

University of Nebraska - Lincoln

DigitalCommons@University of Nebraska - Lincoln

---

UCARE Research Products

UCARE: Undergraduate Creative Activities &  
Research Experiences

---

4-28-2020

## Nanomechanical Characterization of Fuel Cell Ionomers

Jackson Goddard

University of Nebraska - Lincoln, jacksongoddard21@gmail.com

Follow this and additional works at: <https://digitalcommons.unl.edu/ucareresearch>

 Part of the [Polymer Science Commons](#)

---

Goddard, Jackson, "Nanomechanical Characterization of Fuel Cell Ionomers" (2020). *UCARE Research Products*. 222.

<https://digitalcommons.unl.edu/ucareresearch/222>

This Poster is brought to you for free and open access by the UCARE: Undergraduate Creative Activities & Research Experiences at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in UCARE Research Products by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.



# Nanomechanical Characterization of Fuel Cell Ionomers

Author: Jackson Goddard<sup>1,2</sup> Advisor: Shudipto Dishari<sup>1</sup>

<sup>1</sup>Department of Chemical and Biomolecular Engineering

<sup>2</sup>Department of Mathematics



UCARE  
COLLEGE OF  
ENGINEERING

## Introduction

### Renewable Energy

- Energy sustainability can be achieved by improving and adopting clean energy technologies
- Proton exchange membrane fuel cell (PEMFC) is a promising energy conversion device
- Better nanoscale understanding of ionomer-catalyst layers could lead to increased efficiency of PEMFCs<sup>1</sup>

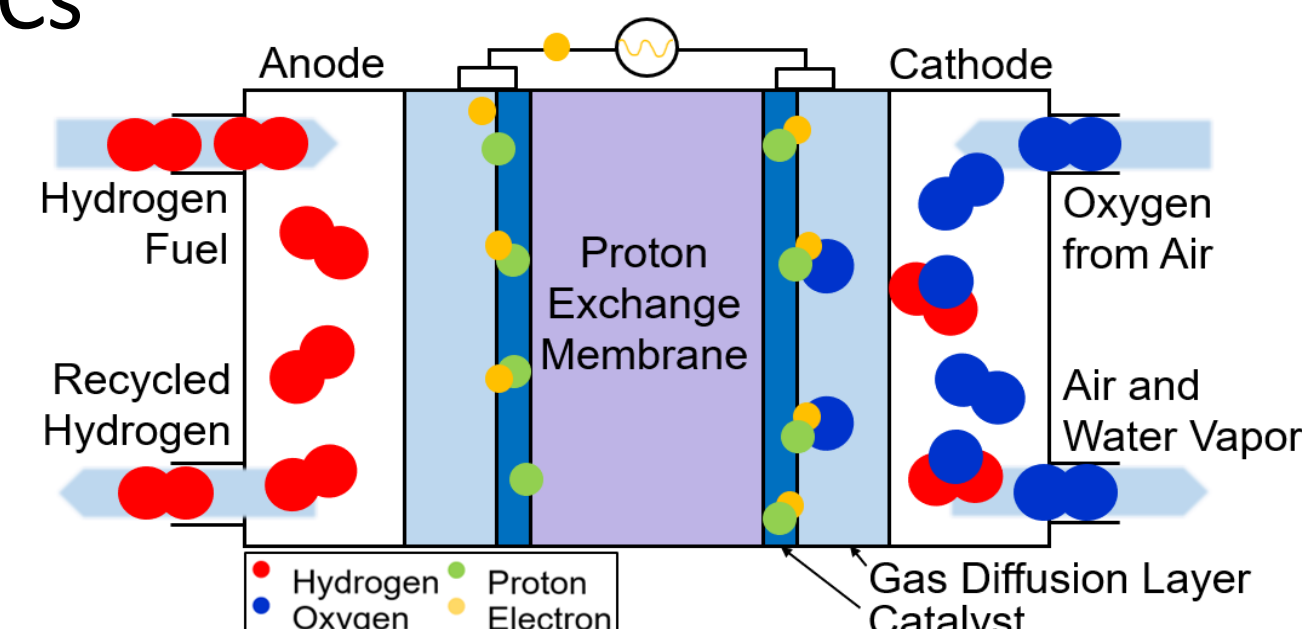


Fig 1. PEMFC schematic.

## Materials

**Nafion:** Perfluorosulfonic acid ionomer

**PFIA:** Perfluoroimide acid ionomer

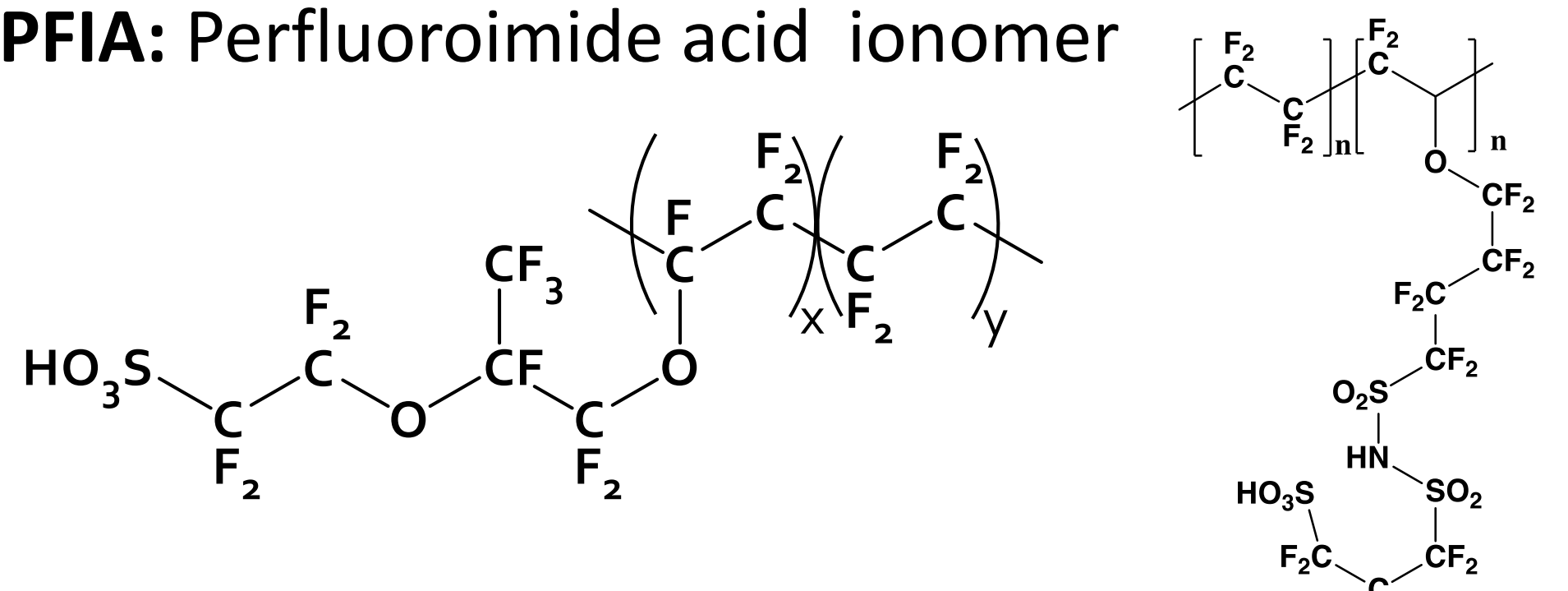


Fig 2. Left: Nafion. Right: PFIA

## Methods

### Sample Preparation

- Ionomer solutions (1, 2, 5, 7, 10 wt %) sonicated for 20 min; spin coated on Si wafer pieces; dried at 42°C for 3 hours then at 83°C for 7 hours; cooled in vacuum for 12 hours

### Contact Resonance Force Microscopy

- Oscillating current of increasing frequency in cantilever

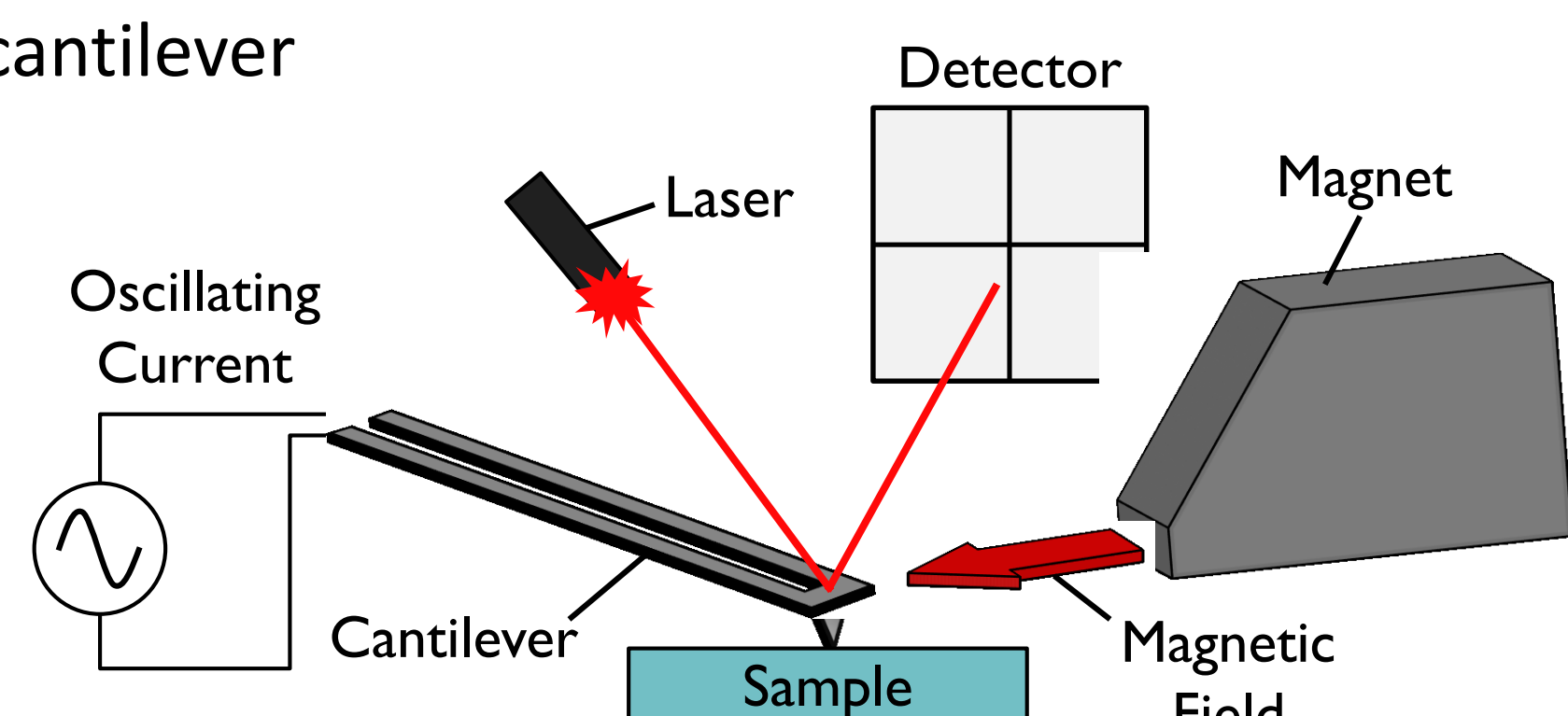


Fig 3. CRFM Schematic

## Results

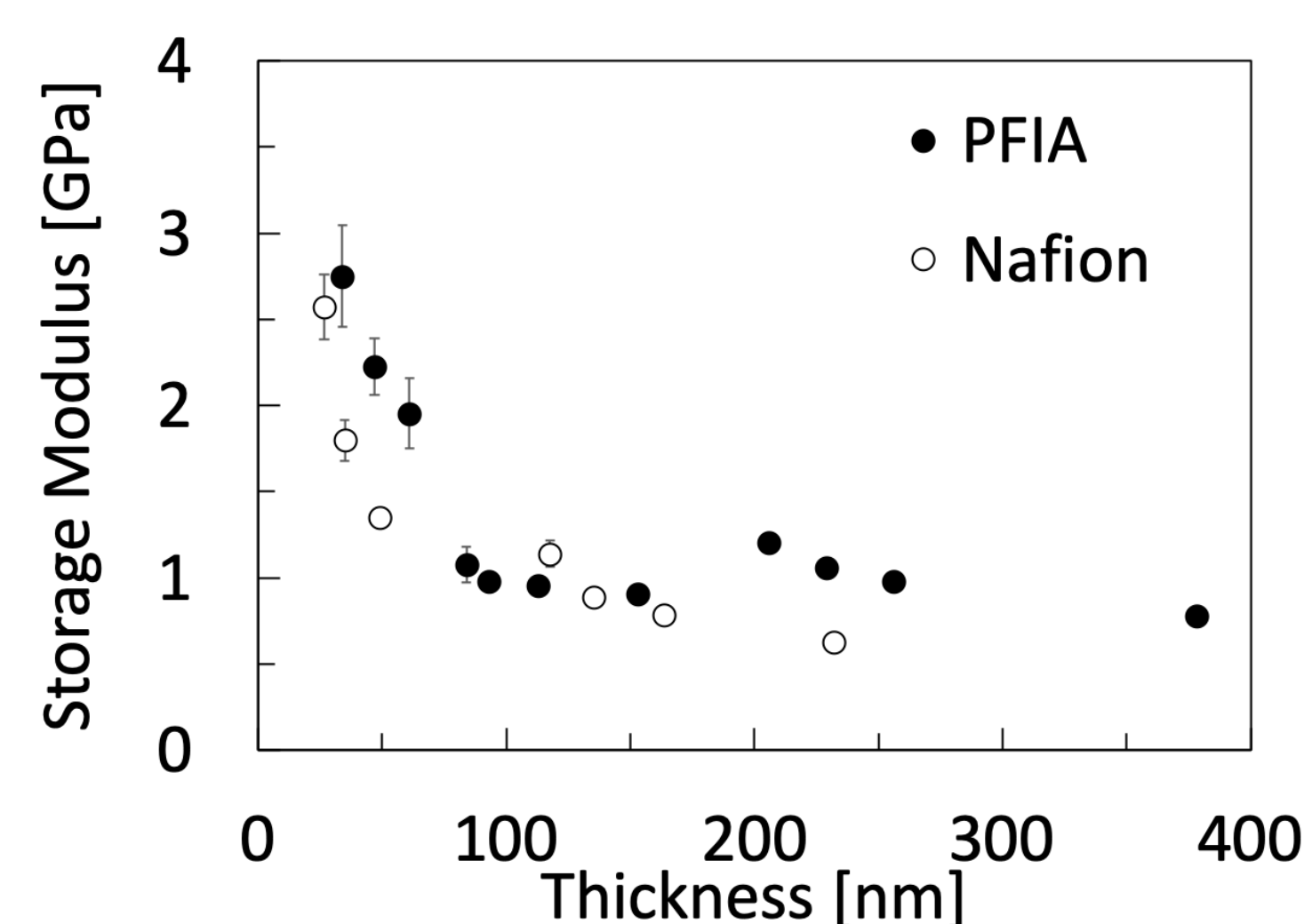


Fig 4. Stiffness of Nafion and PFIA

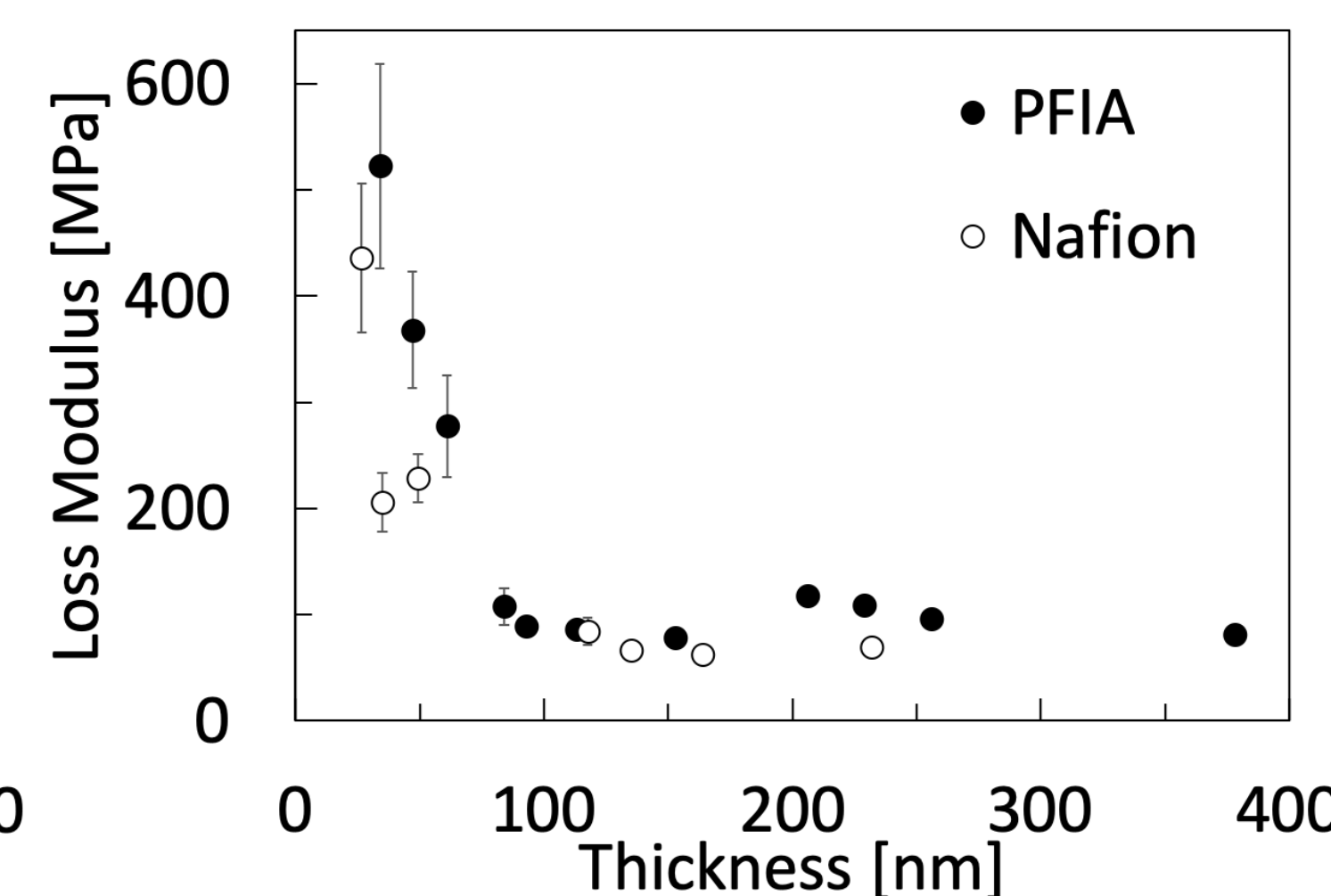


Fig 5. Loss modulus of Nafion and PFIA

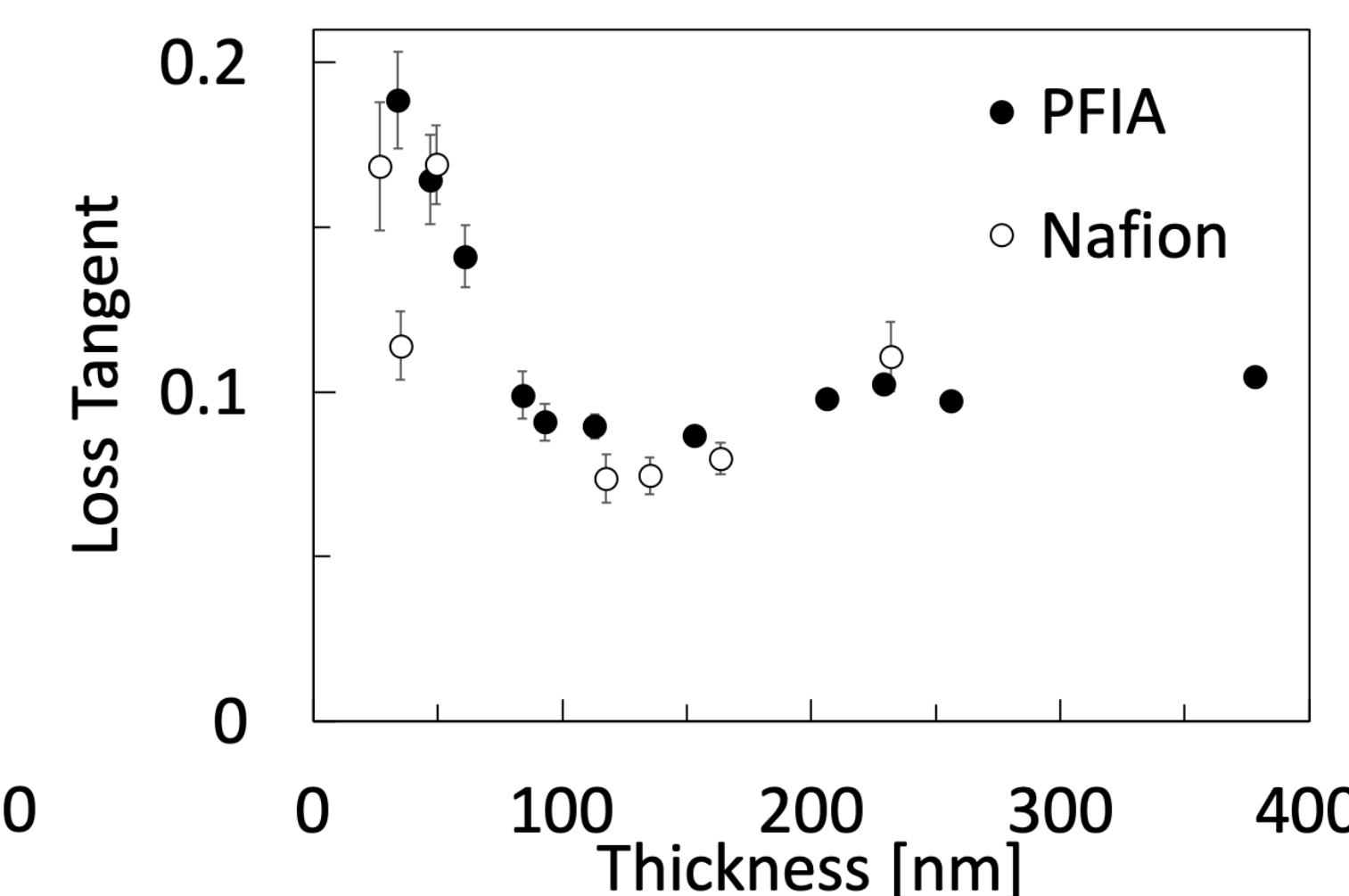


Fig 6. Ratio of viscoelastic properties of Nafion and PFIA

### Comparison of Nafion and PFIA

- As the film thickness decreases, both storage and loss modulus increase.
- The small difference in viscoelastic modulus between these two ionomers suggests the insignificant impact of bisulfonyl imide group on viscoelastic modulus.

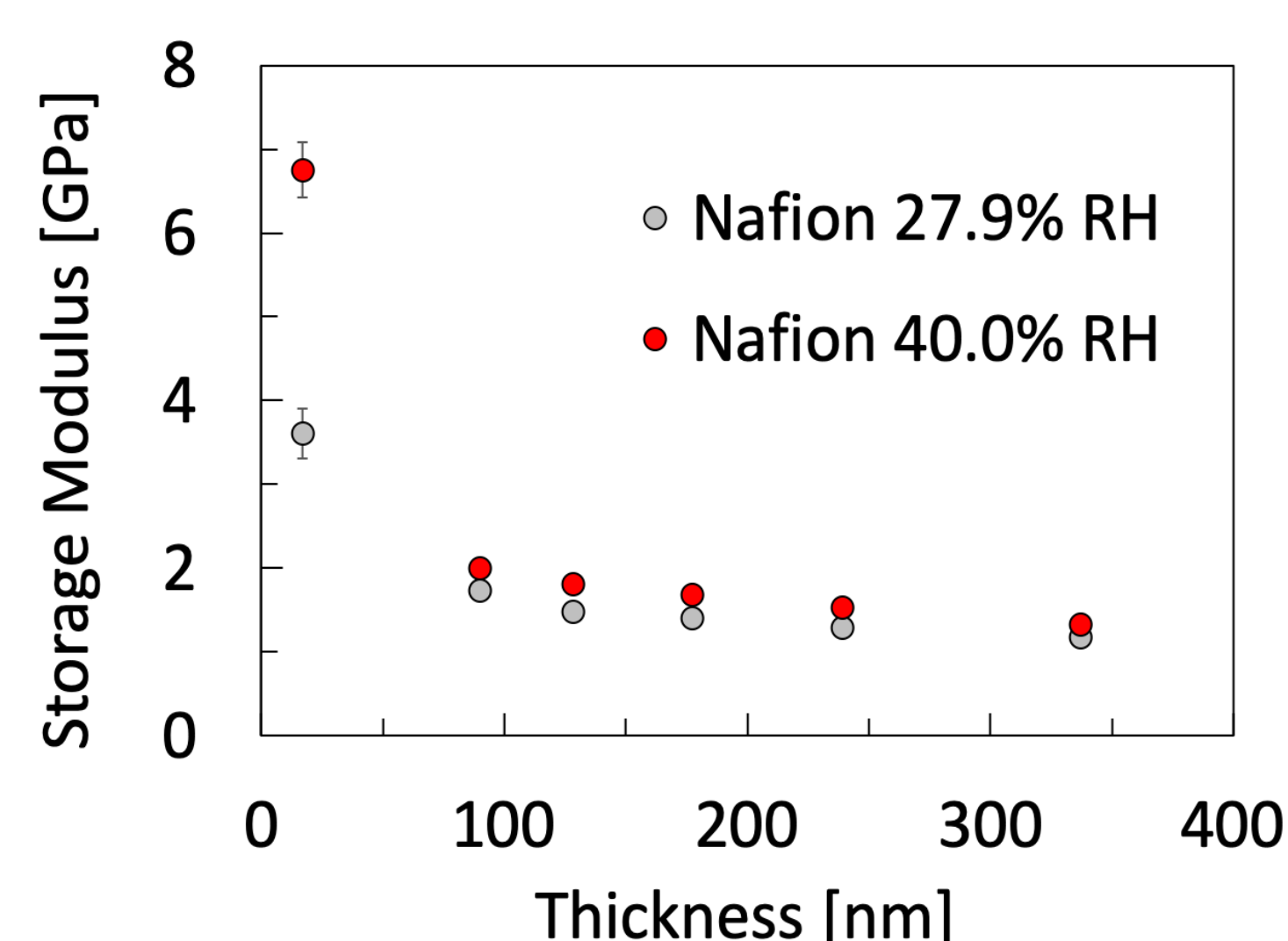


Fig 7. Effect of relative humidity on ionomer stiffness

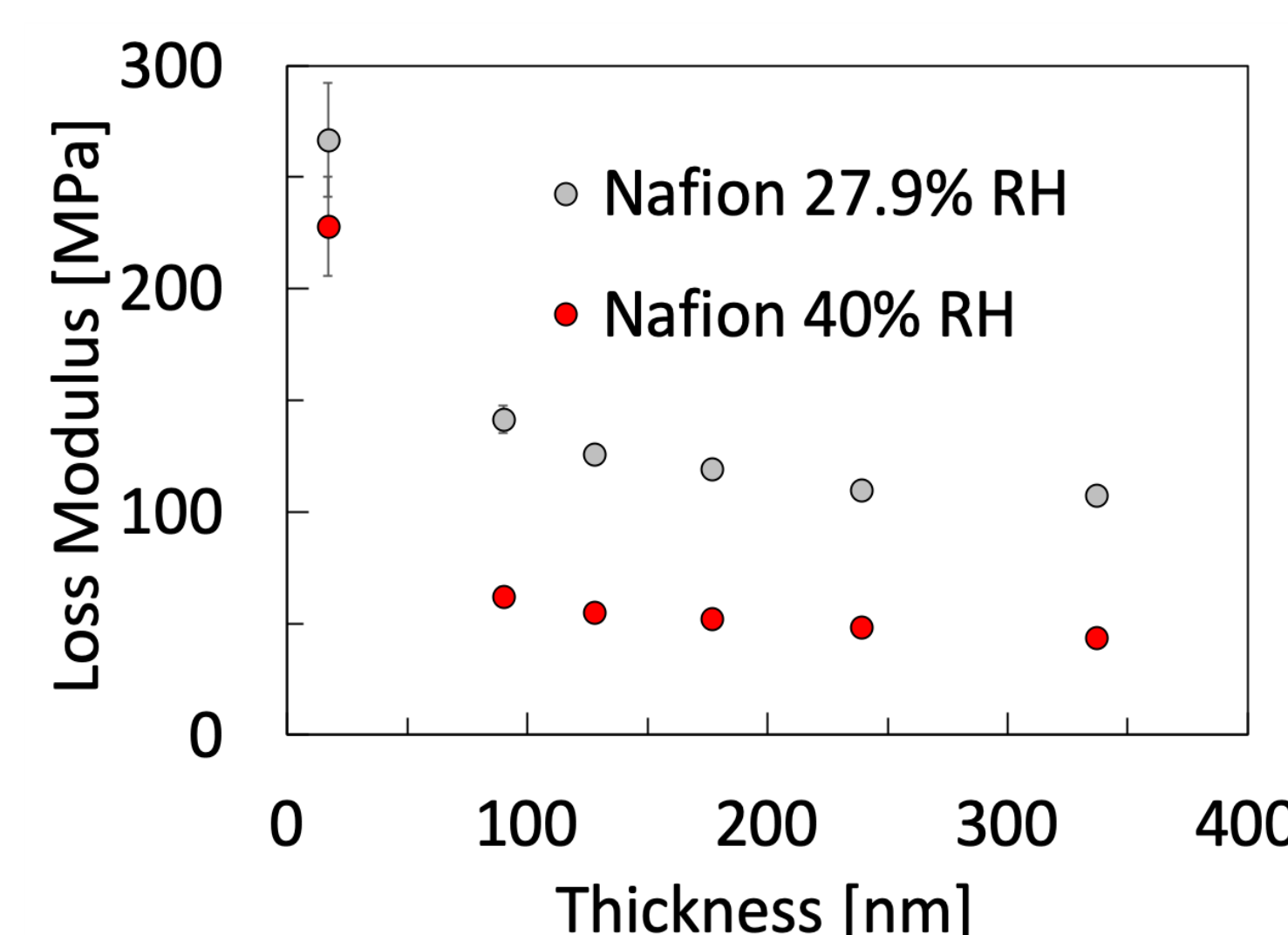


Fig 8. Effect of relative humidity on ionomer viscosity

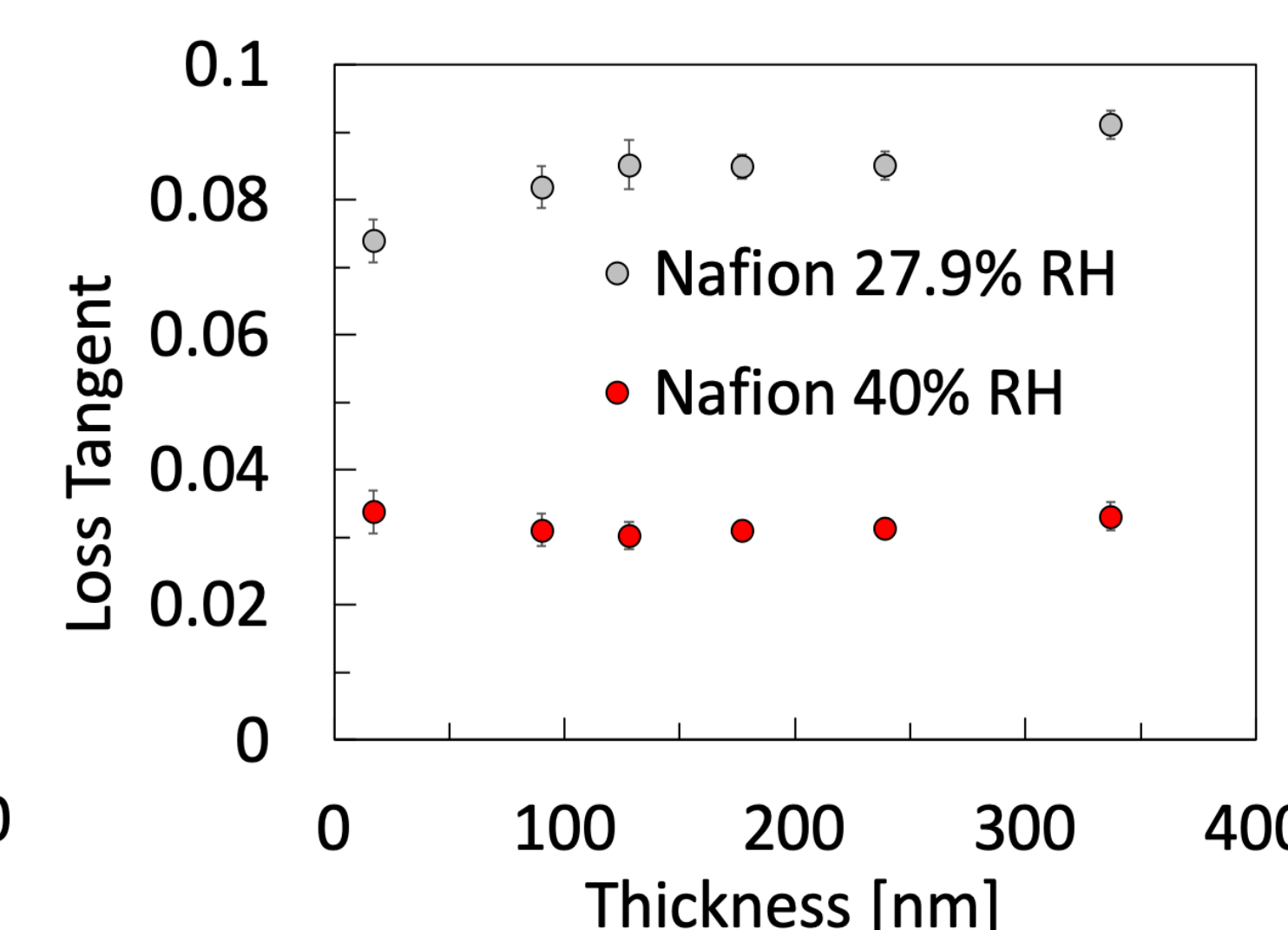


Fig 9. Ratio of viscoelastic properties under different relative humidities

### Relative Humidity and Viscoelastic Properties

- Increasing** the relative humidity **increases** stiffness when the film thickness is < 100 nm.
- At very low thickness, water and substrate both strongly **hydrogen-bond** with ionomer side chains and lead to a **drastic increase** in film stiffness.
- Decrease** in loss modulus when % RH was increased from 28% to 40% is an interesting observation. This suggests **lower energy dissipation** with hydration.

### Example of CRFM Data

- Resonance frequency shift** related to storage modulus
- Width (w)** is related to loss modulus
- Free (no contact) case for **cantilever** properties
- Nafion **bulk** (60  $\mu$ m) for **film** calibration

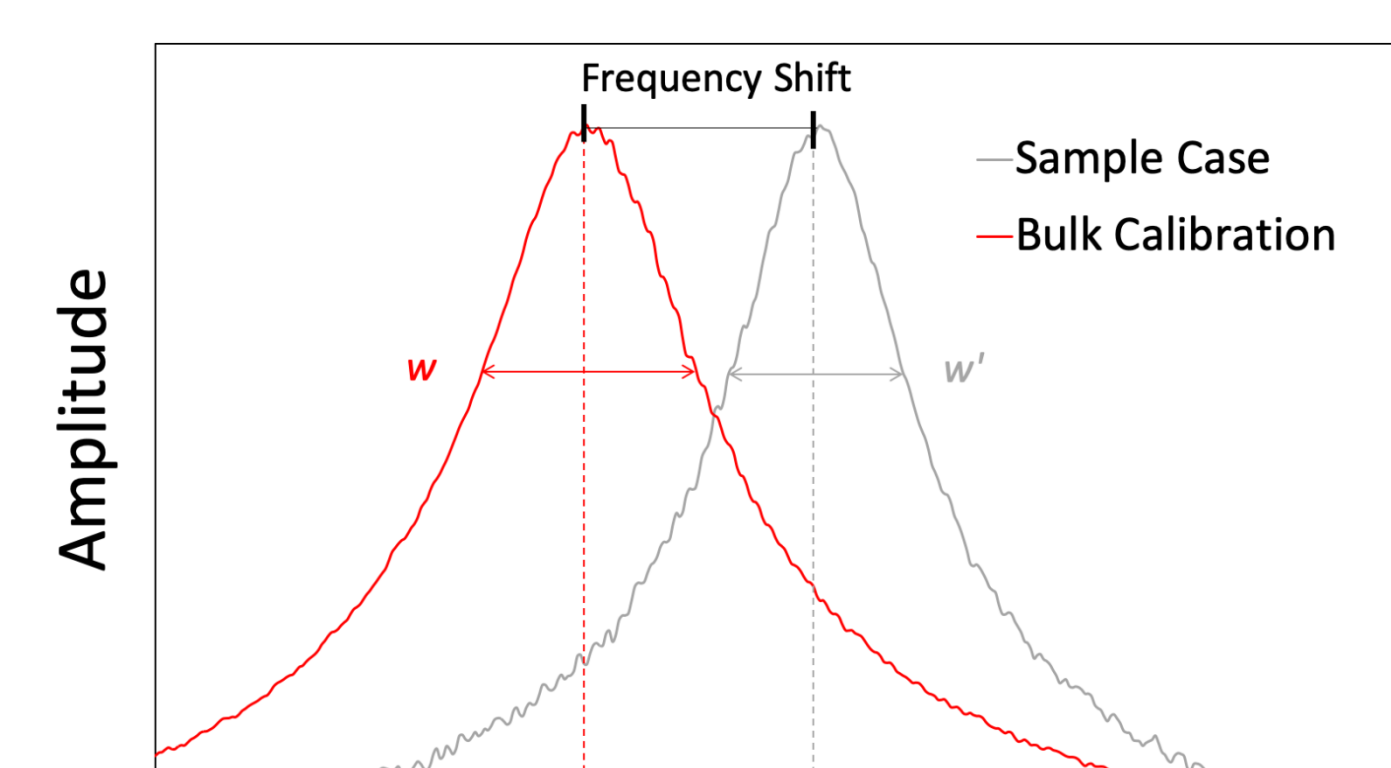


Fig 10. Example data

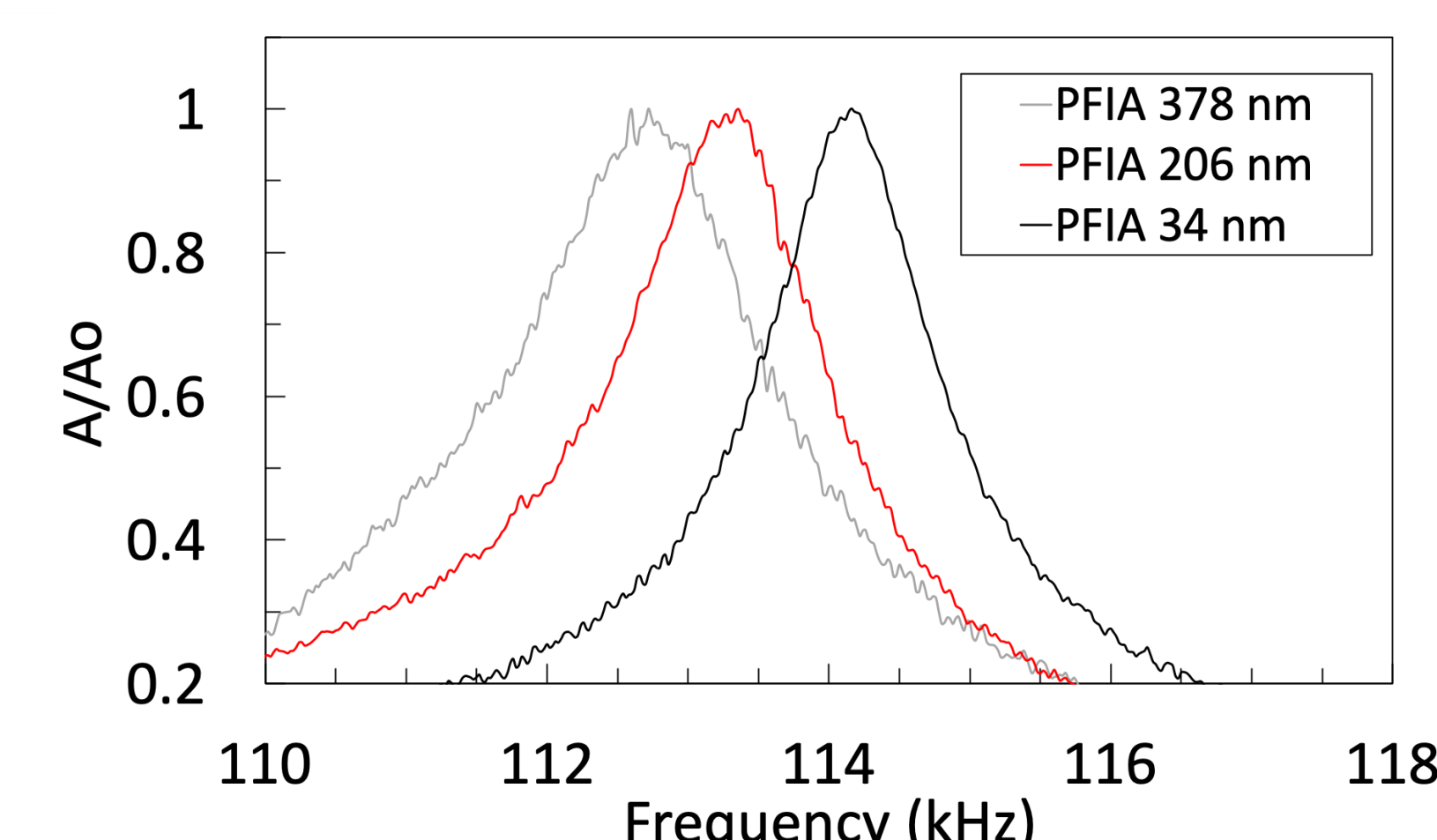


Fig 11. Sample of peaks used for measurements

## Discussion

### Film Properties and Thickness Effect

- In general, stiffness and viscosity increase as film thickness decreases below 100 nm<sup>2,3</sup>
- Increased side chain interactions and thinner films lead to less free volume and therefore more **confinement**<sup>5</sup>

### Relative Humidity

- Relative humidity has a profound impact on viscoelastic properties<sup>4</sup>
- Water creates **strong** intermolecular forces due to highly polar side chains (hydrogen bonds)
- Antiplasticization** occurs as %RH increased from 28% to 40%.

## Conclusions and Future Work

### Thin Film Polymers

- Storage and loss modulus increase as relative humidity increases and ionomer film thickness goes below 100 nm.
- In this specific study, the similar storage and loss modulus of Nafion and PFIA could be attributed to similar backbone structure and only slightly different flexibility of side chains.

### Future Work

- Modify CRFM machine to create chamber to control **relative humidity**
- Implement Electrochemical Impedance Spectroscopy (EIS) to measure film **conductivity** in different relative humidities

## References and Acknowledgments

- Holdcroft, S. *Chem. Mater.* **2014**, *26*, 381-393
- Dishari, S. K.; Hickner, M. A. *Macromolecules* **2013**, *46*, 413-421.
- Dishari, S. K.; Hickner, M. A. *ACS Macro Lett.* **2012**, *1*, 291-295.
- Majsztrik, Paul W., et al. *Macromolecules* **2008**, *41*, 24, 9849-9862
- Yuya, P. A.; Hurley, D. C.; Turner, J. A. *J. Appl. Phys.* **2008**, *104*, 074916

This work is done under the support of UCARE funding for academic year 2019-2020 and Dr. Dishari's NSF CAREER Award (#1750040). Jackson thanks Dr. Dishari for her exceptional advising and guidance in this work from proposal to completion. Jackson thanks Tyler Johnson for mentoring him in research activities, including sample preparation and CRFM training. Jackson also thanks Ehsan Zamani for providing the ellipsometry measurements. A special thanks to NERC for permission to use world class equipment for measurements.