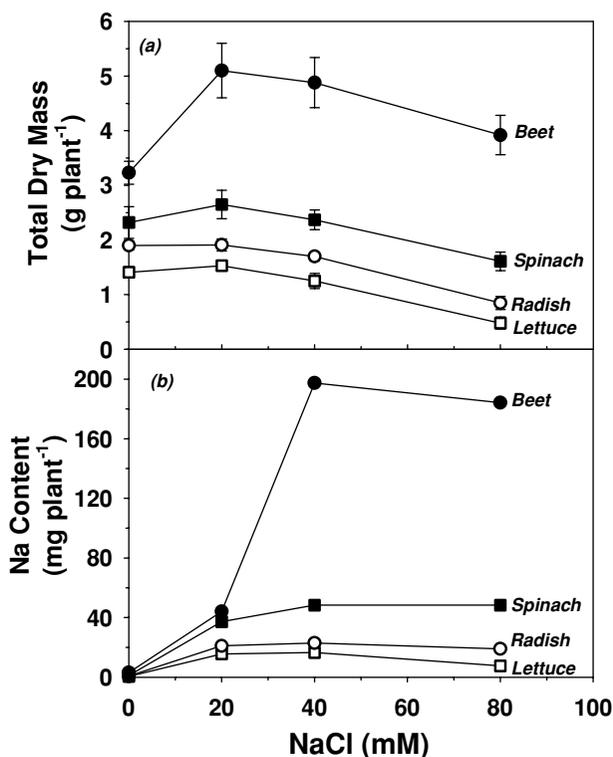
A simulation software program (Stella 6.0, High Performance Systems, Hanover, NH) was used to individually model Na, F, and I stocks (amount residing in system components) and fluxes in a hypothetical BLSS. Model parameters included a single person crew, 100% vegetarian diet using a hydroponic cropping system containing multiple-aged plants (steady-state) with a daily total biomass harvest of 1250 g dry mass, and external element resupply, as required. This equated to 50 m<sup>2</sup> crop growing area and a supporting nutrient solution volume of 500 L per person. The estimated crop production area requirements were somewhat large considering the high productivities that have been achieved with individual crop species (Bugbee and Salisbury, 1988; Edeen et al., 1993; Stutte et al., 1999) but it is a practical estimate assuming a mix of species and productivities (Wheeler et al., 1996). This simulation model

was based on a human (70 kg) ingesting recommended daily quantities of target elements, i.e., 3000 mg (130.4 mmoles) Na, 4 mg F (211  $\mu$ moles), and 0.15 mg (118  $\mu$ moles) I and excreting these amounts each day via waste products. The model was also based on the waste processing system returning 100% of the target elements to the crop system.

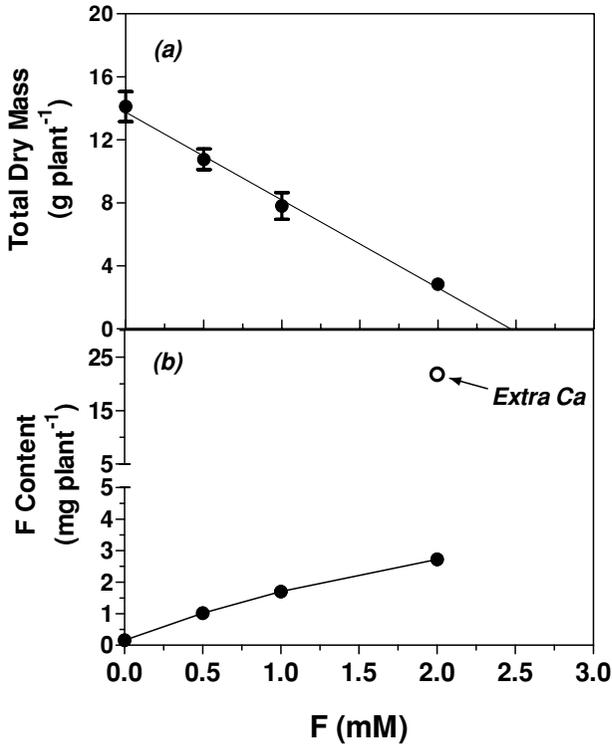
## RESULTS AND DISCUSSION

### Plant Dose-Response and Element Composition

Sodium response was dependent upon crop species. Compared to the control (no Na), beet biomass increased with 20 and 40 mM Na, spinach was unaffected, and lettuce and radish biomass declined at 80 mM Na (Figure 1a). These results coincide with salinity tolerance values for herbaceous crops (Richards, 1954).



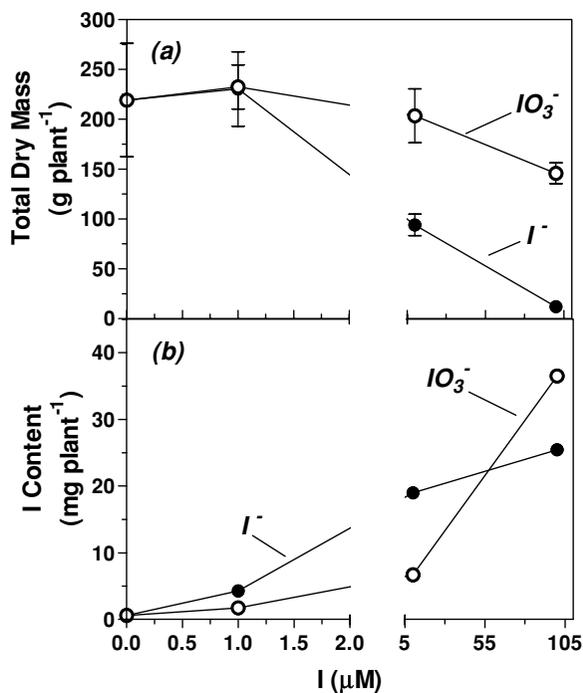
**Figure 1.** Sodium effect on total biomass yield (a) and Na content in the total plant (b) at 21 days for lettuce, radish, and spinach, and 28 days for beet. Vertical bars = standard errors (a).



**Figure 2.** Fluoride effect on total biomass yield (a) and F content in the total plant (b) at 63 days for rice 'Super dwarf'. Vertical bars = standard error (a).

Sodium removal by beet was greatest and least for radish and lettuce (Figure 1b). Based on the growth response and Na accumulation data, it appeared that approximately 40 mM was the upper solution concentration limit for crops in this system.

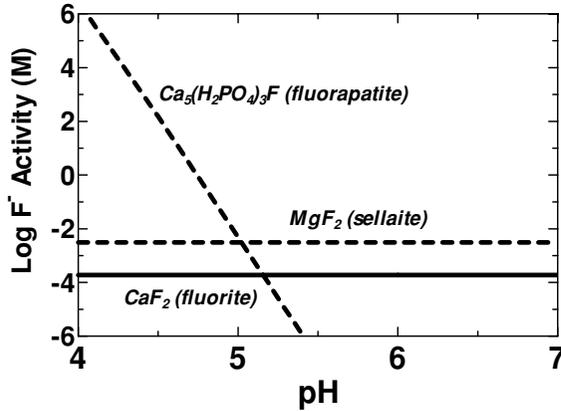
Rice growth was affected by all levels of F, where biomass decreased linearly with increasing solution F<sup>-</sup> (Figure 2a). This study did not continue through seed maturity but results from a previous study showed that rice seed F was less than 0.5% of shoot tissue concentrations (Mackowiak et al., 2003). Plants from this study had begun seed fill at the time it was terminated (63 days). Harvested panicles (immature seed + chaff) contained less than 19 mg kg<sup>-1</sup> F (5% of leaf values) for the 2 mM F treatment and most of the F was likely associated with chaff tissue. This suggests that F is not readily transported from xylem to phloem and so little would be expected in the grain. The total F content increased per plant as solution F increased (Figure 2b). It is interesting to note that when additional Ca was added to the nutrient solution (from 1 mM to 2 mM Ca), total plant F increased from 2.7 to 22 mg plant<sup>-1</sup> (Figure 2b). Over 90% of



**Figure 3.** Iodine effect on total biomass yield (a) and I content in the total plant (b) at 113 days for rice '29-Lu-1'. Vertical bars = standard error (a).

the plant F in the 2x Ca treatment was associated with the root, mostly coating the surface as a CaF<sub>2</sub> solid phase, as determined by EDX spectroscopy. The solid phases are addressed further in the geochemical interactions discussion.

Iodine valency affected I toxicity in rice. The more reduced form, I<sup>-</sup>, was more inhibitory than the oxidized form, IO<sub>3</sub><sup>-</sup> (Figure 3a). Although there tended to be less growth with increasing solution IO<sub>3</sub><sup>-</sup>, the differences were not significant. As with the other test elements, increasing solution I concentrations resulted in a greater accumulation of I in the plant (Figure 3b). At 100 μM I, there was actually more total I removed from solution by the IO<sub>3</sub><sup>-</sup> treatment than with the I<sup>-</sup> treatment since plant growth was greatly inhibited with the I<sup>-</sup> treatment. Although it was not addressed in this trial, I<sub>2</sub>, which has been used to treat International Space Station water (Parker et al., 1999) is even more inhibitory to plants than I<sup>-</sup>. In a recent study, I supplied to rice at 20 μM with an I<sub>2</sub>:I<sup>-</sup> ratio of 0.6:1.0 resulted in a 70% reduction in growth compared to a 35% reduction in growth when I was provided solely as I<sup>-</sup> (Mackowiak et al., 2005). The I<sub>2</sub> form is unstable in non-acidic solutions where it can be readily (within hours) reduced to I<sup>-</sup> and volatilized to some extent.

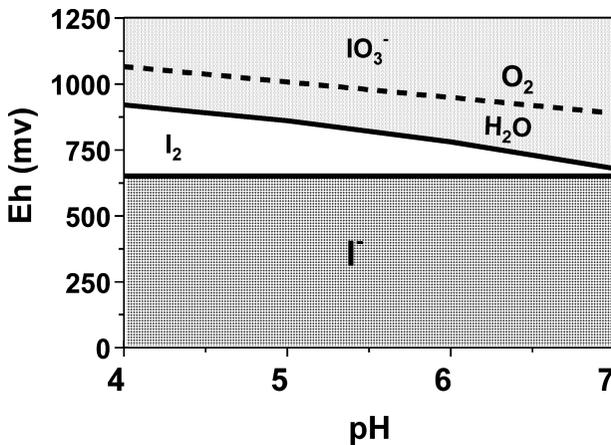


**Figure 4.** Fluorine stability diagram where  $F^-$  activity is a function of solution pH. Fluorite controls  $F^-$  activity below pH 5.2 and fluoroapatite controls  $F^-$  activity from pH 5.2 to 7.0. Adapted from Erashidi and Lindsey (1986).

### Geochemical Interactions

Sodium remains soluble at high NaCl concentrations (up to 6 M). Sodium readily forms complexes with  $Cl$ ,  $CO_{2(g)}$ , and  $SO_4^{2-}$ , which have little effect on overall Na bioavailability. Sodium may form solids with other elements, particularly  $CO_3^{2-}$ , Al, and Si but it is unlikely to occur in these hydroponic cropping systems.

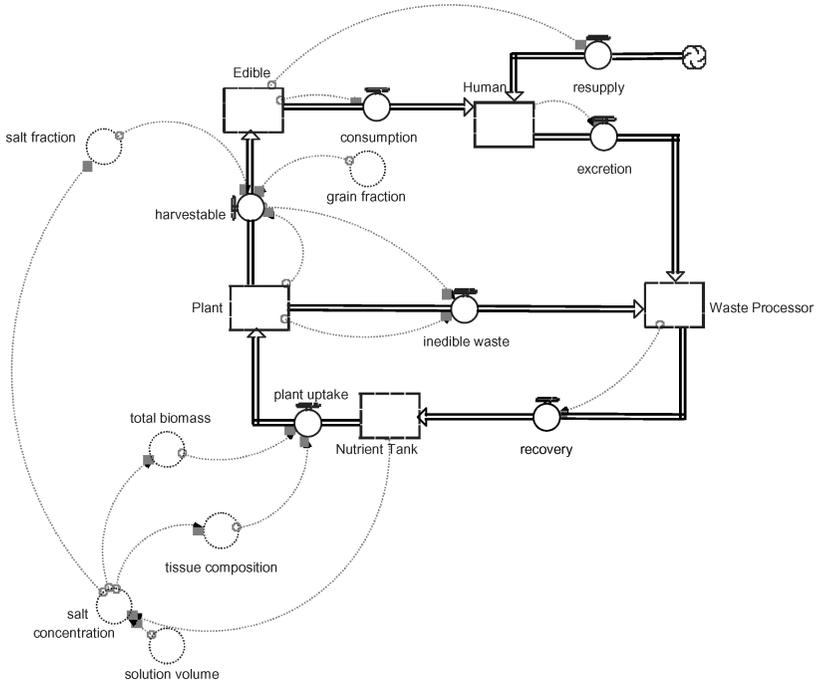
Fluoride readily reacts with several cationic species, such as  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $Fe^{3+}$ , and  $Na^+$ , as well as  $PO_4^{3-}$ , to form mixed solid phases. The fluoride interactions of greatest concern in a BLSS are with Ca and Mg (Figure 4). The model predicts that under acidic conditions (pH less than approximately 5.2),  $CaF_2$  controls  $F^-$  activity in solution, to approximately 0.15 mM (2.9 ppm). If more  $F^-$  is added to solution, it would likely form a  $CaF_2$  precipitate, thereby retaining solution equilibrium near 0.15 mM in the hydroponic solution. At higher pH values, fluoroapatite [ $Ca_5(PO_4)_3F$ ] controls F activity and therefore soluble  $F^-$  concentrations would likely decline at a correspondingly higher solution pH (Figure 4). Based on the geochemical model used in this study and a nutrient solution pH of approximately 5.0, initial solution  $F^-$  at 2 mM was predicted to decrease to 0.53 mM at equilibrium because of the formation of Ca-F solid phases. Additional Ca (from 1 mM to 2 mM) would reduce solution  $F^-$  further to 0.26 mM. Doubling nutrient solution Ca greatly increased the amount of F associated with root tissue (Figure 2b). Root ICP analysis showed total root Ca had increased from 0.63% to 5.12% in the 1x Ca and 2x Ca treatments, respectively. Root surface EDX spectroscopy identified F and Ca as the major elements found on the root, suggesting that a Ca-F solid phase



**Figure 5.** Fields of stability for I species as a function of pH in a natural aquatic system. The dashed line represents the redox equilibrium of H<sub>2</sub>O with O<sub>2</sub>, where decreasing the electron activity (an increased Eh) would result in O<sub>2</sub> evolution. The Eh of a well-aerated hydroponic solution ranges approximately 400 to 700 mv. Adapted from Fuge and Johnson (1986).

coated the root surface rather than these elements being sequestered within the root.

Iodine will not form precipitates in a BLSS system unless heavy metal contaminants, i.e., Ag, or Hg, are present (Lindsay, 1979), resulting in solids and gasses, respectively. Iodine exists in multiple oxidation states, which greatly determine the degree of toxicity to plants and microbes. The most common oxidized I species is IO<sub>3</sub><sup>-</sup> and the most common reduced form is I<sup>-</sup> (Figure 5). Typical equilibrium redox values are around 400 to 700 mv (pe = 6.8 to 11.8) for a well-aerated hydroponic system. At equilibrium, IO<sub>3</sub><sup>-</sup> should not exist at oxidation states below 700 mv (Figure 5). However, we found that the majority of added IO<sub>3</sub><sup>-</sup> remained as IO<sub>3</sub><sup>-</sup> for at least 100 days in previous plant studies (Mackowiak and Grossl, 1999), suggesting that the kinetics of IO<sub>3</sub><sup>-</sup> to lower oxidation states are slow. The I<sup>-</sup> species is likely more bioavailable to plants than IO<sub>3</sub><sup>-</sup> (Böszörményi and Cseh, 1960; Mackowiak et al., 2005; Umaly and Poel, 1971). If reduction of IO<sub>3</sub><sup>-</sup> to I<sup>-</sup> was quick, plant growth inhibition would likely have been greater with the corresponding IO<sub>3</sub><sup>-</sup> treatments (Figure 3a). The relative toxicity of I to plants is I<sub>2</sub> > I<sup>-</sup> > IO<sub>3</sub><sup>-</sup>. In a BLSS, I<sup>-</sup> may become available as a by-product of water treatment, such as the microbial check valve (Colombo et al., 1978) and I<sup>-</sup> is the main form found in human urine (Michalke and Scramel, 1999). If I resins are used to treat potable water, a post treatment will be needed to minimize residual water I<sup>-</sup> and possible I volatilization. Some techniques, such as carbon water filters, are under consideration for treating residual I<sup>-</sup> in water (Parker et al, 1999).



**Figure 6.** A schematic representation of element cycling in a biological life support system. Some adjustments in the given model were added (not shown) to test further resupply/contamination scenarios (i.e., for  $\text{Na}^+$  and  $\text{I}^-$ ) or F loss by  $\text{CaF}_2$  solid phase formation.

### Simulation Models

With the aforementioned data, simulation models were created to characterize these elements and their mass transfers in a BLSS. The basic simulation scenario is given (Figure 6). The model has three main components, stocks ( $\square$ ), flows (valves), and converters ( $\circ$ ). The stocks are reservoirs that accumulate the target element, thus they can act as both, sources and sinks. The flows represent pipes that deliver materials to and from stocks. The converters contain equations or coefficients used to help define and regulate the flows. The dashed lines with arrows (connectors) represent dependencies (relationships) between components. For example, excretion (a flow) is dependent on the Human stock; resupply (a flow) is dependent on how much target element exists in the Edible stock; plant uptake (a flow) is dependent on total biomass (a converter) + tissue composition (a converter) equations (Figure 6). The resupply flow represents importation of the target element from outside the system into the human diet. This would occur if sub-adequate amounts were being provided to the Human

stock via the Edible stock. The grain fraction converter (Figure 6) is the proportion of food supplied by grain crops. Based on data from our hydroponic studies, we assumed that grains would not accumulate Na, F, or I. Therefore, if grain fraction = 1, then no elements will enter the Human stock from plants (100% resupply required). Based on this simple design, several scenarios were simulated for each element.

### Sodium

The model daily human requirement was 130.4 mmoles (3 g) Na per person per day. If no Na came from crops (grain fraction = 1) and all Na was from resupply, then plant productivity (total biomass) declined 10% within 230 days (7.5 months) and the corresponding nutrient solution Na concentration reached 50 mM. If we included non-grain crop plants to supply some of the human Na requirement then the system eventually reached equilibrium. For example, 20% non-grain crops resulted in an end of resupply by 1.7 years, at an equilibrium concentration of 12.4 mM Na in the nutrient solution. If the non-grain component increased to 40% (typical of a BLSS diet), the resupply ended in 11 months, with a similar Na nutrient solution concentration (12.4 mM Na). However, if we assumed that the crew would continue to use Na to flavor meals, such as 10% of their daily requirement (13 mmoles daily resupply) and we had 40% non-grain crops, then the input to the human stock became rather large. Sodium consumption passed the recommended 130.4 mmoles daily intake by 83 days and would increase another 50% by 616 days (1.7 years). In this scenario, plant productivity would remain within 90% of set point for well over 6 years. It is interesting to note that non Na accumulating crops (Chenopods) could supply the total human daily Na requirement (based on 50 m<sup>2</sup> per person growing area). Including additional Na to the system via condiments and flavor enhancements would likely affect the crew's intake level sooner than it would affect plant productivity.

### Fluoride

Since F will exist primarily as F<sup>-</sup> in the BLSS, the models simulated only F<sup>-</sup> effects. In a worst case scenario, [i.e., no F precipitate formation, no F from crops (grain fraction = 1)], and a human intake of 211 μmoles per day), total plant biomass dropped 10% within 1.6 years, at a corresponding nutrient solution concentration of 0.24 mM F<sup>-</sup>. With 20% non-grain crops, total biomass dropped approximately 9% by the eighth year and resupply was no longer required. With 40% non-grain crops, resupply ended in 6.8 years. However, CaF<sub>2</sub> precipitates form when solution F<sup>-</sup> values approach 0.3 mM. Even with no crop recycling (100% resupply), nearly all the human excreted

F would likely become bound in a solid phase with nutrient solution Ca. The model results suggest a nutrient solution equilibrium F concentration of approximately  $1 \mu\text{M}$ , which has little if any effect on plant productivity. Based on these simulations, it appears that F may not be a concern in a BLSS as long as the solid phases do not dissolve (i.e., alkaline solutions).

### Iodine

The prominent I species in the human diet is as  $\text{I}^-$  (118  $\mu\text{moles}$  or 15 mg). In addition, the organic I species in human metabolic waste is transformed to  $\text{I}^-$  by the time it leaves the body via urine (Michalke and Schramel, 1999). If no crops provided I (grain fraction = 1), then plant productivity would drop 10% within seven years at a corresponding nutrient solution concentration of  $3.4 \mu\text{M I}^-$ . If 20% of the crop mix were non-grain crops, then plants could fulfill the entire human requirement by 294 days (9.5 months) or 162 days (5 months) with 40% of the food provided as non-grain crops.

A potential problem exists if water iodination is used as a bactericidal treatment. Research by Wiederhoeft et al. (1999) reported that they could reduce residual  $\text{I}^-$  in potable water from approximately 2 to 0.28 ppm ( $2.2 \mu\text{M}$ ). With a daily water allotment of approximately 25 L per person (Parker et al., 1999), the plant system would receive an additional 55  $\mu\text{moles}$  of  $\text{I}^-$  each day. Based on the simulation (40% non-grain crops), human I ingestion would double within 5 days and total plant biomass would drop 10% within 61 days (2 months). Based on the simulation results, the acceptable limits for residual I should be lowered. If residual  $\text{I}^-$  was decreased 90% then the system could run for 663 days (1.8 years) before reaching a 10% decline in plant biomass (nutrient solution  $\text{I} = 3.6 \mu\text{M}$ ) at a mere 3% increase in daily human I intake.

## CONCLUSIONS

The process of characterizing element cycling through a BLSS requires inputs at several levels. Based on reasonable inputs, simulations can be constructed to help guide future systems studies and the eventual development of a BLSS or other closed life support systems. This work used plant and geochemical data for constructing simple simulations of human essential element cycling in a BLSS.

The daily addition of F to a BLSS might not result in any net nutrient solution F accumulation because of the formation of  $\text{CaF}_2$  solids at Ca concentrations adequate for plant growth. Iodine cycling reached an acceptable steady state in the simulation model, unless residual I was also added to the system via bactericidal water treatment. Low residual I levels that have been used by NASA in the past may result in excessive I problems in both plant

and human systems within months of operation. A mixed cropping scenario would supply the daily human Na requirements. The addition of even 10% more Na to improve food palatability might result in excessive Na uptake by the crew prior to any deleterious effect is seen in crops. In fact, Na may be the biggest challenge in element cycling in a BLSS. Sodium is heavily relied upon as a flavor enhancer and it is not easily separated from other elements in the nutrient-rich waste stream which feeds the cropping system.

### ACKNOWLEDGMENTS

We thank Lisa Ruffe for her lab assistance with the Na studies at the Kennedy Space Center (KSC), Justin and Melissa Crane for their lab assistance with the F and I studies, and the analytic chemistry staff at KSC and Utah State University. Financial support was provided by a NASA/GSRP fellowship.

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