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Accuracy of Lysimetric, Energy Balance, and Stability-corrected Aerodynamic Methods of Estimating Above-canopy Flux of CO₂¹

S. B. Verma and N. J. Rosenberg²

ABSTRACT

Measurement of carbon dioxide exchange is necessary to indicate rates of photosynthesis in the field. Methods which are accurate over short periods of 15 min to 1 hour when are actually increased to changing weather conditions
on crop photosynthetic activity. Three micrometeoro-
logical methods—the lysimetric, energy balance, and sta-
bility-corrected aerodynamic—can be used for field de jective of this paper is to present a detailed comparison
of the performances of the three methods. Measurements were made in an oat field at Mead, Neb. Results obtained
with the three methods were in good agreement, both on
a short-period and on a daily basis. The calculated CO₂
flux agreed reasonably well with estimates from othe oat-photosynthesis studies. Midday CO₂ flux rates ranged
from about 2 to 3×10^{-7} g cm⁻² sec⁻¹. The three micrometerological methods were subjected to detailed error analysis to evaluate the relative influence of errors in measurement of the constituent input parameters used to calculate the flux of carbon dioxide, namely water vapor
flux and gradients of vapor pressure and CO_2 concentra-
tion in the lysimetric method; net radiation, soil heat flux and gradients of vapor pressure, air temperature and CO₂ concentration in the energy balance method; and gradients of windspeed, air temperature and CO₂ concentration in the stability-corrected aerodynamic method.

Additional index words: Micrometeorology, Microclimate, Energy balance, Lysimeters, Aerodynamic, CO₂ Flux, Error analysis, Field-photosynthesis, Stability correction, Exchange coefficients, Turbulent transfer.

W^E wish to know the influence of changing weath-
er conditions and the influence of microclimate and/or plant architecture alterations on the water use efficiency. To do so it is necessary to make short period measurements of photosynthesis and evapotranspiration simultaneously.

Precision weighing lysimeters are adequate for the short period determination of evapotranspiration rates under field conditions (van Bavel and Myers, 1962; Tanner, 1967; Rosenberg et al., 1968). An energy balance approach utilizing the Bowen Ratio concept has also been shown to agree with the lysimetrically measured fluxes within acceptable limits (Tanner, 1960; Denmead and McIlroy, 1970; Blad and Rosenberg, 1974). Less progress has been made, however, in rapid, short-period determination of photosynthesis in the field.

Non-meteorological methods have been proposed by Thomas and Hill (1949) and Baker and Musgrave (1964). These workers have used plastic chambers to enclose small portions of a field in order to measure the exchange of CO₂ between air and crop. The use of such chambers may, however, drastically disturb the plant microenvironment. Despite the fact that controlled environment chambers and cuvettes of quite elaborate design have been developed (for example, see Lange et al., 1969) results obtained with such equipment will remain of limited utility until they are compared with results of open field micrometeorological measurements made simultaneously.

Three micrometeorological tools for field determination of Find the flux of CO_2 (F_6) to a crop surface are considered in this paper. These are the lysimetric, the energy balance and the aerodynamic methods. All are founded on Fick's law of diffusion:

> $F_c = I K_c \frac{\partial C}{\partial E}$ $[1]$

where $F_e = CO_2$ flux, K_e = turbulent exchange coefficient for CO_2 , and $c = CO_2$ concentration at height z. In order to use any of these methods in determining F_e it is customary to assume that the exchange coefficient

and vapor pressure gradients measured over the evaporating
surface with appropriate sensors [e.g., by aspirated, shielded thermocouple psychrometers (Rosenberg and Brown, 1974)]:

$$
w = \frac{LE}{60 \frac{M_w / M_g L \rho}{P}} \tag{2}
$$

where $LE =$ latent heat flux, $L =$ latent heat of vaporization, where LE = latent heat flux, L = latent heat of vaporization,

E = water vapor flux, ρ = air density, P = atmospheric pres-

sure, and M_w . M_a = molecular weights of water and air, re-

spectively. F_c is obtained

soil heat flux (S) and gradients of vapor pressure $(\frac{\partial e}{\partial r})$ and aT

temperature
$$
(\frac{\sigma}{\partial z})
$$
:
\n
$$
K = \frac{-(Ra + S)}{60\varphi C_p \frac{\partial T}{\partial z} + \frac{M_w/M_a L}{\varphi} \frac{\partial \alpha}{\partial z}} \tag{3}
$$

 F_e can then be obtained from equations [1] and [3].

 \mathbf{r}_e can use that the covariance term of 1958) and Lemon (1960) attempted to develop
aerodynamic estimates of \mathbf{F}_e by equating \mathbf{K}_e with the exchange
coefficient for momentum (\mathbf{K}_m) . \mathbf{K}_m was calcu which prevail, generally, for very short periods of time. Under non-neutral conditions these assumptions may lead to significant errors. *Stability corrections* must be applied to acrodynamic
methods if flux estimates are to be correct. Assuming $K_w = K_c$, equation [1] becomes: $F_c = f K_w \frac{\partial c}{\partial z} = f K_m \left(\frac{K_w}{K_m}\right) \frac{\partial c}{\partial z}$

or
$$
F_c = I k^2 z^2 \left(\frac{\partial U}{\partial z}\right) \left(\frac{\partial c}{\partial z}\right) \left(\phi_m^{-2}\right) \left(\frac{K_w}{K_m}\right)
$$
 [4]

where wind speed gradient $\frac{\partial U}{\partial z} = \frac{u}{kz} \phi_m$, with $u = \text{fric}$ -

air

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tion velocity = $(\tau/\rho)^{1/2}$; τ = shear stress; ϕ_m = non-dimensional wind shear (or diabatic correction factor for the log-law wind profile, $\partial U/\partial z = u / k_z v$ \mathbf{u} 2

$$
F^{22112} = a \lambda x \lambda b \lambda m = \frac{1}{(\partial U/\partial z)} = k^2 z^2 (\phi_m)^{-2} \partial U/\partial z, k
$$

= von Karman's constant.

= von Karman's constant. Both ϕ_m and K_w/K_m have been
found to vary with atmospheric thermal stability, which is usually expressed in terms of the Richardson number, Ri. In this study we use stability corrections proposed in some recent micrometeorological investigations (Businger, 1966; Dyer, unpublished;
Dyer and Hicks, 1970; Webb, 1970; Pruitt, Morgan, and Lourence,

\n- 1. Model of Prutt et
$$
y_1 \pm 1
$$
 (1973) (noncatter called the - PML-model) $\phi_{\rm m} = (1 - 16 \text{ RV})^{-1/3}$ for unstable stratification, $= (1 + 16 \text{ RV})^{1/3}$ for stable stratification, $K_{\rm w}/K_{\rm m} = 1$. 13 (1 - 60 R1)^{0.074} for unstable stratification, $= 0.42$.
\n- 2. Model of Buainger (1966) and Dyer (amplitiated) (nresafter called the B-D model): $\phi_{\rm m} = (1 - 16 \text{ R1})^{-0}$. 25 for unstable stratification, $K_{\rm m} \approx (1 - 16 \text{ R1})^{-0}$. 25 for unstable stratification, $K_{\rm m} \approx (1 - 16 \text{ R1})^{0}$. 25 for unstable arratification, $= (1 - 16 \text{ R1})^{0.25}$ for unstable arratification. $= 16 \times 10^{-11}$ for a table stratification. $= 16 \times 10^{-11}$ for a table character, $$

$$
\overline{K}_{m}^{m} \times \overline{K}_{m}^{n} = \{1 - 16 \text{ R1}^{10.52} \text{ for unstable stratification,}
$$

Gradients (or derivatives with respect to height) for the various profiles of e, T, c and U were evaluated using finite differences in the manner suggested by Panofsky (1965). If Ψ is a profile variable, then the gradient is given by:

$$
\frac{\partial \phi}{\partial \alpha} = \frac{\phi_1 - \phi_1}{\sqrt{\pi_1 \pi_2} \ln \left(\frac{\pi_1}{\pi_1}\right)} \tag{7}
$$

where z_1 and z_2 are the heights above the ground. This approximation is, of course, rigorous for logarithmic profiles. Using this approximation and introducing the concept of zero plane
displacement, d, we find that eq. [4] becomes:

$$
0 = \frac{1}{2} \left(\frac{(U_1 - U_1)(c_1 - c_1)}{\alpha_1 \frac{c_1 - d_1}{c_1 - d_1}} \right) \left(\phi_{\frac{m}{m}}^{-2} \frac{K_w}{K_m} \right),
$$

Below we present a detailed comparison of lysimetric, energy
balance and aerodynamic (stability corrected) estimates of F_c
over an oat field. We find that the three methods compare well over an oat neig, we find that the three methods compare wen
on both a short period (30 min) and a daily basis. The calculated CO_2 flux rates agree reasonably well with results from
other studies of oat photosynthesis.

In this paper we also give a detailed error analysis of the three micrometeorological methods. Perhaps disagreements concerning the usefulness of the micrometeorological methods (e.g. Wright and Brown, 1967; Harper et al., 1973) may have been due, at least in part, to artifacts or experimental errors of a nature exposed by the analyses we discuss. The error analysis should serve as a useful tool, in its own right, for investigators considering the use of micrometeorological methods in estimating field photosynthesis since the instrumental accuracies required are clearly established.

ERROR ANALYSIS

The magnitude of errors (or a measure of uncertainty-interval) in the calculation of F_e due to errors in measurement of the constituent input parameters are estimated employing the root sum square error analysis technique (Kline and McClintock, 1953; Blad and Rosenberg, 1974). Using eq. [1], we find that the relative error in F_e , ($\sigma F_e/F_e$), is estimated by:

$$
\frac{\sigma(\mathbf{F}_c)}{\mathbf{F}_c} = \left[\left\{ \frac{\sigma(\mathbf{K}_c)}{\mathbf{K}_c} \right\}^2 + \left\{ \frac{\sigma(\Delta c)}{(\Delta c)} \right\}^2 \right]^{\frac{1}{2}}
$$
\n[9]

The relative error in exchange coefficient is obtained from eqs. [3] and [2] as follows:

$$
\frac{\sigma(K_c)}{K_c} = \left[\left\{ \frac{\sigma(Rn+8)}{(Rn+8)} \right\}^2 + \left\{ \frac{m_1 \sigma(\Delta T)}{m_1(\Delta T) + m_2(\Delta \sigma)} \right\}^2 + \left\{ \frac{m_2 \sigma(\Delta \sigma)}{m_1(\Delta T) + m_2(\Delta \sigma)} \right\}^2 \right] \frac{1}{2} [10]
$$

for the energy balance method, where $m_1 = 60\rho C_p = 0.0165$ and $m_a = 60\rho \left(\frac{M_w / M_a}{p} \right) L = 0.0254$.

For the lysimetric method the relative error in exchange coefficient is:

$$
\frac{\sigma(K_c)}{K_c} = \left[\left\{ \frac{\sigma(LE)}{LE} \right\}^2 + \left\{ \frac{\sigma(\Delta \alpha)}{\Delta \alpha} \right\}^2 \right] ^\frac{1}{2}
$$
\n[11]

where Δe , Δc , and ΔT are, respectively, the gradients of vapor pressure, CO₂ concentration, and air temperature in the layer of air (in our case between 1.0 and 2.0 m) above ground.

Energy Balance Method

Figure 1 shows the relative errors in F_c when the energy balance method is used to compute a value of K_e . The following set of input errors was considered in the preparation of ing set of input errors was considered in the preparation of
this figure: $\sigma(Rn) \approx \sigma(Rn+5) = 0.02$ ly min⁻¹, $\sigma(\Delta T) = 0.025C$, $\sigma(\Delta e) = 0.1$ mb, and $(\Delta c) = 0.1$ ppm. $(\sigma F_e)/F_e$ is shown
as a function of vapor pressure grad

For non-advective conditions⁸ (sensible heat generation at the For non-advective conditions³ (sensible heat generation at the surface, negative ΔT) and moderate vapor pressure gradients ($\Delta c = -0.6$ to -1.0 mb), the relative error in F_e ranges from 10 to 15%. The error is gre in the advective case. Magnitude of the net radiation and soil heat flux, in the range studied, had no significant effect on the value of $\sigma (F_e)/F_e$.

The errors in input parameters assumed in preparation of Fig. 1 are quite reasonable and performance within these limits is achievable with careful instrument calibration and maintenis achievable with careful instrument calibration and mainten-
ance. In Fig. 2 effects of larger input errors are considered. In
this example, the errors in Rn, Δe , ΔT and ΔC are increased
individually by about 2 conditions. When vapor pressure gradient is low, doubling the
croomations. When vapor pressure gradient is low, doubling the
error in Δe increases $\sigma (F_e)/F_e$ by 25 to 30%. The effect on an
increase in $\sigma (\Delta c)$ on $\sigma (F$ strongly advective conditions and low vapor pressure gradient, however, this method is subject to large errors. This combination of environmental conditions is quite uncommon, however, and is not shown in the figure.

Lysimetric Method

Relative errors in F_e applicable to the lysimetric method are shown in Fig. 3 and 4. In Fig. 3 the following set of input errors were used:

 σ (LE) = 0.02 ly min⁻¹ (for LE measured over 30 min),

 σ (Δ e) = 0.1 mb, and σ (Δ c) = 0.1 ppm.

700

^a A detailed discussion of the significance of advective and non-advective conditions is given in a recent paper by Blad and Rosenberg (1974).

⁴ (σ F_e)/F_e is expressed throughout this paper in terms of percentage, i.e. 0.4 = 40%. Alterations in the value of (σ F_e)/F_e due to changes in the independent variables are expressed as due to changes in the inteperment variables are expressed from 1 to 5 ppm, $(\sigma F_c)/F_c$ ranges from 7 to 12% under non-advective conditions, and 17 to 22% for conditions of strong sensible heat advection. (With the set of input errors given above.)

Fig. 1. Relative errors in F_c calculated by the energy balance method for the range of assumed errors in input parameters
and the range of assumed errors in input parameters ly min^{-1}).

Fig. 2. Relative errors in F_c by the energy balance method for a range of increasing errors in input parameters and for the
ambient conditions indicated ($\Delta C = 1$ ppm). Cross-hatched bands the data sets indicated by the symbols to the left.

When vapor pressure gradient is moderate and LE flux and Δc are strong, the error in F_e ranges from 10 to 16%. With weak LE flux and Δc the error ranges from 16 to 21%. How-
ever, with low vapor pressure gradient, $\sigma(F_c)/F_c$ is about 25 to 28% . In each case, the error is reduced by 3 to 4% when strong Δc is substituted for smaller values.

Fig. 3. Relative errors in F_c calculated by the lysimetric method for the ambient conditions and input errors specified. LE flux of 1.2 and 0.3 ly min⁻¹ shown.

Under most experimental circumstances the errors in input Under most experimental circumstances the errors in input
parameters assumed in the preparation of Fig. 3 are achievable.
Figure 4 shows the effect in increasing σ (LE), σ (Δc), and σ (Δc)
to 0.05 ly min⁻¹ is 5 to 9% if Δc is small, and negligible if Δc is large.

Aerodynamic Method

The error in Γ_s applicable to the aerodynamic method is computed in a slightly different way. For a given set of input parameters, ΔU , Δc , ΔT , T and d, the value of Γ_e is computed using eqs. [8], [5] and of input parameters of varying magnitude and their associated
errors. Assuming $\sigma(\Delta U) = 5$ cm/sec, $\sigma(\Delta c) = 0.1$ ppm, $\sigma(\Delta T)$
= 0.025 C, $\sigma(T) = 0.2$ C and $\sigma(d) = 3$ cm, values of $\sigma(F_e)/F_e$
ranged from 1 to 2% of large $\$

At first sight, the error inherent in the aerodynamic method does not seem severe. It should be emphasized, however, that superimposed on the error calculated above is an uncertainty in
the empirical expressions for the stability correction factor
 $(\phi_m^{-a} K_c/K_m)$. This uncertainty can easily lead to an additional
10 to 20% error in estimation

EXPERIMENTAL AND COMPUTATIONAL **PROCEDURES**

Field measurements of evapotranspiration and CO₂ fluxes were made during the period 30 Sept. 1972 to 10 Oct. 1972 at were made during the period av sept. 1374 to 10 Oct. 1374 at
the University of Nebraska Agricultural Meteorology Research
Laboratory (41° 09' N; 96° 30' W; 354 m above m.s.l.) located
near Mead, Neb. The experimental field planted to oats (*Avena saliva* L., market mixture of 'Neal' and planted to oats (*Avena saliva* L., market mixture of 'Neal' and 'Dakota' cultivars) on 17 Aug. 1972 to provide field cover as well as a uniform greensward f Fetch at the site of micrometeorological instrumentation was approximately 150 m to the south and 100 to 125 m to the southeast and southwest. Crop height during the study was about 40 to 50 cm.

 $CO₂$ concentration ($[CO₃$)) and gradients ($\Delta CO₂$) were measured with two Infra Red Gas Analyzers – a Beckman Model 315 for absolute analysis and a Grubb Parsons SB-2 for differential analysis. An automatic calibration system was developed to permit performance checks of both analyzers to be made hourly with a set of standard gases. Details of the sampling and analysis system are reported by Rosenberg and Verma.⁵ CO₂ gradients measured between the elevations of 1.0 and 2.0 m $CO₂$ above the ground were used in this study.

Air temperature and vapor pressure profiles were measured
with thermocouple psychrometer assemblies described by Rosen-
berg and Brown (1974). Measurements were made at 1.0, 1.25,
1.50, and 2.0 m above the ground. Instrume accuracy of the psychrometer assemblies were checked once each hour by automatically moving all sensors into the same horizontal plane and noting the agreement in dry and wet bulb temperatures. Corrections were made during the later computational process.

Profiles of wind speed were measured with a set of 6 or 7 Casella-Sheppard type sensitive cup anemometers mounted at 0.75, 1.0, 1.25, 1.50, 1.75, 2.0, and 2.25 m above the ground.

The average values of LE measured with two improved van Bavel-Myers automatic precision weighing lysimeters (Rosenberg and Brown, 1970) were used in this study. Net radiation was measured with Swissteco type S1 and Middleton model CN6 net radiometers. Soil heat flux was obtained with Middleton heat flux plates buried 4 cm deep in the soil.

All meteorological measurements were made twice on the quarter hour. Data were logged on punched tape with an auto-matic analog-to-digital data system. These data were converted from the digitized emf or count record of individual sensors into parametric forms with a series of computer programs. All data were averaged over half-hour periods.

Fig. 5. CO₂ flux over oats at Mead, Neb., computed by the lysimetric, energy balance and two aerodynamic methods. A, 30 Sept. 1972; B, 5 Oct. 1972; C, 7 Oct. 1972; and D, 10 Oct. 1972.

Gradients of vapor pressure, CO_2 , air temperature, and wind
speed measured between 1.0 and 2.0 m above ground were used
in calculating F_e . Lysimetric CO_2 flux was calculated with eqs.
[1] and [2]. Energy balance est the aerodynamic method. The stability correction factors pro-
posed in eq. [5] (the PML model) and eq. [6] (the B-D model)
were used. Values of Ri (based on virtual temperature) in the

⁶ A system and program for monitoring CO₂ concentration and gradients in an agricultural region. Submitted for Publication.

Table 1. Weather conditions at Mead, Neb., on days of oat photosynthesis observation.

Parameter	Units	Elevation	Date in 1972			
			30 Sept.	5 Oct.	7 Oct.	10 Oct.
		m				
Solar radiation	ly day ⁻¹		471	172	456	188
Latent heat flux	day^{-1} ÌУ		- 281	-162	-230	-227
Max, temp.	c		22.9	21.4	23.0	23.8
Min, temp.	C		0.4	B.O	-6.6	12.7
Mean temp.	c		11.6	15.2	10.0	17.6
Мах. уарог ргеза.	mbar		11.4	15.7	14.3	20.9
Min. vapor press.	mbar		5.6	10.0	3.6	11.7
Mean vapor press.	mbar		9.0	13.6	8.2	15.9
Mean wind speed	me^{-1}		2,4	2.8	2.1	4.2

Table 2. Comparison of average daytime CO₂ fluxes (mg CO₂ dm⁻² hr⁻¹ or g cm⁻² hr⁻¹ \times 10^b) over oats at Mead, Neb., measured by three micrometeorological methods.

air layer 1.0 to 2.0 m were computed from the profiles of wind
speed, air temperature, and vapor pressure. These values (rang-
ing from -0.06 to 0.02) were used to quantify the stability cor-
rection factor. Values of z made under nearly neutral conditions.

RESULTS

Comparison of the Micrometeorological Methods in the Field

The CO₂ flux, computed with the three methods, is presented in Fig. 5A through 5D for four different days. (A summary of weather conditions on these days is given in Table 1). In view of the large number of independent data sets used and the different sources of error outlined in the previous section, the agreement in fluxes estimated by the three methods is good. We choose to consider the lysimetric method as the standard and most direct measure against which to compare the other methods. Both the energy balance method and the B-D aerodynamic methods tend to underestimate F_e by 10 to 20% and 10 to 30%, respectively. The PML aerodynamic method seems to overestimate F_c by 15 to 20% at times. Significantly better agreement is obtained when average daytime fluxes are considered (Table 2).

It should be noted that the fluxes of both water vapor or sensible heat will be similarly estimated by the three methods since these methods assume that $K_c = K_w = K_h$. Water vapor flux is underestimated by the energy balance method (Blad and Rosenberg, 1974). The PML aerodynamic model assumes that $K_w/K_m = 1.13$ under neutral conditions and that von Karman's constant k has a value of 0.42. The B-D model uses K_h/K_m (= K_w/K_m) = 1.0 under
neutral conditions and $k = 0.40$. There is some evidence in the micrometeorological literature that K_w/K_m (or K_h/K_m) can be greater than 1.0 under neutral conditions (e.g. Businger et al., 1971). Use of K_w/K_m (or K_h/K_m) = 1.10 to 1.15 under neutral conditions will increase the Businger-Dyer stability

correction factor by 10 to 15% which will, in turn, increase the predicted F_c values to more closely agree with the lysimetric estimates. Lysimeters provide the most direct and accurate measurement of water vapor flux. We must also note that, while their performance is generally better, the precision lysimeters at Mead may differ by 10 to 20% at times because of mechanical and/or crop uniformity problems. In view of these considerations the overall comparison between the lysimetric, energy balance, and aerodynamic methods of estimating CO₂ flux seems quite good.

Comparison With Other Studies

It is instructive to establish whether any of the methods tested provide estimates of F_c which are reasonable in terms of field photosynthesis. The literature on field photosynthesis by oats is not abundant. Criswell and Shibles (1971) determined the photosynthesis rate of flag leaves of 20 oat genotypes in an artificially lighted, water-jacketed chamber. Light intensity ranged from 0.43 to 0.24 ly min⁻¹ (400 to 700 nm waveband). The rates of photosynthesis during 1 year of study ranged from 0.67 to 1.33 \times 10⁻⁷ g CO₂ cm⁻² (leaf area) sec⁻¹ for these varieties and from 0.61 to 0.89 \times 10⁻⁷ g CO₂ cm⁻² (leaf area) sec⁻¹ during a second year of study. Specific leaf dry weight ranged from about 0.36 to about 0.53 g dm⁻² in the varieties tested during both years. If this chamber research realistically mimicks net photosynthesis in the field, we should expect peak field $CO₂$ flux rates to range from about 1.8_z to 3.9×10^{-7} g cm⁻² (ground area) sec⁻¹ (assuming a reasonable value of leaf area/ground area of about 3). Our observations show that during midday these values are, indeed, realistic. The comparison is intended only as an order of magnitude check on our results.

SUMMARY AND CONCLUSIONS

Three micrometeorological methods, the lysimetric, energy balance and stability-corrected aerodynamic, were used to determine $CO₂$ exchange rates from oats under field conditions. Results of these methods were in good agreement, both on a short-period and on a daily basis. These estimates compared well with those from other oat-photosynthesis studies.

An error analysis of the lysimetric and energy balance methods reveals that errors in vapor pressure gradient can be important. Errors in $CO₂$ gradient are also important except when these gradients are strong. Errors in net radiation, air temperature gradient and lysimetrically measured water vapor flux are not important.

Under a set of given weather conditions the lysimetric method is normally subject to relatively smaller errors than is the energy balance method. Under conditions of strong sensible heat advection and with small vapor pressure gradients, the energy balance method is subject to large errors. This combination of conditions occurs very rarely, however.

The results of our error analysis and field tests show that althrough aerodynamic methods are frequently criticized, they can if properly corrected for stability conditions, be effectively used to estimate flux of CO₂. Errors are greatest under calm conditions. The uncertainly in the value of empirical constants used for stability corrections must also be considered.

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APPENDIX: NOTATION AND CUSTOMARY **UNITS**

\equiv sensible heat flux (ly min⁻¹)
 \equiv CO₂ concentration (ppm) \mathbf{A}

-
- C_{p} \equiv specific heat at constant pressure (cal gm⁻¹ C⁻¹)
- d $=$ zero plane displacement (cm)
- ${\bf E}$ $=$ water vapor flux (gm cm⁻² min⁻¹)
- e $=$ vapor pressure (mbar)
- F.
- \equiv CO₂ flux (gm cm⁻² sec⁻¹)
= conversion factor for CO₂ from ppm to specific weight $\mathbf f$
- acceleration due to gravity (cm sec-2) $\frac{g}{k}$
- von Karman's constant
- K_a , K_m , K_w = Exchange coefficients for CO₂, sensible heat, K_{c} momentum and water vapor (cm² sec⁻¹) L
- latent heat of vaporization (cal gm-1)
- LE = latent heat flux (ly min⁻¹)
- = Richardson number = $g(\frac{\partial \theta_v}{\partial x})/\theta_v(\frac{\partial U}{\partial x})^2$ Ri
-
-
- Rn = net radiation (ly min⁻¹)

S = soil heat flux (ly min⁻¹)

T = air temperature (C or Absolute)
- = virtual temperature (C or Absolute)
= wind speed (m sec⁻¹ or cm sec⁻¹) $T_{\rm v}$
- \mathbf{U}
- $=$ friction velocity (m sec⁻¹ or cm sec⁻¹) u
-
- = height above ground (cm or m) M_w , M_a = molecular weight of water and air $m_1 = 60 \rho C_p = 0.0165$

$$
m_a = 60 \rho \frac{\text{M}_\text{w}}{\text{M}} L = 0.0254
$$

- ϕ_{m} $=$ diabatic correction factor
- $=$ a function
- θ
- θ .
- = potential temp. (C or Absolute)
= virtual potential temp. (C or Absolute)
= shear stress (gm cm⁻¹ sec⁻²)
- $=$ air density (gm cm⁻³)
-
- $=$ adiabatic lapse rate (C cm⁻¹) σ .
- $=$ error in various parameters

NOTES: Subscripts 1 and 2 refer to the two heights z_1 and z_2 . In this study, $z_1 = 1m$ and $z_2 = 2m$ were used. Differences of c, e, U, and T measured between z_1 and z_2 are indicated in terms on c,
e, U, and T measured between z_1 and z_2 are indicated in terms
of the respective gradients (or finite differences) Δc , Δe , ΔU
and Δ

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